

ALL ENVIRONMENTAL, INC.

Environmental Engineering & Construction

March 7, 1997
AEI Project No. 1412

Mr. John Jay
Jay Phares Corporation
10700 MacArthur Boulevard, Suite 200
Oakland, CA 94605

Subject: **Soil Remediation Summary**
10700 MacArthur Boulevard, Oakland, California

Dear Mr. Jay:

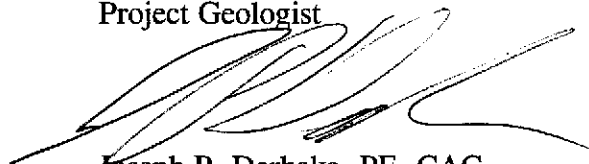
All Environmental Inc. (AEI) has prepared this Soil Remediation Summary to document the remedial activities performed at the above referenced site.

Please do not hesitate to contact Bryan Campbell at (510) 283-6000 or Joseph P. Derhake at (310) 328-8878 if you have any questions.

Sincerely,
ALL ENVIRONMENTAL, INC.



Bryan Campbell
Project Geologist



Joseph P. Derhake, PE, CAC
Senior Author

cc: Mr. Will Mast, PES Environmental
1682 Novato Boulevard, Suite 100, Novato, CA 94947

Mr. Barney Chan, Alameda County Health Care Services Agency
1131 Harbor Bay Parkway, Rm. 250, Alameda, CA 94502-6577

Corporate Headquarters:

3364 Mt. Diablo Blvd.
Lafayette, CA 94549
Phone: (510) 283-6000
Fax: (510) 283-6121

Sacramento Office:

5524 Assembly Ct., Suite 10
Sacramento, CA 95823
Phone: (916) 429-0776
Fax: (916) 424-0182

Los Angeles Office:

111 N. Sepulveda Blvd., #250
Manhattan Beach, CA 90266
Phone: (310) 328-8878
Fax: (310) 798-2841

ENVIRONMENTAL
PROTECTION

March 7, 1997

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SOIL REMEDIATION SUMMARY

10700 MacArthur Boulevard
Oakland, California

3/7/97

Project No. 1412

Prepared for:

Jay Phares Corporation
10700 MacArthur Boulevard, Suite 200
Oakland, CA 94605

Prepared by:

All Environmental, Inc.
3364 Mt. Diablo Boulevard
Lafayette, CA 94549
(800) 801-3224

AEI

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1.0 INTRODUCTION

All Environmental Inc. (AEI) has prepared this report to document the remedial action performed at Foothill Square, 10700 MacArthur Boulevard, Oakland, California. This reports summarizes the on-site aeration of 2,400 cubic yards (cy) of contaminated soil, which has taken place between April of 1996 through January of 1997.

In general, this project has followed the course of action, which was set forth in AEI's Soil Remediation Workplan (herein referred to as the "Workplan"), dated March 25, 1996, which was submitted to and approved by the Alameda County Health Care Serves Agency (ACHCSA).

1.1 Authorization

AEI was contracted to perform sampling and consulting during the remediation of the soil on-site by Jay-Phares Corporation. Mr. John Jay, (referred to as the "Client") acted as the project manager and his staff performed all of the physical work, specifically, the soil tilling. AEI's role was as a consultant. AEI prepared a Site Safety Plan, collected baseline and confirmation soil samples, and prepared this report.

employees
trained?

1.2 Target Cleanup Levels

The proposed Target Cleanup Levels were developed using the EPA Region IX Preliminary Remediation Goals (PRG) for soil (Second Half of 1995). In order to provide a factor of safety, AEI proposed to remediate the soil stockpiled on-site until contaminant levels are two orders of magnitude below the PRGs for soil at residential properties.

Table 1 shows the proposed Target Cleanup Levels next to the PRG for each chemical.

Table 1: Target Cleanup Levels

Compound	Target Cleanup Level	Residential PRGs
Tetrachloroethene	70 ug/Kg	7,000 ug/Kg
Trichloroethene	71 ug/Kg	7,100 ug/Kg
1,2 Dichloroethene	590 ug/Kg	59,000 ug/Kg*
1,1 Dichloroethene	<5 ug/Kg **	38 ug/Kg

* The PRG for Cis 1,2 Dichloroethene was used, in order to be conservative because the PRG for Trans 1,2 Dichloroethene is greater. The Target Cleanup Level for 1, 2 Dichloroethene will apply to mixtures of Cis and Trans 1,2 Dichloroethene.

** 5 ug/Kg is the method detection limit.

AEI chose the target cleanup levels conservatively to ensure that the soil could be safely reused on-site.

These action levels were presented in AEI's Workplan and were approved by Barney Chan of the ACHCSA.

1.4 Site History

Approximately 2,400 cubic yards of soil contaminated with chlorinated solvents was excavated from beneath and in the vicinity of Young's Cleaners, a former dry cleaning operation, previously located in Space #9 of Foothill Square Shopping Center at the above referenced site (refer to Figure 1). This excavation began on October 12, 1995 and was completed on December 22, 1995. The backfilling of the site was completed on January 13, 1996.

Excavated soil had elevated levels of tetrachloroethene, trichloroethene, 1,1 dichloroethene, and 1,2 dichloroethene, as documented by the 82 soil samples collected during the excavation. The contaminated soil was stockpiled on-site and covered with plastic. The soil was eventually moved to the extreme southeast corner of the property (refer to Figure 1), where it was spread out for aeration.

2.0 SOIL SAMPLING PROCEDURES

2.1 *Field Screening and Sampling Procedures*

AEI collected baseline soil samples and confirmation soil samples at the beginning and end of the remediation effort. In both instances, the soil samples were collected from random locations within the aeration cell. However, all samples were collected from six inches or deeper from the soil surface.

2.1.1 *Field Screening Procedure*

AEI's field engineer would collect approximately 100 grams of soil from the desired location. This soil would be immediately inserted into a clean plastic sandwich bag and the bag was inserted into a glass mason jar. Aluminum foil was placed over the top of the open jar and the screw cap without the lid was screwed on over the aluminum foil. After approximately 10 minutes the probe of the PID was inserted through the aluminum foil and into the jar, and then a head space reading was taken. These readings were found to be rather consistent with laboratory results.

← not rec. for CHe

2.1.2 *Discrete Soil Sample Collection Procedure*

The desired sample location was identified by the field engineer and then approximately four inches of soil was removed from the surface of the sample location. A clean thin-wall sample tube was then pounded into the soil using a wooden hammer. The tube was retrieved and immediately capped with Teflon tape and plastic caps. The samples were placed in an iced cooler and shipped to the laboratory under proper chain of custody.

2.6 *Baseline Soil Sampling and Analysis*

The contaminated soil was characterized in February, 1996 by collecting ten soil samples from random locations within the aeration cell and analyzing the soil samples for chlorinated volatile organic hydrocarbons by EPA Method 8010. The samples were sent to American Analytics Laboratory in Chatsworth for analysis of chlorinated volatile organic hydrocarbons (EPA Method 8010). The results are listed in Table 2.

Table 2: Baseline Sampling Results

Analyte \ Sample #	1	2	3	4	5	6	7	8	9	10	Mean
(micrograms per Kilogram)											n
Tetrachloroethene	88	110	18	40	31	380	14	180	270	<5*	110
Trichloroethene	<5	11	<5	<5	<5	38	<5	<5	29	<5	10
1,1 Dichloroethene	<5	<5	<5	<5	<5	<5	<5	<5	<5	<5	NA
1,2 Dichloroethene	<5	<5	<5	<5	<5	<5	<5	<5	<5	<5	NA

Results below the detection limit of 5 ug/Kg were given a value of 2.5 ug/Kg for the purposes of statistical analysis.

(VC?)

The stockpile was found to have a mean tetrachloroethene concentration of 110 parts per billion with a standard deviation of 130 parts per billion. This data served as the baseline soil profile prior to remediation.

3.0 SOIL REMEDIATION

The 2,400 cy of contaminated soil was remediated over the span of nine months by aerating the soil on-site. Aeration relies on the volatile nature of the contaminant, which simply evaporates and is released to the atmosphere. Aeration is enhanced by temperature, exposure to the atmosphere, and soil movement. By spreading the soil thinly and by tilling it the aeration process can be accelerated.

3.1 Aeration Cell Construction

The Bay Area Air Quality Management District (BAAQMD) restricts the amount of soil that can be aerated at one time. The volume of soil that can be aerated depends on the concentrations of the hydrocarbons within the soil. As the mean soil concentration is below 50 parts per million, the volume of soil that can be aerated at once is not restricted by the BAAQMD (as per rule 8-40-301).

Excavated soil was stockpiled on approximately one acre portion of Foothill Square in the extreme southeast corner of the property. The remediation cell consisted of 6 mil plastic to prevent the vertical migration and on top of the plastic the contaminated soil was spread to a thickness of 18". Hay bales were placed around the stockpile to inhibit water or soil run-off during times of high precipitation.

