

350 California Street 22nd Floor San Francisco, CA 94104-1435 tel 415 228 5400 fax 415 228 5450 www.bcltlaw.com

September 15, 2008

Alameda County SEP 1 6 2008

**Environmental Health** 

Mr. Jerry Wickham Senior Hazardous Materials Specialist Alameda County Health Care Services Agency Environmental Health Services Environmental Protection 1131 Harbor Bay Parkway, Suite 250 Alameda, CA 94502-6577

Re:

9201 San Leandro Street, Oakland, CA; Former Paco Pumps, Inc. Facility Fuel Leak Case No. RO0000320 and Geo Tracker Global ID T0600101592

Dear Mr. Wickham:

I am writing on behalf of our client, H. Mark Vignoles and 9201 San Leandro, LLC, the current owner of the above-referenced property located at 9201 San Leandro Street in Oakland.

On August 21, 2007, the Alameda County Health Care Services Agency ("Alameda County") sent a letter to John Lilla, of Paco Pumps, Inc., Dallas Nelson, of GP Holdings, LLC, and Mr. Vignoles to request copies of five referenced technical reports concerning previous environmental investigations at the property, dated between 1985 and 1991, and preparation of a work plan for a further investigation that would address the technical comments set forth your letter. As you know, Mr. Lilla failed to respond to the August 21, 2007 letter even though Alameda County has properly designated Paco Pumps, Inc. as the primary responsible party, and GP Holdings no longer owns the property. As a result, Mr. Vignoles complied with Alameda County's requirements by retaining a consultant, Eras Environmental, Inc. ("Eras"), to prepare and implement the requested work plan. On or about July 31, 2008, Eras submitted its Subsurface Investigation and Groundwater Monitoring Report, Quarter 2, 2008, which is currently under review by Alameda County. However, neither Mr. Vignoles nor Eras was able to provide any of the five earlier technical reports because Mr. Vignoles did not have copies of those documents.

Mr. Jerry Wickham September 15, 2008 Page 2

9201 San Leandro, LLC is currently involved in litigation with PCC Flow Technologies Holdings, Inc. ("PCC Flow"), which is the corporate successor to Paco Pumps, Inc., concerning PCC Flow's failure to complete required investigation and remediation activities at the subject property. In the context of that litigation, PCC Flow recently produced to our office two of the five requested reports: (1) Danes & Moore. 1987. Site Contamination Study -- Paco Pumps Facility, Oakland, for Amsted Industries; and (2) Jonas & Associates, Inc. 1991. Soil Characterization Report, Stained Asphalt/Concrete Area -- Paco Pumps, Inc., 9201 San Leandro Street, Oakland, California, October 30, 1991. Therefore, in response to Alameda County's request, a copy of each of these reports is enclosed with this letter. PCC Flow did not produce, and thus Mr. Vignoles continues to be unable to provide, the other three technical reports.

John Lilla was formerly Vice President of Paco Pumps, Inc. and is currently a Vice President of PCC Flow. The corporate headquarters of PCC Flow is located at 16801 Greenspoint Park Drive, Houston, Texas, which is the address used for Mr. Lilla in Alameda County's August 21, 2007 letter. Thus, although Mr. Lilla and PCC Flow presumably received that letter, PCC Flow failed and refused for over a year to provide the enclosed reports directly to Alameda County and continues to refuse to comply with Alameda County's directives to further investigate and remediate the contamination caused by Paco Pumps, Inc. at the property.

In addition to the August 21, 2007 letter, Alameda County sent at least three subsequent letters concerning the property to the same three addressees (Messrs. Lilla, Nelson, and Vignoles) -- on October 11, 2007, and January 31 and May 9, 2008. With respect to Mr. Lilla, the three subsequent letters were addressed to him, for Paco Pumps, Inc., at 800 Koomey Road, Brookshire, Texas. This is to advise Alameda County that the Paco Pumps, Inc. located in Brookshire, Texas is not the same corporate entity as the Paco Pumps, Inc. which formerly owned and operated the property at 9201 San Leandro Street in Oakland.

Through the litigation, we have learned that in 1994, PCC Flow acquired all the stock of the Paco Pumps, Inc. that formerly owned and operated at 9201 San Leandro Street in Oakland, and Paco Pumps, Inc. later merged into PCC Flow. In 2004, PCC Flow sold the assets of Paco Pumps, Inc. to Sulzer Pumps, Inc. ("Sulzer"). However, Sulzer did not assume any of the pre-existing environmental liabilities of Paco Pumps, Inc. To the contrary, PCC Flow reportedly agreed to indemnify Sulzer for any such liabilities. In 2006, Sulzer in turn sold the assets of Paco Pumps, Inc. to another company, Grundfos CBS, Inc. ("Grundfos"). Thus, the Paco Pumps, Inc. located at Koomey Road, Brookshire, Texas is apparently owned by Grundfos.

PCC Flow has admitted in the litigation that it is the corporate successor by merger to Paco Pumps, Inc., the former owner and operator of the subject property in Oakland. For this reason, we respectfully suggest that Alameda County direct further correspondence concerning this property to Mr. John Lilla, Vice President, PCC Flow Technologies Holdings, Inc., 16801 Greenspoint Park Drive, Suite 355, Houston, Texas, 77060.