3.2 Soil Tilling

Significant soil aeration was achieved during the placement and grading of the soil. Then, during the nine months in which the soil aerated, Jay Phares Corporation's crew tilled the soil once. The soil was turned by attaching a ripper to the back of a skip loader and dragging the blades of the ripper through the stockpile soil.

3.3 Confirmation Sampling

Confirmation soil samples were taken on January 23, 1997, in accordance with the EPA's Test Methods for Evaluating Solid Waste (SW-846). The initial number of soil samples collected was based on the recommendations of AEI's workplan which states that approximately 10 soil samples would be collected. Soil samples were collected from random locations within the aeration cell at least six inches below the surface and were sent to McCampbell Analytical, Inc. in Pacheco (State Certification #1644) for analysis of chlorinated volatile organic hydrocarbons (EPA Method 8010). Refer to Figure 2 for sample locations. Laboratory results and chain of custody documentation are included in Appendix C. Results of this analysis are listed in Table 3.

Table 3: Confirmation Sampling Results

Analyte \ Sample #	1	2	3	4	5	6	7	8	9	10	Mean
(micrograms per kilogram)											
Tetrachloroethene	12	49	28	66	110	<10	15	<10	23	<10	31.8
Trichloroethene	<5	<5	<5	5.8	<5	<5	<5	<5	<5	<5	NA
1,1 Dichloroethene	<5	<5	<5	<5	<5	<5	<5	<5	<5	<5	NA
1,2 Dichloroethene	<5	<5	<5	<5	<5	<5	<5	<5	<5	<5	NA

Results below the detection limit of 10 ug/Kg were given a value of 5 ug/Kg for the purposes of statistical analysis.

Analysis of the soil samples revealed that almost all of the samples had trichloroethene concentrations below the detection limit; and there was no 1,1 dichloroethene, and 1,2 dichloroethene detection in any of the samples. Therefore, calculating the standard deviation and confidence interval for these three data sets would be meaningless.

The analytical results do show appreciable concentrations of tetrachloroethene and a calculation of this data set is necessary to show that this contaminant is present in concentrations below the previously proposed Target Cleanup Level. Refer to Appendix A for excerpts of EPA's SW-846 (Third Edition, Part 4 of 4, pages 1-17) for specific equations and general procedures of obtaining and evaluating random sample results of contaminated solid wastes. Refer to Appendix B, a summary of the calculation of the referenced equations used for the confirmation sampling.

The conditions of SW-846 requires statistical analyses of sample data to ensure the 90% confidence interval has been reached. The Confidence Interval for concentrations of tetrachloroethene confirmation sampling was calculated to be $31.80 \text{ ug/Kg} \pm 15.05 \text{ ug/Kg}$ which, at the upper limit of 46.86 ug/Kg is still less than the Target Cleanup Level of 70 ug/Kg for this contaminant. ?

Also, an appropriate number of sample were used in the chemical characterization of the aerating soil for tetrachloroethene. The Appropriate Number of Samples to Collect was calculated to be 2 which is much lower than the number of samples actually analyzed. Therefore, this data set satisfies AEI's Workplan's remediation goals.

4.0 CONCLUSIONS AND RECOMMENDATIONS

The remediation goals of this project have been achieved. As was shown by the sampling data discussed in the previous section, the 2,400 cy of remediated soil contains only trace levels of solvents. The soil should be considered eligible for limited reuse on-site or disposal off-site.

106TH AVENUE

ADJACENT RESIDENTIAL

LUCKY SUPERMARKET

EXCAVATED AREA

RETAIL STORES

MEDICAL CLINIC & GOVT. OFFICE

RETAIL STORES

GOVT. OFFICE

PARKING LOT

PARKING LOT

DAY CARE

AERATION CELL

108TH AVENUE

MACARTHUR BLVD.

FOOTHILL BLVD.



ALL ENVIRONMENTAL, INC.
3364 MT. DIABLO BOULEVARD, LAFAYETTE, CA

SCALE:

APPROVED BY:

DRAWN BY:

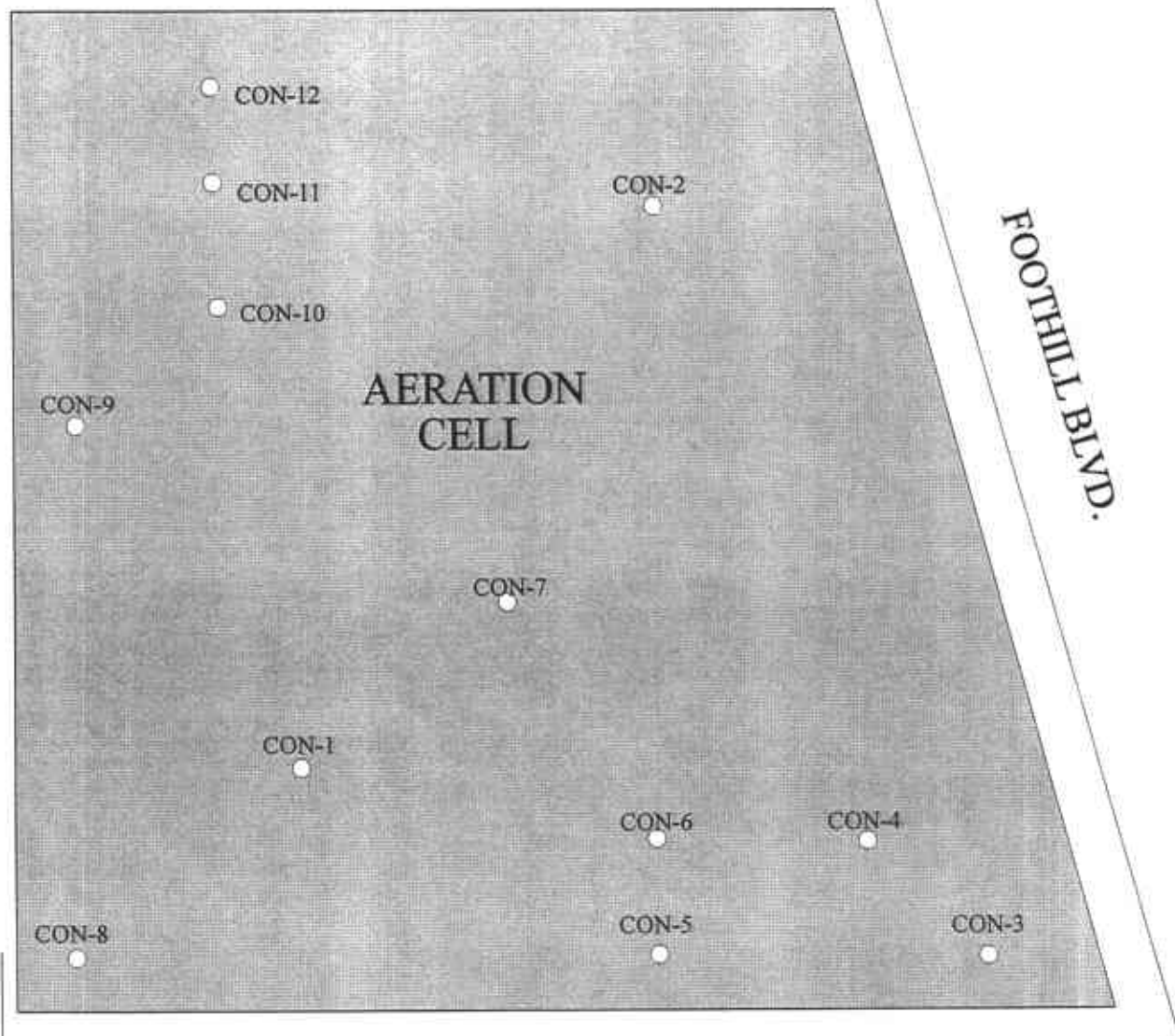
DATE:

REVISED:

SITE MAP

10700 MACARTHUR BOULEVARD
OAKLAND, CA

DRAWING NUMBER:
FIGURE 1



ENTRANCE TO
PARKING LOT

108TH AVENUE



CON-1 CONFIRMATION SAMPLE #1
(1/23/97)

ALL ENVIRONMENTAL, INC.
3364 MT. DIABLO BOULEVARD, LAFAYETTE, CA

SCALE:	APPROVED BY:	DRAWN BY:
DATE:		REVISED:

SAMPLE LOCATION MAP

10700 MACARTHUR BOULEVARD
OAKLAND, CA

DRAWING NUMBER:
FIGURE 2

APPENDIX A
LABORATORY DATA

All Environmental, Inc. 3364 Mt. Diablo Blvd. Lafayette, CA 94549	Client Project ID: # 1412; Foothill Square	Date Sampled: 01/23/97
		Date Received: 01/24/97
	Client Contact: Joe Derhake	Date Extracted: 01/24/97
	Client P.O.:	Date Analyzed: 01/24/97

Volatile Halocarbons

EPA method 601 or 8010

Lab ID	73151	73152	73153	73154
Client ID	CON-1	CON-2	CON-3	CON-4
Matrix	S	S	S	S
Compound	Concentration*			
Bromodichloromethane	ND	ND	ND	ND
Bromoform ^(b)	ND	ND	ND	ND
Bromomethane	ND	ND	ND	ND
Carbon Tetrachloride ^(c)	ND	ND	ND	ND
Chlorobenzene	ND	ND	ND	ND
Chloroethane	ND	ND	ND	ND
2-Chloroethyl Vinyl Ether ^(d)	ND	ND	ND	ND
Chloroform ^(e)	ND	ND	ND	ND
Chloromethane	ND	ND	ND	ND
Dibromochloromethane	ND	ND	ND	ND
1,2-Dichlorobenzene	ND	ND	ND	ND
1,3-Dichlorobenzene	ND	ND	ND	ND
1,4-Dichlorobenzene	ND	ND	ND	ND
Dichlorodifluoromethane	ND	ND	ND	ND
1,1-Dichloroethane	ND	ND	ND	ND
1,2-Dichloroethane	ND	ND	ND	ND
1,1-Dichloroethene	ND	ND	ND	ND
cis 1,2-Dichloroethene	ND	ND	ND	ND
trans 1,2-Dichloroethene	ND	ND	ND	ND
1,2-Dichloropropane	ND	ND	ND	ND
cis 1,3-Dichloropropene	ND	ND	ND	ND
trans 1,3-Dichloropropene	ND	ND	ND	ND
Methylene Chloride ^(f)	ND	ND	ND	ND
1,1,2,2-Tetrachloroethane	ND	ND	ND	ND
Tetrachloroethene	12	49	28	66
1,1,1-Trichloroethane	ND	ND	ND	ND
1,1,2-Trichloroethane	ND	ND	ND	ND
Trichloroethene	ND	ND	ND	5.8
Trichlorofluoromethane	ND	ND	ND	ND
Vinyl Chloride ^(g)	ND	ND	ND	ND
% Recovery Surrogate	90	98	93	95
Comments				

* water and vapor samples are reported in ug/L, soil and sludge samples in ug/kg and all TCLP extracts in ug/L.

Reporting limit unless otherwise stated: water/TCLP extracts, ND < 0.5ug/L; soil and sludge, ND < 5ug/kg

ND means not detected above the reporting limit; N/A means analyte not applicable to this analysis

(b) tribromomethane; (c) tetrachloromethane; (d) (2-chloroethoxy) ethene; (e) trichloromethane; (f) dichloromethane; (g) chloroethene;
(h) a lighter than water immiscible sheen is present; (i) liquid sample that contains greater than ~ 5 vol. % sediment.

All Environmental, Inc. 3364 Mt. Diablo Blvd. Lafayette, CA 94549	Client Project ID: # 1412; Foothill Square	Date Sampled: 01/23/97
		Date Received: 01/24/97
	Client Contact: Joe Derhake	Date Extracted: 01/24/97
	Client P.O.:	Date Analyzed: 01/24/97

Volatile Halocarbons

EPA method 601 or 8010

Lab ID	73155	73156	73157	73158
Client ID	CON-5	CON-6	CON-7	CON-8
Matrix	S	S	S	S
Compound	Concentration *			
Bromodichloromethane	ND	ND	ND	ND
Bromoform ^(b)	ND	ND	ND	ND
Bromomethane	ND	ND	ND	ND
Carbon Tetrachloride ^(c)	ND	ND	ND	ND
Chlorobenzene	ND	ND	ND	ND
Chloroethane	ND	ND	ND	ND
2-Chloroethyl Vinyl Ether ^(d)	ND	ND	ND	ND
Chloroform ^(e)	ND	ND	ND	ND
Chloromethane	ND	ND	ND	ND
Dibromochloromethane	ND	ND	ND	ND
1,2-Dichlorobenzene	ND	ND	ND	ND
1,3-Dichlorobenzene	ND	ND	ND	ND
1,4-Dichlorobenzene	ND	ND	ND	ND
Dichlorodifluoromethane	ND	ND	ND	ND
1,1-Dichloroethane	ND	ND	ND	ND
1,2-Dichloroethane	ND	ND	ND	ND
1,1-Dichloroethene	ND	ND	ND	ND
cis 1,2-Dichloroethene	ND	ND	ND	ND
trans 1,2-Dichloroethene	ND	ND	ND	ND
1,2-Dichloropropane	ND	ND	ND	ND
cis 1,3-Dichloropropene	ND	ND	ND	ND
trans 1,3-Dichloropropene	ND	ND	ND	ND
Methylene Chloride ^(f)	ND	ND	ND	ND
1,1,2,2-Tetrachloroethane	ND	ND	ND	ND
Tetrachloroethene	110	ND < 10	15	ND < 10
1,1,1-Trichloroethane	ND	ND	ND	ND
1,1,2-Trichloroethane	ND	ND	ND	ND
Trichloroethene	ND	ND	ND	ND
Trichlorofluoromethane	ND	ND	ND	ND
Vinyl Chloride ^(g)	ND	ND	ND	ND
% Recovery Surrogate	102	93	97	95
Comments				

* water and vapor samples are reported in ug/L, soil and sludge samples in ug/kg and all TCLP extracts in ug/L.

Reporting limit unless otherwise stated: water/TCLP extracts, ND < 0.5ug/L; soil and sludge, ND < 5ug/kg

ND means not detected above the reporting limit; N/A means analyte not applicable to this analysis

(b) tribromomethane; (c) tetrachloromethane; (d) (2-chloroethoxy) ethene; (e) trichloromethane; (f) dichloromethane; (g) chloroethene; (h) a lighter than water immiscible sheen is present; (i) liquid sample that contains greater than ~ 5 vol. % sediment.

All Environmental, Inc. 3364 Mt. Diablo Blvd. Lafayette, CA 94549	Client Project ID: # 1412; Foothill Square	Date Sampled: 01/23/97
		Date Received: 01/24/97
	Client Contact: Joe Derhake	Date Extracted: 01/24/97
	Client P.O.:	Date Analyzed: 01/24/97

Volatile Halocarbons

EPA method 601 or 8010

Lab ID	73159	73160		
Client ID	CON-9	CON-10		
Matrix	S	S		
Compound	Concentration *			
Bromodichloromethane	ND	ND		
Bromoform ^(b)	ND	ND		
Bromomethane	ND	ND		
Carbon Tetrachloride ^(c)	ND	ND		
Chlorobenzene	ND	ND		
Chloroethane	ND	ND		
2-Chloroethyl Vinyl Ether ^(d)	ND	ND		
Chloroform ^(e)	ND	ND		
Chloromethane	ND	ND		
Dibromochloromethane	ND	ND		
1,2-Dichlorobenzene	ND	ND		
1,3-Dichlorobenzene	ND	ND		
1,4-Dichlorobenzene	ND	ND		
Dichlorodifluoromethane	ND	ND		
1,1-Dichloroethane	ND	ND		
1,2-Dichloroethane	ND	ND		
1,1-Dichloroethene	ND	ND		
cis 1,2-Dichloroethene	ND	ND		
trans 1,2-Dichloroethene	ND	ND		
1,2-Dichloropropane	ND	ND		
cis 1,3-Dichloropropene	ND	ND		
trans 1,3-Dichloropropene	ND	ND		
Methylene Chloride ^(f)	ND	ND		
1,1,2,2-Tetrachloroethane	ND	ND		
Tetrachloroethene	23	ND < 10		
1,1,1-Trichloroethane	ND	ND		
1,1,2-Trichloroethane	ND	ND		
Trichloroethene	ND	ND		
Trichlorofluoromethane	ND	ND		
Vinyl Chloride ^(g)	ND	ND		
% Recovery Surrogate	97	89		
Comments				

* water and vapor samples are reported in ug/L, soil and sludge samples in ug/kg and all TCLP extracts in ug/L.

Reporting limit unless otherwise stated: water/TCLP extracts, ND < 0.5ug/L; soil and sludge, ND < 5ug/kg

ND means not detected above the reporting limit; N/A means analyte not applicable to this analysis

(b) tribromomethane; (c) tetrachloromethane; (d) (2-chloroethoxy) ethene; (e) trichloromethane; (f) dichloromethane; (g) chloroethene; (h) a lighter than water immiscible sheen is present; (i) liquid sample that contains greater than ~ 5 vol. % sediment.

QC REPORT FOR EPA 8010/8020/EDB

Date: 01/24/97

Matrix: Soil

Analyte	Concentration (ug/kg)				% Recovery		
	Sample (#68845)	MS	MSD	Amount Spiked	MS	MSD	RPD
1,1-DCE	0	90	99	100	90	99	9.5
Trichloroethene	0	83	91	100	83	91	9.2
EDB	N/A	N/A	N/A	N/A	N/A	N/A	N/A
Chlorobenzene	0	82	90	100	82	90	9.3
Benzene	N/A	N/A	N/A	N/A	N/A	N/A	N/A
Toluene	N/A	N/A	N/A	N/A	N/A	N/A	N/A
Chlorobz (PID)	N/A	N/A	N/A	N/A	N/A	N/A	N/A

$$\% \text{ Rec.} = (\text{MS} - \text{Sample}) / \text{amount spiked} \times 100$$

$$\text{RPD} = (\text{MS} - \text{MSD}) / (\text{MS} + \text{MSD}) \times 2 \times 100$$

APPENDIX B

SW 846 STATISTICAL ANALYSIS

SW-846 CALCULATIONS

Confirmation Sampling

Confirmation soil samples were taken on January 23, 1997, in accordance with EPA's Test Methods for Evaluating Solid Waste (SW-846). Refer to Appendix A for excerpts of EPA's SW-846 (Third Edition, Part 4 of 4, pages 1-17) for specific equations and general procedures of obtaining and evaluating randomized sampling results of contaminated solid wastes. Below is a summary of the referenced equations:

Steps 1., 2., and 3.