Mr. Jerry Wickham September 15, 2008 Page 3

Finally, this is to advise Alameda County that, in 1995, Paco Pumps, Inc. entered into an Indemnification Agreement in connection with its sale of the subject property to GP Holdings, LLC. Under the Indemnification Agreement, Paco Pumps, Inc. agreed to investigate, monitor, and remediate all contaminants released into or on the property prior to the transfer of the property to GP Holdings, LLC, as requested or directed by any regulatory agency exercising authority over such matters, including Alameda County, until such time as the agency provided written notice that no further action is required. Paco Pumps, Inc. further agreed to indemnify GP Holdings, LLC, its lenders, and any of their assigns and successors in interest which may take title to the property, including Mr. Vignoles and 9201 San Leandro, LLC, from any liabilities, damages, fees, penalties, or losses arising out of any demand, claim, or suit by any agency or other party relating to any contaminants released into or onto the property prior to the transfer of the property to GP Holdings, LLC. (For your reference, a copy of the Indemnification Agreement is enclosed with this letter.) This Indemnification Agreement by Paco Pumps, Inc. and in favor subsequent property owners certainly supports Alameda County's August 15, 2007 Notice of Responsibility making Paco Pumps, Inc. the primary responsible party, and also supports making our client secondarily responsible inasmuch as our client is merely the current owner and did not cause or contribute to the unauthorized releases. Thus, in failing to comply with Alameda County's requirements for the property, PCC Flow, as successor to Paco Pumps, Inc., is in breach of its contractual obligations under the Indemnification Agreement as well as in breach of its obligations to Alameda County as primary responsible party.

Please contact me if you have any questions.

Sincerely,

Man Zeppetello

Marc A. Zeppetello

MAZ/fmc

Enclosures

cc: H. Mark Vignoles, 9201 San Leandro, LLC (w/o enclosures)

Mr. John Lilla, Vice President, PCC Flow (w/o enclosures)

Mr. Scott Kaplan, Counsel for PCC Flow (w/o enclosures)

# SOIL CHARACTERIZATION REPORT STAINED ASPHALT/CONCRETE AREA

PACO PUMPS, INC. 9201 SAN LEANDRO STREET OAKLAND, CALIFORNIA October 30, 1991



# JONAS & ASSOCIATES INC. Environmental Consultants

SOIL CHARACTERIZATION REPORT STAINED ASPHALT/CONCRETE AREA

PACO PUMPS, INC. 9201 SAN LEANDRO STREET OAKLAND, CALIFORNIA October 30, 1991

## SOIL CHARACTERIZATION REPORT STAINED ASPHALT/CONCRETE AREA

PACO PUMPS, INC. 9201 SAN LEANDRO STREET OAKLAND, CALIFORNIA

October 30, 1991

Prepared for

PACO Pumps, Inc. Oakland, California

Prepared by

Jonas & Associates Inc. 1056 Dale Place Concord, California 94518 (415) 676-8554

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#### **EXECUTIVE SUMMARY**

Soil Characterization Investigation PACO Pumps, Inc. October 30, 1991

In September of 1991, PACO Pumps retained Jonas & Associates Inc. to perform a soil investigation of an area located at the western boundary of their property, adjacent to the Central Pacific Railroad tracks. This area is approximately 35 feet by 15 feet in dimensions. The floor of this area is covered with asphalt and concrete pad, which extends into an asphalted parking lot. Drums and bins containing scrap metals are temporarily stored at this location. In this report, this area is identified as the Asphalt/Concrete Area.

Stained ground surface was observed at this location. In order to determine whether any or a release of chemical substances has occurred in this area, two soil samples were collected. Soil samples were collected in brass sleeves from the first six inches of the native soil encountered underneath the asphalt/concrete pad. Following the soil collection, the boring locations were plugged with the cuttings and re-asphalted.

Both samples were analyzed by Chromalab Inc., a California State Certified Analytical Laboratory for the following parameters:

- Extractable Petroleum Hydrocarbons (EPH) EPA Method 3550/8015;
- Volatile Petroleum Hydrocarbons (VPH) EPA Method 5030/8015;
- Volatile Organic Compounds (VOCs) EPA Method 8240;
- Semi-Volatile Organic Compounds (Semi-VOCs) EPA Method 8270;
- Pesticides and Polychlorinated Biphenyi EPA Method 8080; and
- Title 22 Metals EPA Method 6010 and 7000 series.

On October 12, 1991, laboratory analytical data was received by Jonas & Associates Inc. Extractable Petroleum Hydrocarbons, Volatile Petroleum Hydrocarbons, Volatile Organic Compounds, Semi-Volatile Organic Compounds, and Pesticides were not detected above the method detection limits in both of the soil samples. Polychlorinated Biphenyl (PCB 1260) was detected in both samples B6 and B7 at concentrations of 0.40 milligram/kilogram (mg/kg) and 0.67 mg/kg, respectively.

A source for PCB found in the soil is not yet identified. According to 40 CFR, Part 761, Subpart G - PCB Spill Cleanup Policy:

"PCB contaminated soil should be removed to 10 mg/kg, provided that soil is excavated to a minimum depth of 10 inches. The excavated soil should be replaced with clean soil (less than 1 mg/kg PCBs)."