Steps 1., 2., and 3. outline the procedures for the calculation of the initial number of random samples to be collected. However, the initial number of soil samples collected was based on the recommendations of AEI's workplan entitled "Soil Remediation Workplan" dated March 25, 1996, which states that approximately ten soil samples would be collected. In actuality a total of 13 samples were collected and only 10 were submitted for analysis. The three extra samples were placed on hold at the laboratory in case the analysis of more samples were to be required in order to reach an acceptable 90% confidence level.

Step 4.

On January 23, 1997, soil samples were collected from random locations within the aeration cell at least six inches below the surface and were sent to McCampbell Analytical, Inc. in Pacheco (State Certification #1644) for analysis of chlorinated volatile organic hydrocarbons (EPA Method 8010). Results of this analysis are listed in Table 3. These results are not characterized by any obvious abnormality.

Step 5.

The values of the sample mean (\bar{x}), variance of sample (s^2), standard deviation of sample (s), and standard error ($s_{\bar{x}}$) are calculated:

$$\begin{aligned}\bar{x} &= (12 + 49 + 28 + \dots) / 10 \\ &= 31.8, \\ s^2 &= (144 + 2401 + 784 + \dots) \times (318^2 / 10) / 9 \quad \text{Std Dev.} \\ &= 1166.84, \\ s &= \sqrt{s^2} \\ &= 34.16, \text{ and} \\ s_{\bar{x}} &= 34.16 / \sqrt{10} \quad \left(\frac{s}{\sqrt{N}} \right) \\ &= 10.80.\end{aligned}$$

Step 6.

The Regulatory Threshold (RT) for Tetrachloroethene is the Target Cleanup Level of 70 ppb listed in Table 1. The study is continued with nontransformed data.

Step 7.

The confidence interval (CI) is now calculated using \bar{x} , the student's "t" value (t_{20}), and $s_{\bar{x}}$.

$$\begin{aligned} \text{CI} &= 31.8 \pm (1.393) \times (10.80) \\ &= 31.8 \pm 15.05 \end{aligned}$$

Step 8.

The difference between RT and \bar{x} as well as the appropriate number of samples to collect from the solid waste (n_2) are now calculated.

$$\begin{aligned} \Delta &= 70 - 31.8 \\ &= 38.2 \text{ and} \\ n_2 &= ((1.393)^2 \times (34.16)^2) / (38.2)^2 \\ &= 1.55 \end{aligned}$$

$$\left(\frac{t_{20}^2 S^2}{A^2} \right)$$

Due to the large difference between the RT and empirical mean the calculated number of samples is a low number. Clearly, 10 samples is enough.

APPENDIX C

EXCERPTS FROM SW 846

CHAPTER NINE

SAMPLING PLAN

9.1 DESIGN AND DEVELOPMENT

The initial -- and perhaps most critical -- element in a program designed to evaluate the physical and chemical properties of a solid waste is the plan for sampling the waste. It is understandable that analytical studies, with their sophisticated instrumentation and high cost, are often perceived as the dominant element in a waste characterization program. Yet, despite that sophistication and high cost, analytical data generated by a scientifically defective sampling plan have limited utility, particularly in the case of regulatory proceedings.

This section of the manual addresses the development and implementation of a scientifically credible sampling plan for a solid waste and the documentation of the chain of custody for such a plan. The information presented in this section is relevant to the sampling of any solid waste, which has been defined by the EPA in its regulations for the identification and listing of hazardous wastes to include solid, semisolid, liquid, and contained gaseous materials. However, the physical and chemical diversity of those materials, as well as the dissimilarity of storage facilities (lagoons, open piles, tanks, drums, etc.) and sampling equipment associated with them, preclude a detailed consideration of any specific sampling plan. Consequently, because the burden of responsibility for developing a technically sound sampling plan rests with the waste producer, it is advisable that he/she seek competent advice before designing a plan. This is particularly true in the early developmental stages of a sampling plan, at which time at least a basic understanding of applied statistics is required. Applied statistics is the science of employing techniques that allow the uncertainty of inductive inferences (general conclusions based on partial knowledge) to be evaluated.

9.1.1 Development of Appropriate Sampling Plans

An appropriate sampling plan for a solid waste must be responsive to both regulatory and scientific objectives. Once those objectives have been clearly identified, a suitable sampling strategy, predicated upon fundamental statistical concepts, can be developed. The statistical terminology associated with those concepts is reviewed in Table 9-1; Student's "t" values for use in the statistics of Table 9-1 appear in Table 9-2.

9.1.1.1 Regulatory and Scientific Objectives

The EPA, in its hazardous waste management system, has required that certain solid wastes be analyzed for physical and chemical properties. It is mostly chemical properties that are of concern, and, in the case of a number of chemical contaminants, the EPA has promulgated levels (regulatory thresholds) that cannot be equaled or exceeded. The regulations pertaining to the management of hazardous wastes contain three references regarding the

TABLE 9-1. BASIC STATISTICAL TERMINOLOGY APPLICABLE TO SAMPLING PLANS FOR SOLID WASTES

Terminology	Symbol	Mathematical equation	(Equation)
• Variable (e.g., barium or endrin)	x	—	
• Individual measurement of variable	x_i	—	
• Mean of all possible measurements of variable (population mean)	μ	$\mu = \frac{\sum_{i=1}^N x_i}{N}$, with N = number of possible measurements	(1)
• Mean of measurements generated by sample (sample mean)	\bar{x}	<p><u>Simple random sampling and systematic random sampling</u></p> $\bar{x} = \frac{\sum_{i=1}^n x_i}{n}$, with n = number of sample measurements	(2a)
		<p><u>Stratified random sampling</u></p> $\bar{x} = \sum_{k=1}^r W_k \bar{x}_k$, with \bar{x}_k = stratum mean and W_k = fraction of population represented by Stratum k (number of strata [k] range from 1 to r)	(2b)
• Variance of sample	s^2	<p><u>Simple random sampling and systematic random sampling</u></p> $s^2 = \frac{\sum_{i=1}^n x_i^2 - (\sum_{i=1}^n x_i)^2/n}{n - 1}$	(3a)
		<p><u>Stratified random sampling</u></p> $s^2 = \sum_{k=1}^r W_k s_k^2$, with s_k^2 = stratum variance and W_k = fraction of population represent by Stratum k (number of strata [k] ranges from 1 to r)	(3b)

$$s^2 = \frac{\sum_{i=1}^n (x_i - \bar{x})^2}{n}$$

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TABLE 9-1. (Continued)

Terminology	Symbol	Mathematical equation	(Equation)
• Standard deviation of sample	s	$s = \sqrt{s^2}$	(4)
• Standard error (also standard error of mean and standard deviation of mean) of sample	$s_{\bar{x}}$	$s_{\bar{x}} = \frac{s}{\sqrt{n}}$	(5)
• Confidence interval for μ^a	CI	CI = $\bar{x} \pm t_{.20} s_{\bar{x}}$, with $t_{.20}$ obtained from Table 2 for appropriate degrees of freedom	(6)
• Regulatory threshold ^a	RT	Defined by EPA (e.g., 100 ppm for barium in elutriate of EP toxicity)	(7)
• Appropriate number of samples to collect from a solid waste (financial constraints not considered)	n	$n = \frac{t_{.20}^2 s^2}{\Delta^2}$, with $\Delta = RT - \bar{x}$	(8)
• Degrees of freedom	df	$df = n - 1$	(9)
• Square root transformation	---	$X_i + 1/2$	(10)
• Arcsin transformation	---	Arcsin p; if necessary, refer to any text on basic statistics; measurements must be converted to percentages (p)	(11)

^aThe upper limit of the CI for μ is compared with the applicable regulatory threshold (RT) to determine if a solid waste contains the variable (chemical contaminant) of concern at a hazardous level. The contaminant of concern is not considered to be present in the waste at a hazardous level if the upper limit of the CI is less than the applicable RT. Otherwise, the opposite conclusion is reached.

TABLE 9-2. TABULATED VALUES OF STUDENT'S "t" FOR EVALUATING SOLID WASTES

Degrees of freedom (n-1) ^a	Tabulated "t" value ^b
1	3.078
2	1.886
3	1.638
4	1.533
5	1.476
6	1.440
7	1.415
8	1.397
9	1.393
10	1.372
11	1.363
12	1.356
13	1.350
14	1.345
15	1.341
16	1.337
17	1.333
18	1.330
19	1.328
20	1.325
21	1.323
22	1.321
23	1.319
24	1.318
25	1.316
26	1.315
27	1.314
28	1.313
29	1.311
30	1.310
40	1.303
60	1.296
120	1.289
	1.282

^aDegrees of freedom (df) are equal to the number of samples (n) collected from a solid waste less one.

^bTabulated "t" values are for a two-tailed confidence interval and a probability of 0.20 (the same values are applicable to a one-tailed confidence interval and a probability of 0.10).

sampling of solid wastes for analytical properties. The first reference, which occurs throughout the regulations, requires that representative samples of waste be collected and defines representative samples as exhibiting average properties of the whole waste. The second reference, which pertains just to petitions to exclude wastes from being listed as hazardous wastes, specifies that enough samples (but in no case less than four samples) be collected over a period of time sufficient to represent the variability of the wastes. The third reference, which applies only to ground water monitoring systems, mandates that four replicates (subsamples) be taken from each ground water sample intended for chemical analysis and that the mean concentration and variance for each chemical constituent be calculated from those four subsamples and compared with background levels for ground water. Even the statistical test to be employed in that comparison is specified (Student's t-test).

The first of the above-described references addresses the issue of sampling accuracy, and the second and third references focus on sampling variability or, conversely, sampling precision (actually the third reference relates to analytical variability, which, in many statistical tests, is indistinguishable from true sampling variability). Sampling accuracy (the closeness of a sample value to its true value) and sampling precision (the closeness of repeated sample values) are also the issues of overriding importance in any scientific assessment of sampling practices. Thus, from both regulatory and scientific perspectives, the primary objectives of a sampling plan for a solid waste are twofold: namely, to collect samples that will allow measurements of the chemical properties of the waste that are both accurate and precise. If the chemical measurements are sufficiently accurate and precise, they will be considered reliable estimates of the chemical properties of the waste.

It is now apparent that a judgment must be made as to the degree of sampling accuracy and precision that is required to estimate reliably the chemical characteristics of a solid waste for the purpose of comparing those characteristics with applicable regulatory thresholds. Generally, high accuracy and high precision are required if one or more chemical contaminants of a solid waste are present at a concentration that is close to the applicable regulatory threshold. Alternatively, relatively low accuracy and low precision can be tolerated if the contaminants of concern occur at levels far below or far above their applicable thresholds. However, a word of caution is in order. Low sampling precision is often associated with considerable savings in analytical, as well as sampling, costs and is clearly recognizable even in the simplest of statistical tests. On the other hand, low sampling accuracy may not entail cost savings and is always obscured in statistical tests (i.e., it cannot be evaluated). Therefore, although it is desirable to design sampling plans for solid wastes to achieve only the minimally required precision (at least two samples of a material are required for any estimate of precision), it is prudent to design the plans to attain the greatest possible accuracy.

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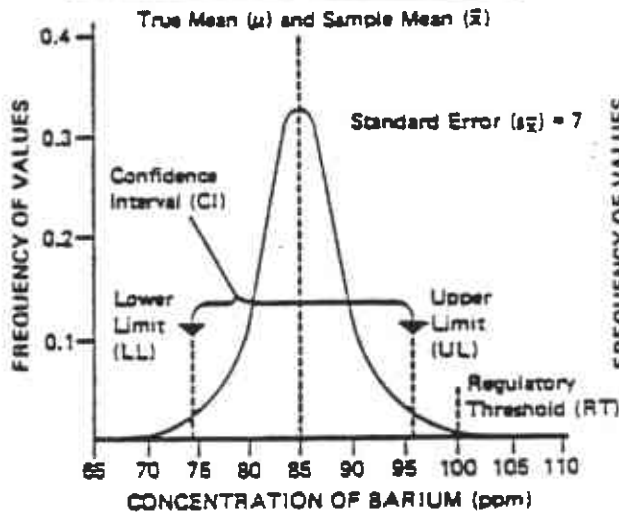
The roles that inaccurate and imprecise sampling can play in causing a solid waste to be inappropriately judged hazardous are illustrated in Figure 9-1. When evaluating Figure 9-1, several points are worthy of consideration. Although a sampling plan for a solid waste generates a mean concentration (\bar{x}) and standard deviation (s , a measure of the extent to which individual sample concentrations are dispersed around \bar{x}) for each chemical contaminant of concern, it is not the variation of individual sample concentrations that is of ultimate concern, but rather the variation that characterizes \bar{x} itself. That measure of dispersion is termed the standard deviation of the mean (also, the standard error of the mean or standard error) and is designated as $s_{\bar{x}}$. Those two sample values, \bar{x} and $s_{\bar{x}}$, are used to estimate the interval (range) within which the true mean (μ) of the chemical concentration probably occurs, under the assumption that the individual concentrations exhibit a normal (bell-shaped) distribution. For the purposes of evaluating solid wastes, the probability level (confidence interval) of 80% has been selected. That is, for each chemical contaminant of concern, a confidence interval (CI) is described within which μ occurs if the sample is representative, which is expected of about 80 out of 100 samples. The upper limit of the 80% CI is then compared with the appropriate regulatory threshold. If the upper limit is less than the threshold, the chemical contaminant is not considered to be present in the waste at a hazardous level; otherwise, the opposite conclusion is drawn. One last point merits explanation. Even if the upper limit of an estimated 80% CI is only slightly less than the regulatory threshold (the worst case of chemical contamination that would be judged acceptable), there is only a 10% (not 20%) chance that the threshold is equaled or exceeded. That is because values of a normally distributed contaminant that are outside the limits of an 80% CI are equally distributed between the left (lower) and right (upper) tails of the normal curve. Consequently, the CI employed to evaluate solid wastes is, for all practical purposes, a 90% interval.

9.1.1.2 Fundamental Statistical Concepts

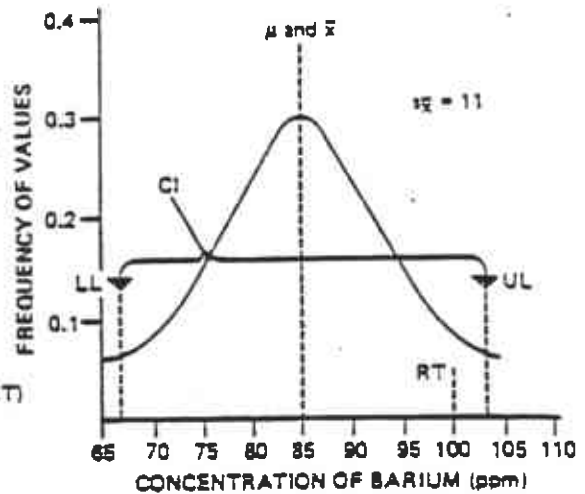
The concepts of sampling accuracy and precision have already been introduced, along with some measurements of central tendency (\bar{x}) and dispersion (standard deviation [s] and $s_{\bar{x}}$) for concentrations of a chemical contaminant of a solid waste. The utility of \bar{x} and $s_{\bar{x}}$ in estimating a confidence interval that probably contains the true mean (μ) concentration of a contaminant has also been described. However, it was noted that the validity of that estimate is predicated upon the assumption that individual concentrations of the contaminant exhibit a normal distribution.

Statistical techniques for obtaining accurate and precise samples are relatively simple and easy to implement. Sampling accuracy is usually achieved by some form of random sampling. In random sampling, every unit in the population (e.g., every location in a lagoon used to store a solid waste) has a theoretically equal chance of being sampled and measured. Consequently, statistics generated by the sample (e.g., \bar{x} and, to a lesser degree, $s_{\bar{x}}$) are unbiased (accurate) estimators of true population parameters (e.g., the CI for μ). In other words, the sample is representative of the population. One of the commonest methods of selecting a random sample is to divide the

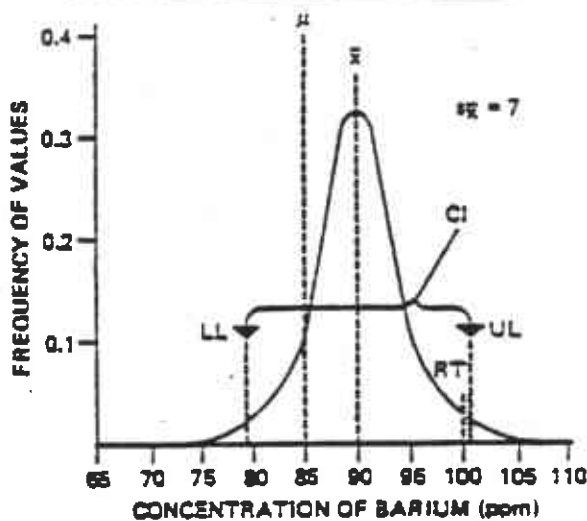
ACCURATE AND PRECISE SAMPLE
(Waste Appropriately Judged Nonhazardous)



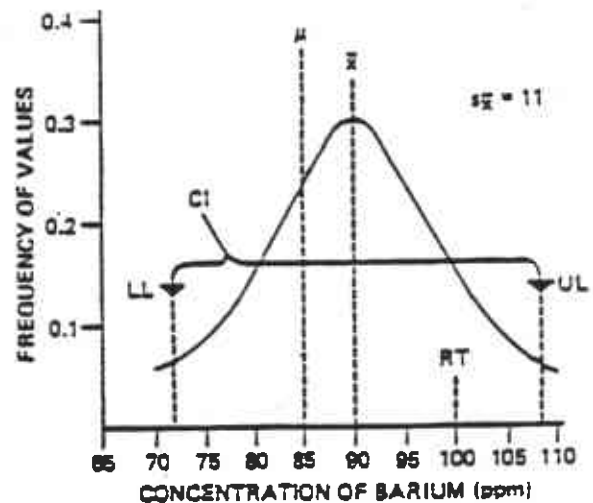
ACCURATE AND IMPRECISE SAMPLE
(Waste Inappropriately Judged Hazardous)



INACCURATE AND PRECISE SAMPLE
(Waste Inappropriately Judged Hazardous)



INACCURATE AND IMPRECISE SAMPLE
(Waste Inappropriately Judged Hazardous)



NOTE: In All Cases, Confidence Interval for $\mu = \bar{x} \pm 2.0 s_{\bar{x}}$.