The PCB detected in this area is well below 10 mg/kg. In addition, PCB concentrations detected in this area also are below the Total Threshold Limit Concentration Value of 50 mg/kg.

Metals were detected in both of the samples. Barium (B6:10.7 mg/kg; B7:138 mg/kg), Cadmium (B6:0.595 mg/kg; B7:1.99 mg/kg), Cobalt (B6:0.804 mg/kg; B7:0.991 mg/kg), Chromium (B6:2.63 mg/kg; B7:57.8 mg/kg), and Nickel (B6:1.28 mg/kg; B7:348 mg/kg) were detected in both of the samples. Lead and Zinc were detected in sample B6 at concentrations of 85.4 mg/kg and 22.5 mg/kg, respectively. Molybdenum and Vanadium were detected in Sample B7 at concentration levels of 8.85 mg/kg and 146 mg/kg, respectively.

None of the metals exceeded the Total Threshold Limit Concentrations (TTLC). Since the detected PCB and metals are below the required regulatory levels, cleanup of this area may not be required. However, the final determination must be made by the regulatory agencies. It is recommended that a copy of this report be submitted to the Alameda County Health Care Services Agency for their review and approval. Their address is as follows:

Alameda County Health Care Services Agency Department of Environmental Health Hazardous Materials Program 80 Swan Way, Room 200 Oakland, California 94621

Attention: Mr. Barney M. Chan, Hazardous Materials Specialist

\* NOTE: A \$400 deposit is required for submittal to the Alameda County Health Care Services Agency.

#### Soil Characterization Investigation PACO Pumps, Inc. October 30, 1991

#### SECTION 1.0

#### INTRODUCTION

#### 1.1 FACILITY DESCRIPTION

PACO Pumps, Inc. is located at 9201 San Leandro Street, in the City of Oakland, Alameda County, California. Figure 1-1 is a regional site location map. This parcel is approximately 4.8 acres in size with approximately 92,240 square feet of office and manufacturing space. It is bounded to the north by Q.A. Products Inc., to the west by the Central Pacific Railroad, to the south by St. Vincent de Paul Resale Shop, and to the east by San Leandro Street, the Bay Area Transit (BART) and Western Pacific Railroad (WPRP).

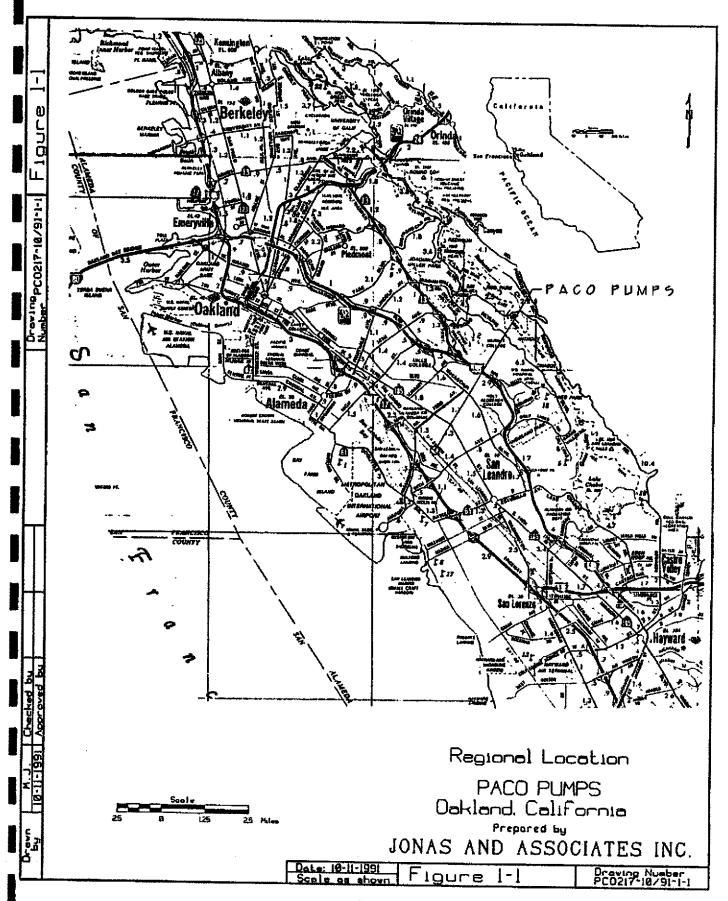
PACO Pumps has been operating at this facility since 1945. Currently PACO Pumps is owned by Newpac Industries Inc. (Newpac). Prior to Newpac, the facility was owned by Amsted Industries, Inc. Before PACO's occupancy, a tent manufacturer operated at the site. Previous to that, this was the site of a foundry.