Figure 9-1.—Important theoretical relationships between sampling accuracy and precision and regulatory objectives for a chemical contaminant of a solid waste that occurs at a concentration marginally less than its regulatory threshold. In this example, barium is the chemical contaminant. The true mean concentration of barium in the elutriate of the EP toxicity test is 85 ppm, as compared to a regulatory threshold of 100 ppm. The upper limit of the confidence interval for the true mean concentration, which is estimated from the sample mean and standard error, must be less than the regulatory threshold if barium is judged to be present in the waste at a nonhazardous level.

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population by an imaginary grid, assign a series of consecutive numbers to the units of the grid, and select the numbers (units) to be sampled through the use of a random-numbers table (such a table can be found in any text on basic statistics). It is important to emphasize that a haphazardly selected sample is not a suitable substitute for a randomly selected sample. That is because there is no assurance that a person performing undisciplined sampling will not consciously or subconsciously favor the selection of certain units of the population, thus causing the sample to be unrepresentative of the population.

Sampling precision is most commonly achieved by taking an appropriate number of samples from the population. As can be observed from the equation for calculating $s_{\bar{x}}$, precision increases ($s_{\bar{x}}$ and the CI for μ decrease) as the number of samples (n) increases, although not in a 1:1 ratio. For example, a 100% increase in the number of samples from two to four causes the CI to decrease by approximately 62% (about 31% of that decrease is associated with the critical upper tail of the normal curve). However, another 100% increase in sampling effort from four to eight samples results in only an additional 39% decrease in the CI. Another technique for increasing sampling precision is to maximize the physical size (weight or volume) of the samples that are collected. That has the effect of minimizing between-sample variation and, consequently, decreasing $s_{\bar{x}}$. Increasing the number or size of samples taken from a population, in addition to increasing sampling precision, has the secondary effect of increasing sampling accuracy.

In summary, reliable information concerning the chemical properties of a solid waste is needed for the purpose of comparing those properties with applicable regulatory thresholds. If chemical information is to be considered reliable, it must be accurate and sufficiently precise. Accuracy is usually achieved by incorporating some form of randomness into the selection process for the samples that generate the chemical information. Sufficient precision is most often obtained by selecting an appropriate number of samples.

There are a few ramifications of the above-described concepts that merit elaboration. If, for example, as in the case of semiconductor etching solutions, each batch of a waste is completely homogeneous with regard to the chemical properties of concern and that chemical homogeneity is constant (uniform) over time (from batch to batch), a single sample collected from the waste at an arbitrary location and time would theoretically generate an accurate and precise estimate of the chemical properties. However, most wastes are heterogeneous in terms of their chemical properties. If a batch of waste is randomly heterogeneous with regard to its chemical characteristics and that random chemical heterogeneity remains constant from batch to batch, accuracy and appropriate precision can usually be achieved by simple random sampling. In that type of sampling, all units in the population (essentially all locations or points in all batches of waste from which a sample could be collected) are identified, and a suitable number of samples is randomly selected from the population. More complex stratified random sampling is appropriate if a batch of waste is known to be nonrandomly heterogeneous in terms of its chemical properties and/or nonrandom chemical heterogeneity is known to exist from batch to batch. In such cases, the population is stratified to isolate the known sources of nonrandom chemical heterogeneity.

After stratification, which may occur over space (locations or points in a batch of waste) and/or time (each batch of waste), the units in each stratum are numerically identified, and a simple random sample is taken from each stratum. As previously intimated, both simple and stratified random sampling generate accurate estimates of the chemical properties of a solid waste. The advantage of stratified random sampling over simple random sampling is that, for a given number of samples and a given sample size, the former technique often results in a more precise estimate of chemical properties of a waste (a lower value of s_x) than the latter technique. However, greater precision is likely to be realized only if a waste exhibits substantial nonrandom chemical heterogeneity and stratification efficiently "divides" the waste into strata that exhibit maximum between-strata variability and minimum within-strata variability. If that does not occur, stratified random sampling can produce results that are less precise than in the case of simple random sampling. Therefore, it is reasonable to select stratified random sampling over simple random sampling only if the distribution of chemical contaminants in a waste is sufficiently known to allow an intelligent identification of strata and at least two or three samples can be collected in each stratum. If a strategy employing stratified random sampling is selected, a decision must be made regarding the allocation of sampling effort among strata. When chemical variation within each stratum can be estimated with a great degree of detail, samples should be optimally allocated among strata, i.e., the number of samples collected from each stratum should be directly proportional to the chemical variation encountered in the stratum. When detailed information concerning chemical variability within strata is not available, samples should be proportionally allocated among strata, i.e., sampling effort in each stratum should be directly proportional to the size of the stratum.

Simple random sampling and stratified random sampling are types of probability sampling, which, because of a reliance upon mathematical and statistical theories, allows an evaluation of the effectiveness of sampling procedures. Another type of probability sampling is systematic random sampling, in which the first unit to be collected from a population is randomly selected, but all subsequent units are taken at fixed space or time intervals. An example of systematic random sampling is the sampling of a waste lagoon along a transect in which the first sampling point on the transect is 1 m from a randomly selected location on the shore and subsequent sampling points are located at 2-m intervals along the transect. The advantages of systematic random sampling over simple random sampling and stratified random sampling are the ease with which samples are identified and collected (the selection of the first sampling unit determines the remainder of the units) and, sometimes, an increase in precision. In certain cases, for example, systematic random sampling might be expected to be a little more precise than stratified random sampling with one unit per stratum because samples are distributed more evenly over the population. As will be demonstrated shortly, disadvantages of systematic random sampling are the poor accuracy and precision that can occur when unrecognized trends or cycles occur in the population. For those reasons, systematic random sampling is recommended only when a population is essentially random or contains at most a modest stratification. In such cases, systematic random sampling would be employed for the sake of convenience, with little expectation of an increase in precision over other random sampling techniques.

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Probability sampling is contrasted with authoritative sampling, in which an individual who is well acquainted with the solid waste to be sampled selects a sample without regard to randomization. The validity of data gathered in that manner is totally dependent on the knowledge of the sampler and, although valid data can sometimes be obtained, authoritative sampling is not recommended for the chemical characterization of most wastes.

It may now be useful to offer a generalization regarding the four sampling strategies that have been identified for solid wastes. If little or no information is available concerning the distribution of chemical contaminants of a waste, simple random sampling is the most appropriate sampling strategy. As more information is accumulated for the contaminants of concern, greater consideration can be given (in order of the additional information required) to stratified random sampling, systematic random sampling, and, perhaps, authoritative sampling.

The validity of a CI for the true mean (μ) concentration of a chemical contaminant of a solid waste is, as previously noted, based on the assumption that individual concentrations of the contaminant exhibit a normal distribution. This is true regardless of the strategy that is employed to sample the waste. Although there are computational procedures for evaluating the correctness of the assumption of normality, those procedures are meaningful only if a large number of samples are collected from a waste. Because sampling plans for most solid wastes entail just a few samples, one can do little more than superficially examine resulting data for obvious departures from normality (this can be done by simple graphical methods), keeping in mind that even if individual measurements of a chemical contaminant of a waste exhibit a considerably abnormal distribution, such abnormality is not likely to be the case for sample means, which are our primary concern. One can also compare the mean of the sample (\bar{x}) with the variance of the sample (s^2). In a normally distributed population, \bar{x} would be expected to be greater than s^2 (assuming that the number of samples [n] is reasonably large). If that is not the case, the chemical contaminant of concern may be characterized by a Poisson distribution (\bar{x} is approximately equal to s^2) or a negative binomial distribution (\bar{x} is less than s^2). In the former circumstance, normality can often be achieved by transforming data according to the square root transformation. In the latter circumstance, normality may be realized through use of the arcsine transformation. If either transformation is required, all subsequent statistical evaluations must be performed on the transformed scale.