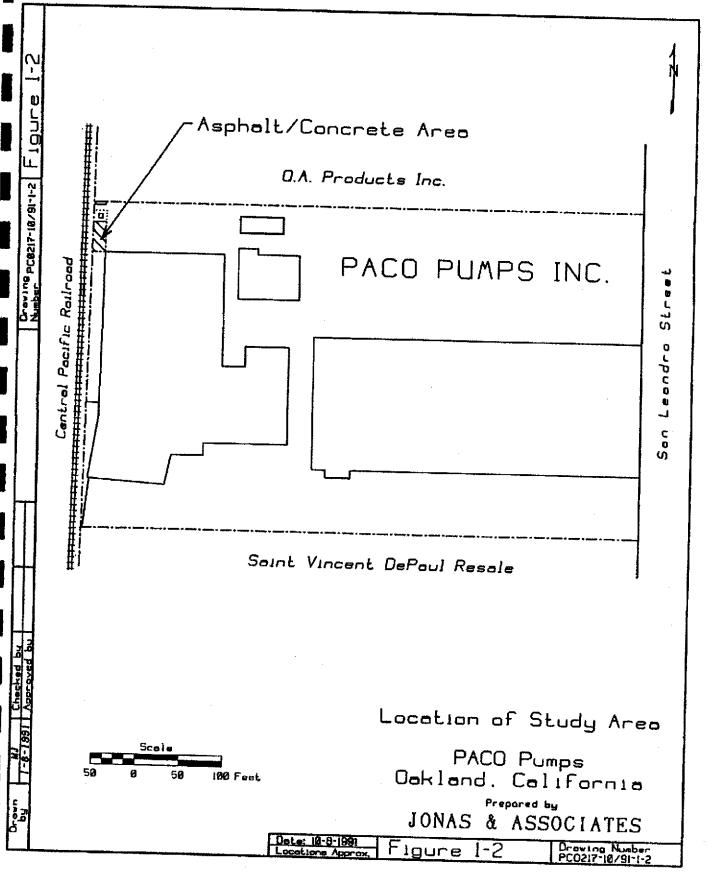
PACO Pumps manufactures pumps. Pump manufacturing is a 5-step process: machining, welding, assembly, painting, and packaging. Rough castings are received that require machining. The machining process includes milling, turning, and grinding, often using water-soluble cutting fluids. After welding and assembly, the parts are painted. Painting is performed in one of the two paint booths equipped with a water wash system to capture overspray. Waste paint is separated from the water with a centrifuge and the water is recirculated. Finished pumps are then packaged for shipment.

#### 1.2 PURPOSE AND SCOPE

In September of 1991, PACO Pumps retained Jonas & Associates Inc. to perform a soil investigation of an area located at the western boundary of their property, adjacent to the Central Pacific Railroad tracks. This area is approximately 35 feet by 15 feet in dimensions. The floor of this area is covered with asphalt and concrete pads, which extend into an asphalted parking lot. Drums and bins containing scrap metals are temporarily stored at this location. Figure 1-2 identifies the location of the study area. In this report, this area will be referred to as Asphalt/Concrete Area.

The purpose of this report is to present a description of the facility and the study area, to discuss general environmental setting of the study area, describe the soil sampling activities performed by Jonas & Associates Inc., discuss the results of the soil sampling and analysis, and make recommendations for future actions.





This Soil Characterization Investigation report consists of the following three sections:

Section 1.0 is the introduction and presents an overview of the project, lists the contents of this report, and presents the environmental setting of the study site.

Section 2.0 discusses soil sampling results and recommendations.

Section 3.0 lists references used in preparation of this report.

#### 1.3 ENVIRONMENTAL SETTING

#### 1.3.1 GEOGRAPHY

PACO Pumps, Inc. is located in Alameda County, within the City of Oakland, approximately 25 miles northeast of the City of San Francisco. The city of Oakland is part of the San Francisco Bay region. The San Francisco Bay region lies between northern latitudes 36° and 39°. and extends across the lowest and narrowest segment of the Coast Ranges in central California. This region covers an area of 7,500 square miles consisting of northwest trending mountain ranges, broad basins, and narrow valleys generally paralleling major geologic structures and the coastline of central California (Helley, 1979).

About 65 percent of the San Francisco Bay consists of rounded hills and rugged mountain uplands with many ridge crests rising above 1,000 feet and a few peaks rising above 4,000 feet. Almost 11 percent of the San Francisco Bay consists of the open water and tidal marshlands adjacent to the Bay. The remaining 24 percent consists of relatively flat lowland areas (generally less than 200 feet above sea level) that constitute the broad alluvial plain surrounding the bay, the broad to narrow valley bottoms extending from the bay plains into the surrounding hills, and the narrow elevated marine terraces cut into the mountains along the Pacific Coast (Helley, 1979).

#### 1.3.1.1 Climate

The San Francisco Bay region has a mediterranean climate with mild west winters and warm dry summers. The climate along the coast is marked by moderate and even temperatures, heavy persistent summer fog, and winds from the west-northwest. In contrast, inland areas have a wide range of temperature and have less wind. Temperatures are influenced by elevation and local topography. Higher summer temperatures and lower winter temperatures occur in low areas isolated by mountainous terrain. This kind of climate is also true of areas far distant from the bay and its temperature-moderating waters. Precipitation is distinctly seasonal, most falling between November and March, very little between June and September. The seasonal distribution of precipitation is largely controlled by the location of the anticyclonic cell that is normally found off the California coast, particularly in the summer. Winter precipitation occurs when this anticyclone is absent or far south of its normal position, which blocks storm systems from the Gulf of Alaska.

Almost all precipitation is in the form of rain and what little snow falls usually soon melts (Helley, 1979). Average annual precipitation ranges from 10-20 inches in the dry interior valleys to 40-60 inches and locally to 80 inches in the high coastal mountains.

The native vegetation in the bay region north through Sonoma County and south through Santa Clara County is dominated by plants adapted to mild climatic conditions including a summer drought. The native plants evolved here in their own ecological niches, which are referred to as biotic communities. These plant communities are bay and salt marsh, freshwater marsh, the open coast, chaparral, grasslands, broad-leaved forest, and coniferous forest. The flatland deposits underlie almost all but the last two communities. However, the broad-leaved and coniferous forests do cover stream terrace deposits in the outer valleys (Helley, 1979).

#### 1.3.2 SOILS

Limited information is available regarding the stratigraphy below the PACO Pumps facility. However, shallow soil stratigraphy at the study site was observed by Jonas & Associates during 1991 soil sampling effort. Soil consisted of gravely fill material from the ground surface to an approximate depth of 1.5 feet. From 1.5 feet to 3.5 feet below the ground surface a well compacted, dark brown clay layer was encountered.

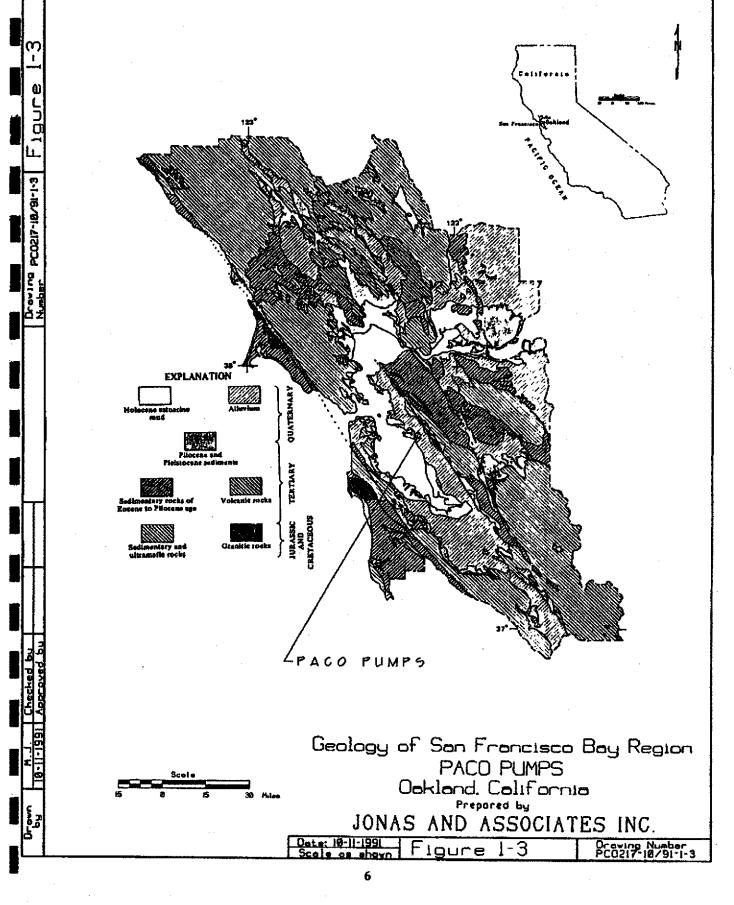
#### 1.3.3 GEOLOGY

The geologic units underlying the East Bay area consist of bedrock of the Franciscan group and an unconsolidated sedimentary sequence (Radbruch, 1957). The bedrock surface beneath San Francisco Bay tilts eastward and becomes deeper towards the south. Figure 1-3 illustrates the geology of the San Francisco Bay Region. As illustrated in this figure, beneath the PACO Pumps property are alluvium formations and Quaternary in age. Published information on the geology of the area occupied by PACO Pumps facility was not found.

The geologic information presented below are obtained from a geologic log of a well drilled on Bay Farm Island, north of Oakland Airport. This information was presented in a report titled Geologic Study, Oakland Outer Harbor, Oakland, California, prepared by Geomatrix Consultants for Port of Oakland, in 1986. According to this report, a well on Bay Farm Island, north of Oakland Airport, encountered bedrock at a depth greater than 1,000 feet. Overlying this bedrock surface is a sequence of sediments that have been subdivided differently by various authors. These various interpretations are presented in Table 1-1.

The unit names proposed by Radbruch (1957) are used for this project because they have wide acceptance among local professionals. These units are described below (Track and Rolston, 1951; Radbruch, 1957; Louderback, 1951; and Helley and others, 1979).

Artificial fill. Manmade. Miscellaneous Bay Mud or sand dredged from the Bay, rock and soil used for fill and construction materials. Permeability generally varies with composition from low to high.