Finally, it is necessary to address the appropriate number of samples to be employed in the chemical characterization of a solid waste. As has already been emphasized, the appropriate number of samples is the least number of samples required to generate a sufficiently precise estimate of the true mean (μ) concentration of a chemical contaminant of a waste. From the perspective of most waste producers, that means the minimal number of samples needed to demonstrate that the upper limit of the CI for μ is less than the applicable regulatory threshold (RT). The formula for estimating appropriate sampling effort (Table 9-1, Equation 8) indicates that increased sampling effort is generally justified as s^2 or the "t.20" value (probable error rate) increases

and as $\Delta(RT - \bar{X})$ decreases. In a well-designed sampling plan for a solid waste, an effort is made to estimate the values of \bar{X} and s^2 before sampling is initiated. Such preliminary estimates, which may be derived from information pertaining to similar wastes, process engineering data, or limited analytical studies, are used to identify the approximate number of samples that must be collected from the waste. It is always prudent to collect a somewhat greater number of samples than indicated by preliminary estimates of \bar{X} and s^2 since poor preliminary estimates of those statistics can result in an underestimate of the appropriate number of samples to collect. It is usually possible to process and store the extra samples appropriately until analysis of the initially identified samples is completed and it can be determined if analysis of the additional samples is warranted.

9.1.1.3 Basic Sampling Strategies

It is now appropriate to present general procedures for implementing the three previously introduced sampling strategies (simple random sampling, stratified random sampling, and systematic random sampling) and a hypothetical example of each sampling strategy. The hypothetical examples illustrate the statistical calculations that must be performed in most situations likely to be encountered by a waste producer and, also, provide some insight into the efficiency of the three sampling strategies in meeting regulatory objectives.

The following hypothetical conditions are assumed to exist for all three sampling strategies. First, barium, which has an RT of 100 ppm as measured in the EP elutriate test, is the only chemical contaminant of concern. Second, barium is discharged in particulate form to a waste lagoon and accumulates in the lagoon in the form of a sludge, which has built up to approximately the same thickness throughout the lagoon. Third, concentrations of barium are relatively homogeneous along the vertical gradient (from the water-sludge interface to the sludge-lagoon interface), suggesting a highly controlled manufacturing process (little between-batch variation in barium concentrations). Fourth, the physical size of sludge samples collected from the lagoon is as large as practical, and barium concentrations derived from those samples are normally distributed (note that we do not refer to barium levels in the samples of sludge because barium measurements are actually made on the elutriate from EP toxicity tests performed with the samples). Last, a preliminary study of barium levels in the elutriate of four EP toxicity tests conducted with sludge collected from the lagoon several years ago identified values of 86 and 90 ppm for material collected near the outfall (in the upper third) of the lagoon and values of 98 and 104 ppm for material obtained from the far end (the lower two-thirds) of the lagoon.

For all sampling strategies, it is important to remember that barium will be determined to be present in the sludge at a hazardous level if the upper limit of the CI for μ is equal to or greater than the RT of 100 ppm (Table 9-1, Equations 6 and 7).

9.1.1.3.1. Simple Random Sampling

Simple random sampling (Box 1) is performed by general procedures in which preliminary estimates of \bar{X} and s^2 , as well as a knowledge of the RT, for each chemical contaminant of a solid waste that is of concern are employed to estimate the appropriate number of samples (n) to be collected from the waste. That number of samples is subsequently analyzed for each chemical contaminant of concern. The resulting analytical data are then used to conclude definitively that each contaminant is or is not present in the waste at a hazardous concentration or, alternatively, to suggest a reiterative process, involving increased sampling effort, through which the presence or absence of hazard can be definitively determined.

In the hypothetical example for simple random sampling (Box 1), preliminary estimates of \bar{X} and s^2 indicated a sampling effort consisting of six samples. That number of samples was collected and initially analyzed, generating analytical data somewhat different from the preliminary data (s^2 was substantially greater than was preliminarily estimated). Consequently, the upper limit of the CI was unexpectedly greater than the applicable RT, resulting in a tentative conclusion of hazard. However, a reestimation of appropriate sampling effort, based on statistics derived from the six samples, suggested that such a conclusion might be reversed through the collection and analysis of just one more sample. Fortunately, a resampling effort was not required because of the foresight of the waste producer in obtaining three extra samples during the initial sampling effort, which, because of their influence in decreasing the final values of \bar{X} , s_x , $t_{.20}$, and, consequently, the upper limit of the CI -- values obtained from all nine samples -- resulted in a definitive conclusion of nonhazard.

9.1.1.3.2 Stratified Random Sampling

Stratified random sampling (Box 2) is conducted by general procedures that are similar to the procedures described for simple random sampling. The only difference is that, in stratified random sampling, values of \bar{X} and s^2 are calculated for each stratum in the population and then integrated into overall estimates of those statistics, the standard deviation (s), s_x , and the appropriate number of samples (n) for all strata.

The hypothetical example for stratified random sampling (Box 2) is based on the same nine sludge samples previously identified in the example of simple random sampling (Box 1) so that the relative efficiencies of the two sampling strategies can be fully compared. The efficiency generated through the process of stratification is first evident in the preliminary estimate of n (Step 2 in Boxes 1 and 2), which is six for simple random sampling and four for stratified random sampling. (The lesser value for stratified sampling is the consequence of a dramatic decrease in s^2 , which more than compensated for a modest increase in Δ .) The most relevant indication of sampling efficiency is the value of s_x , which is directly employed to calculate the CI. In the case of simple random sampling, s_x is calculated as 2.58 (Step 9 in Box 1), and, for stratified random sampling, s_x is determined to be 2.35 (Steps 5 and 7 in Box 2). Consequently, the gain in efficiency attributable to stratification is approximately 9% ($0.23/2.58$).

BOX 1. STRATEGY FOR DETERMINING IF CHEMICAL CONTAMINANTS OF SOLID WASTES
— ARE PRESENT AT HAZARDOUS LEVELS - SIMPLE RANDOM SAMPLING

Step

General Procedures

1. Obtain preliminary estimates of \bar{x} and s^2 for each chemical contaminant of a solid waste that is of concern. The two above-identified statistics are calculated by, respectively, Equations 2a and 3a (Table 9-1).
2. Estimate the appropriate number of samples (n_1) to be collected from the waste through use of Equation 8 (Table 9-1) and Table 9-2. Derive individual values of n_1 for each chemical contaminant of concern. The appropriate number of samples to be taken from the waste is the greatest of the individual n_1 values.
3. Randomly collect at least n_1 (or $n_2 - n_1$, $n_3 - n_2$, etc., as will be indicated later in this box) samples from the waste (collection of a few extra samples will provide protection against poor preliminary estimates of \bar{x} and s^2). Maximize the physical size (weight or volume) of all samples that are collected.
4. Analyze the n_1 (or $n_2 - n_1$, $n_3 - n_2$ etc.) samples for each chemical contaminant of concern. Superficially (graphically) examine each set of analytical data for obvious departures from normality.
5. Calculate \bar{x} , s^2 , the standard deviation (s), and $s_{\bar{x}}$ for each set of analytical data by, respectively, Equations 2a, 3a, 4, and 5 (Table 9-1).
6. If \bar{x} for a chemical contaminant is equal to or greater than the applicable RT (Equation 7, Table 9-1) and is believed to be an accurate estimator of μ , the contaminant is considered to be present in the waste at a hazardous concentration, and the study is completed. Otherwise, continue the study. In the case of a set of analytical data that does not exhibit obvious abnormality and for which \bar{x} is greater than s^2 , perform the following calculations with nontransformed data. Otherwise, consider transforming the data by the square root transformation (if \bar{x} is about equal to s^2) or the arcsine transformation (if \bar{x} is less than s^2) and performing all subsequent calculations with transformed data. Square root and arcsine transformations are defined by, respectively, Equations 10 and 11 (Table 9-1).
7. Determine the CI for each chemical contaminant of concern by Equation 6 (Table 9-1) and Table 9-2. If the upper limit of the CI is less than the applicable RT (Equations 6 and 7, Table 9-1), the chemical contaminant is not considered to be present in the waste at a hazardous concentration and the study is completed. Otherwise, the opposite conclusion is tentatively reached.

8. If a tentative conclusion of hazard is reached, reestimate the total number of samples (n_2) to be collected from the waste by use of Equation 8 (Table 9-1) and Table 9-2. When deriving n_2 , employ the newly calculated (not preliminary) values of \bar{x} and s^2 . If additional $n_2 - n_1$ samples of waste cannot reasonably be collected, the study is completed, and a definitive conclusion of hazard is reached. Otherwise, collect extra $n_2 - n_1$ samples of waste.
9. Repeat the basic operations described in Steps 3 through 8 until the waste is judged to be nonhazardous or, if the opposite conclusion continues to be reached, until increased sampling effort is impractical.

Hypothetical Example

Step

1. The preliminary study of barium levels in the elutriate of four EP toxicity tests, conducted with sludge collected from the lagoon several years ago, generated values of 86 and 90 ppm for sludge obtained from the upper third of the lagoon and values of 98 and 104 ppm for sludge from the lower two-thirds of the lagoon. Those two sets of values are not judged to be indicative of nonrandom chemical heterogeneity (stratification) within the lagoon. Therefore, preliminary estimates of \bar{x} and s^2 are calculated as:

$$\bar{x} = \frac{\sum_{i=1}^n x_i}{n} = \frac{86 + 90 + 98 + 104}{4} = 94.50, \text{ and} \quad (\text{Equation 2a})$$

$$s^2 = \frac{\sum_{i=1}^n x_i^2 - (\sum_{i=1}^n x_i)^2/n}{n - 1} \quad (\text{Equation 3a})$$

$$= \frac{35,916.00 - 35,721.00}{3} = 65.00.$$

2. Based on the preliminary estimates of \bar{x} and s^2 , as well as the knowledge that the RT for barium is 100 ppm,

$$n_1 = \frac{t_{.20}^2 s^2}{\Delta^2} = \frac{(1.638^2)(65.00)}{5.50^2} = 5.77. \quad (\text{Equation 8})$$

3. As indicated above, the appropriate number of sludge samples (n_1) to be collected from the lagoon is six. That number of samples (plus three extra samples for protection against poor preliminary estimates of \bar{x} and s^2) is collected from the lagoon by a single randomization process (Figure 9-2). All samples consist of the greatest volume of sludge that

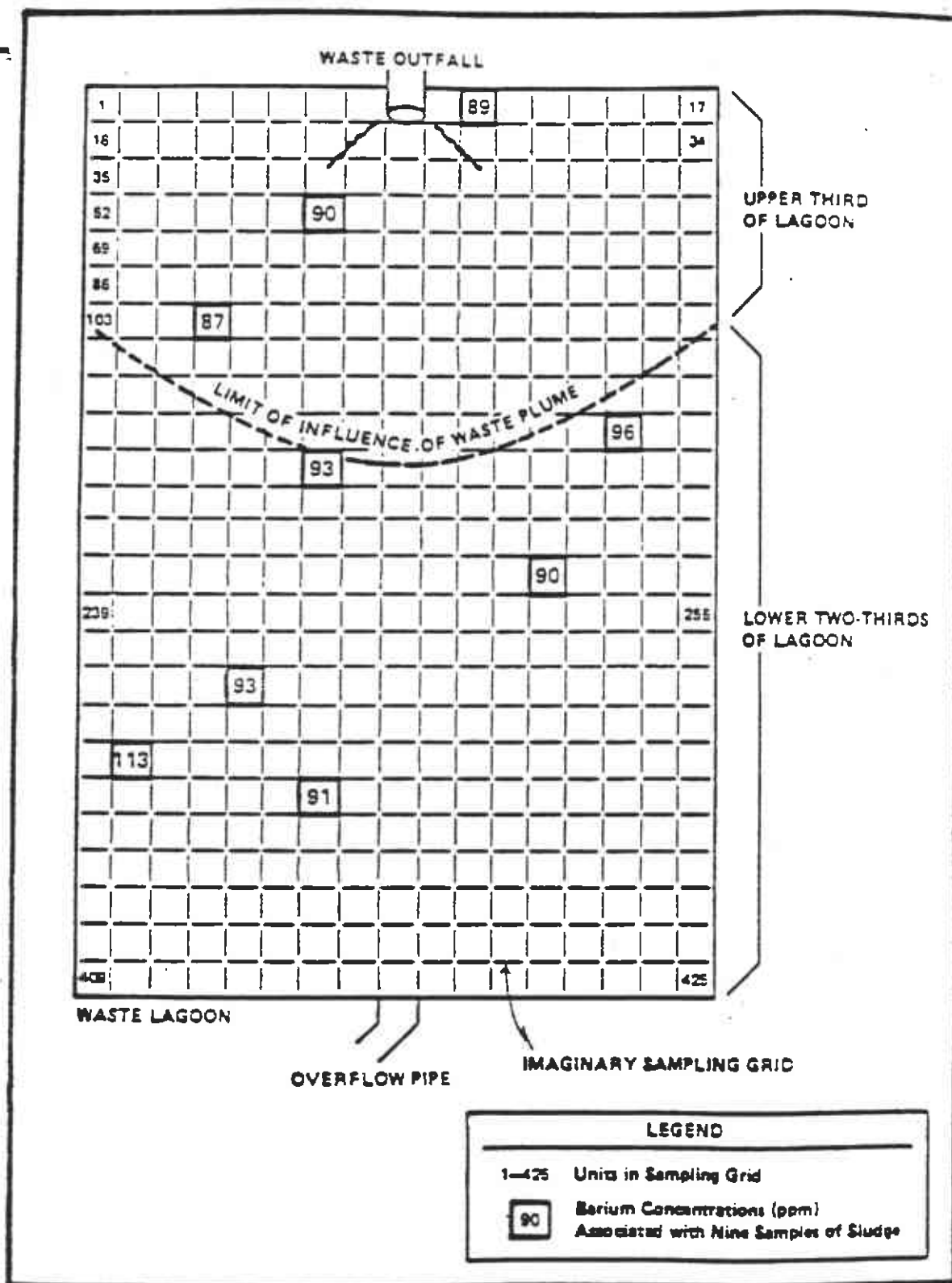


Figure 9-2.—Hypothetical sampling conditions in waste lagoon containing sludge contaminated with barium. Barium concentrations associated with samples of sludge refer to levels measured in the elutriate of EP toxicity tests conducted with the samples.

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can be practically collected. The three extra samples are suitably processed and stored for possible later analysis.

- The six samples of sludge (n_1) designated for immediate analysis generate the following concentrations of barium in the EP toxicity test: 89, 90, 87, 96, 93, and 113 ppm. Although the value of 113 ppm appears unusual as compared with the other data, there is no obvious indication that the data are not normally distributed.
- New values for \bar{x} and s^2 and associated values for the standard deviation (s) and $s_{\bar{x}}$ are calculated as:

$$\bar{x} = \frac{\sum_{i=1}^n X_i}{n} = \frac{89 + 90 + 87 + 96 + 93 + 113}{6} = 94.67, \quad (\text{Equation 2a})$$

$$s^2 = \frac{\sum_{i=1}^n X_i^2 - (\sum_{i=1}^n X_i)^2/n}{n - 1} \quad (\text{Equation 3a})$$
$$= \frac{54,224.00 - 53,770.67}{5} = 90.67,$$

$$s = \sqrt{s^2} = 9.52, \text{ and} \quad (\text{Equation 4})$$

$$s_{\bar{x}} = s/\sqrt{n} = 9.52/\sqrt{6} = 3.89. \quad (\text{Equation 5})$$

- The new value for \bar{x} (94.67) is less than the RT (100). In addition, \bar{x} is greater (only slightly) than s^2 (90.67), and, as previously indicated, the raw data are not characterized by obvious abnormality. Consequently, the study is continued, with the following calculations performed with nontransformed data.
- $CI = \bar{x} \pm t_{.20} s_{\bar{x}} = 94.67 \pm (1.476)(3.89)$ (Equation 6)
 $= 94.67 \pm 5.74.$

Because the upper limit of the CI (100.41) is greater than the applicable RT (100), it is tentatively concluded that barium is present in the sludge at a hazardous concentration.

8. n is now reestimated as:

$$n_2 = \frac{t_{.20}^2 s^2}{\Delta^2} = \frac{(1.476^2)(90.67)}{5.33^2} = 6.95. \quad (\text{Equation 8})$$

The value for n_2 (approximately 7) indicates that an additional ($n_2 - n_1 = 1$) sludge sample should be collected from the lagoon.

9. The additional sampling effort is not necessary because of the three extra samples that were initially collected from the lagoon. All extra samples are analyzed, generating the following levels of barium for the EP toxicity test: 93, 90, and 91 ppm. Consequently, \bar{x} , s^2 , the standard deviation (s), and $s_{\bar{x}}$ are recalculated as:

$$\bar{x} = \frac{\sum_{i=1}^n x_i}{n} = \frac{86 + 90 + \dots + 91}{9} = 93.56, \quad (\text{Equation 2a})$$

$$s^2 = \frac{\sum_{i=1}^n x_i^2 - (\sum_{i=1}^n x_i)^2/n}{n - 1} \quad (\text{Equation 3a})$$

$$= \frac{79,254.00 - 78,773.78}{8} = 60.03,$$

$$s = \sqrt{s^2} = 7.75, \text{ and} \quad (\text{Equation 4})$$

$$s_{\bar{x}} = s/\sqrt{n} = 7.75/\sqrt{9} = 2.58. \quad (\text{Equation 5})$$

The value for \bar{x} (93.56) is again less than the RT (100), and there is no indication that the nine data points, considered collectively, are abnormally distributed (in particular, \bar{x} is now substantially greater than s^2). Consequently, CI, calculated with nontransformed data, is determined to be:

$$\begin{aligned} CI &= \bar{x} \pm t_{.20} s_{\bar{x}} = 93.56 \pm (1.397)(2.58) \quad (\text{Equation 6}) \\ &= 93.56 \pm 3.60. \end{aligned}$$

The upper limit of the CI (97.16) is now less than the RT of 100. Consequently, it is definitively concluded that barium is not present in the sludge at a hazardous level.

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