#### Table 1-1 Correlation Chart of Sedimentary Units (Geomatrix, 1986)

Track, P. and Roiston, J.W. (1951)	Treasher (1963)	Radbruch (1957)	Helley and Others (1979)
		Artificial Fill	
Bay Mud	Younger Bay Mud	Bay Mud	Bay Mud
Merritt Sand	San Deposits	Merritt Sand	Merritt Sand
—		Temescal Formation (Non Marine)	Late Pleistocene Alluvium
Track, P. and Rolston, J.W. (1951)	Treasher (1963)	Radbruch (1957)	Helley and Others (1979)
Posey Formation			
San Antonio Formation	Older Bay Mud	Alameda Formation	Early Pleistocene Alluvium (Marine & Non-Marine)
Alameda Formation			
Bedrock	Bedrock	Bedrock	Bedrock

- Bay Mud. Silty clay; sandy, with lenses of sand. Permeability is low; most mud is saturated. Recent young Bay Mud overlies the Merritt Sand along the edge of the Bay or the Alameda Formation where the Merritt has been removed by erosion. Bay Mud underlies most of the Artificial Fill.
- Merritt Sand. Sand; fine-grained, silty, clayey. Slightly indurated. The Merritt Sand is well-sorted, in part showing evidence of wind-blown origin. Permeability of these deposits is high. The sand deposit filled in valleys previously formed on an older erosion surface. Following this deposition the valleys were partly re-excavated. Lake Merritt was formed during this period.

- Temescal Formation. The Merritt Sand grades laterally into the Temescal Formation, which is the non-marine extension of the Merritt sand. The Temescal is an alluvial-fan deposit located on the plains below the steep hill front. The deposit extends toward the Bay and grades into the Merritt Sand. It is thought that the two were deposited at about the same time. The Temescal consists of clayey gravel, sandy silty clay, and sand mixtures. In some places the gravel is as much as 20 feet thick (Radbruch, 1957). Overall, the permeability of this unit is moderate, but some gravel lenses have relatively high permeability.
- Alameda Formation. The Alameda Formation is a marine deposit that commonly contains alternating layers of sandy clay and sand with the sand containing fine gravel. The lower portion of formation appears to be similar to the Santa Clara Formation observed at the surface in the southern part of the Bay Area. Permeability is moderate to low.
- Bedrock of the Franciscan group. The Franciscan Formation is a complex assemblage of serpentinite, greenstone, graywacke, chert, shale, sandstone, and schist, and is found on may ridges and mountains of the San Francisco Bay Region. Permeability is relatively low.

#### 1.3.4 HYDROGEOLOGY

#### 1.3.4.1 <u>Hydrogeologic Units</u>

A library search was performed to obtain information on the hydrogeology of Oakland. Very little information was found. According to the Department of Water Resources, Merritt Sand and the Alameda Formation are potential aquifers within the Oakland area (DWR, 1982). The groundwater level within the area underlaid by the Merritt Sand appears to be within 5 to 20 feet of surface (City of Oakland, 1986). However, specifics as to the quality, yield, and groundwater use from this aquifer is unknown. Unpublished information provided by County of Alameda, Public Works Agency identifies an aquifer in the zone from slightly above sea level to about 100 feet below sea level. It is reported to probably represent the Merritt Sand.

The Alameda Formation is used as a limited source of water within the Oakland area (DWR, 1982, Alameda County, 1986). However, very limited data are available on the number of active wells, recharge rate, yield, groundwater levels, and water quality.

#### SECTION 2.0

#### SOIL SAMPLING ACTIVITIES AND RESULTS

#### 2.1 SOIL SAMPLE COLLECTION PROCEDURES

As stated in Section 1.0 of this report, stained ground surface was observed at the Asphalt/Concrete Area. In order to determine whether release of chemical substances have occurred in this area, two soil samples were collected. Figure 2-1 denotes the soil sampling locations. Soil samples were collected in brass sleeves from the first six inches of the native soil encountered underneath the asphalt/concrete pad. Following the soil collection, the boring locations were plugged with the cuttings and re-asphalted.

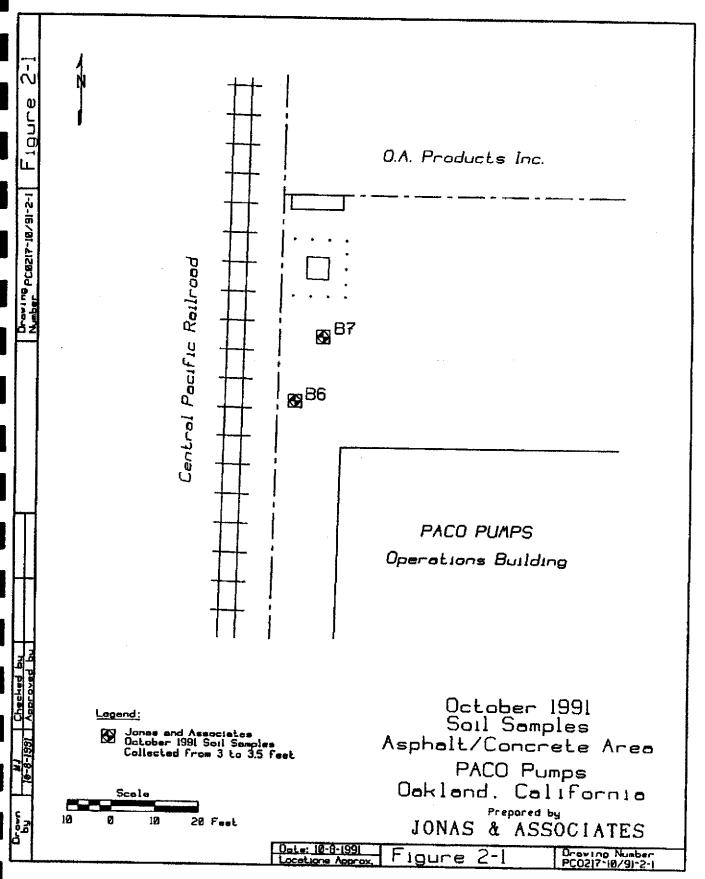
Once collected, samples were placed in an ice chest with ice packs. Each cooler contained sufficient ice and/or ice packs to ensure that a proper temperature of 4° Celsius is maintained. The samples were packed in a manner to decrease the potential for any damage of the sample containers. Field Chain-of-Custody records, completed at the time of sample collection accompanied the samples inside the cooler for shipment to the Chromalab Laboratory. All coolers were delivered to the laboratory by Jonas & Associates field personnel, within 24 hours after sampling. After initially inspecting the samples, the laboratory completed a Sample Condition Report form. This form summarizes the sample conditions as reported by the laboratory. Both samples were analyzed for the following parameters:

- Extractable Petroleum Hydrocarbons (EPH) EPA Method 3550/8015;
- » Volatile Petroleum Hydrocarbons (VPH) EPA Method 5030/8015;
- Volatile Organic Compounds (VOCs) EPA Method 8240;
- Semi-Volatile Organic Compounds (Semi-VOCs) EPA Method 8270;
- Pesticides and Polychlorinated Biphenyl EPA Method 8080; and
- Title 22 Metals EPA Method 6010 and 7000 series.

#### 2.1.1 <u>DECONTAMINATION AND POST-SAMPLING PROCEDURES</u>

Decontamination of equipment used for sampling took place in a specific decontamination zone designated at the site. Sampling equipment was decontaminated prior to initial use and at the completion of sampling activities. A manual scrubbing to remove foreign material followed by a thorough wash with a non-phosphate soap, was used to decontaminate all the equipment between and prior to sampling. All non-disposable equipment was decontaminated according to the procedures summarized below:

- » Manual scrub without water or soap
- Manual scrub and wash with a non-phosphate soap solution



- » Tap water rinse
- Distilled water rinse
- Air dry
- » Distilled water rinse

#### 2.2 SAMPLE DOCUMENTATION

Sample documentation included field logbooks, sample labels, Chain-of-Custody records, and sample condition report forms. All field documentation was written legibly in waterproof ink. Errors was crossed out with a single line, initialed, and dated.

#### 2.2.1 SAMPLE IDENTIFICATION NUMBERS

Each sample was assigned a unique identification number that allow retrieval of information regarding the sample. The sample identification number consisted of three main segments, which were separated by a hyphen. The first segment is made up of the letter "B" and a one or two-digit number representing the boring number. The second segment represents the sampling depth. The third segment represents the sampling date.

Example: B1-0.5-10191

Boring one, collected at a depth of 0.5 feet on October 1, 1991.

#### 2.2.2 FIELD LOGBOOKS

A project field logbook was used during the field effort to document the following:

- » Date and time of log entries:
- Field conditions (weather, terrain, hazards, etc.);
- Personnel present during field operations;
- Waste containment procedures, and a daily inventory of wastes present onsite;
- Field measurements taken, instrumentation used, and frequency of instrument calibration;
- Information recorded on sample labels, as well as the site identification number and the sampling depth;
- Any unusual sample characterization:
- Other specific considerations pertaining to sample acquisition; and
- Boring data (depth, lithology, and etc.).

#### 2.2.3 SAMPLE LABELS

A pre-printed adhesive label was affixed to each sample (Figure 2-2). The information below was written on every sample label;

- » Project number;
- Sample identification number;
- Date and time of sampling:
- Name of sample collector:
- Type of analysis to be performed.

#### FIGURE 2-2 Sample Label

CHROMALAB, INC. 2239 Omega Road #1 San Ramon, CA 94583

PROJECT NO	DATE
BORING NO	TIME
SAMPLE NO	DEPTH
SAMPLE TYPE	
PRESERVATIVES	
TYPE OF ANALYSIS	
NAME OF COLLECTOR	

#### 2.2.4 CHAIN-OF-CUSTODY RECORDS / SAMPLE CONDITION REPORT FORM

A Chain-of-Custody record accompanied all samples when they were shipped to the analytical laboratory. A copy of the Chain-of-Custody form is presented in Figure 2-3. The Chain-of-Custody record documents transfer of samples from one party to another.

Additional information that were noted on the form were as follows:

- » Project number;
- Sample identification number;
- Date and time of sampling;
- » Type of sample; and
- Type of analysis to be performed.

# CHROMALAB, INC.

2239 Omega Road, #1 • San Ramon, California 94583 415/831-1788 • Facsimile 415/831-8798

## **Chain of Custody**

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Figure 2-4 provides a copy of a Sample Condition Report form. The form was completed by the laboratory at the time the samples were received. This form summarizes sample conditions. Information which were noted on this form include:

- Date of request;
- Job number;
- » Client;
- Source of samples;
- » Shipper;
- » Arrival date:
- Sample number;
- Condition received at laboratory; and
- Specific analyses requested.

#### 2.3 SAMPLE SHIPMENT

The Jonas and Associates Technical Manager notified the Project Chemist at the Chromalab analytical laboratory a week prior to sampling. This provided an opportunity for the laboratory to anticipate the arrival of the samples. In addition, coolers and sample bottles was sent from the laboratory to Jonas and Associates.

Within 24 hours after the samples were collected, Jonas and Associates transported all samples to the analytical laboratory. Samples were packaged for shipment in a cooler chilled with bags of ice. Foam padding was used to protect sample containers. The original Chain-of-Custody record was placed in a plastic pouch affixed to the inside lid of the cooler. The Field Manager retained a copy of the form. To secure the cooler during transport, the lid was sealed with tape. When possession of the samples is transferred, the individuals relinquishing and accepting custody wrote their names, the names of their organizations, and the time and date of custody transfer on the Chain-of-Custody record.

#### Figure 2-4 Sample Condition Report Form

#### SAMPLE CONDITION REPORT FORM

Pageof Date of Request_	Job No	<u>.                                    </u>	Analysis Authorized by					
Client				Phone				
Source of Sample Shipper	Arrival D	ate	Repor	Phone				
Lab # (Lab to fill in)	Sample Type <sup>1</sup>	Identify Information (e.g. site taken, field I.D. number, etc.)	Condition Received at Laboratory <sup>2</sup>	Date of Collection	Specific Analyses Requested			
			1. Custody Form?					

2. Package O.K.? 3. In Ice?

Remarks:

4. Container Type Suitable?

6. Proper preservation? 7. Custody Seal?

5. Air bubbles present/absent in VOA?

Specify as water (W), soil (S), or other (describe on back if need to).

Questions are to be answered (Y) or no (N) if possible. Special concerns (re, condition of sample, detection limits desired, important information re, sample history, etc.) should be noted under remarks. information re. sample history, etc.) should be noted under remarks.

#### SECTION 3.0

#### 1991 SOIL SAMPLING RESULTS

#### 3.1 SOIL SAMPLING RESULTS

Laboratory analytical results for these samples are summarized in Tables 3-1, 3-2, 3-3, 3-4, and 3-5. As illustrated in these tables, Extractable Petroleum Hydrocarbons, Volatile Petroleum Hydrocarbons, Volatile Organic Compounds, Semi-Volatile Organic Compounds, and Pesticides were not detected above the method detection limits in both of the soil samples. Polychlorinated Biphenyi (PCB 1260) was detected in both samples B6 and B7 at concentrations of 0.40 mg/kg and 0.67 mg/kg, respectively.

A source for PCB found in the soil is not yet identified. According to 40 CFR, Part 761, Subpart G - PCB Spill Cleanup Policy:

"PCB contaminated soil should be removed to 10 mg/kg, provided that soil is excavated to a minimum depth of 10 inches. The excavated soil should be replaced with clean soil (less than 1 mg/kg PCBs)."

The PCB detected in this area is well below 10 mg/kg. In addition, PCB concentrations detected in this area also are below the Total Threshold Limit Concentration Value of 50 mg/kg.

Metals were detected in both of the samples. Barium (B6:10.7 mg/kg; B7:138 mg/kg), Cadmium (B6:0.595 mg/kg; B7:1.99 mg/kg), Cobalt (B6:0.804 mg/kg; B7:0.991 mg/kg), Chromium (B6:2.63 mg/kg; B7:57.8 mg/kg), and Nickel (B6:1.28 mg/kg; B7:348 mg/kg) were detected in both of the samples. Lead and Zinc were detected in sample B6 at concentrations of 85.4 mg/kg and 22.5 mg/kg, respectively. Molybdenum and Vanadium were detected in Sample B7 at concentration levels of 8.85 mg/kg and 146 mg/kg, respectively.

None of the metals exceeded the Total Threshold Limit Concentrations (TTLC). These limits are listed in Table 3-5.

Since the detected PCB and metals are below the required regulatory levels, cleanup of this area may not be required. However, the final determination must be made by the regulatory agencies. It is recommended that a copy of this report be submitted to the Alameda County Health Care Services Agency for their review and approval. Their address is as follows:

Alameda County Health Care Services Agency Department of Environmental Health Hazardous Materials Program 80 Swan Way, Room 200 Oakland, California 94621

Attention: Mr. Barney M. Chan, Hazardous Materials Specialist

Table 3-1 Gasoline/BTEX and Diesel Analysis Sampling Date October 1, 1991 Analysis Date October 4-7, 1991

Sample Number	Gasoline (Method 5030/8015) (1.0 mg/kg) <sup>3</sup> (mg/kg)	Diesel (Method 3550/8015) (1.0 mg/kg) (mg/kg)	Benzene (Method 8020) (0.005 mg/kg) (mg/kg)	Toluene (Method 8020) (0.005 mg/kg) (mg/kg)	Ethylbenzene (Method 8020) (0.005 mg/kg) (mg/kg)	Total Xylenes (Method 8020) (0.005 mg/kg) (mg/kg)
B6-0.5-1019 <sup>4</sup>	ND <sup>6</sup>	ND	ND	ND	ND	ND
B7-0.5-1019	ND	ND	ND	ND	ND	ND

B6 = Soil sampling boring number one 0.5 = Sample depth in Feet

10191 = Sampling Date October 1, 1991 (Laboratory reports deleted the last number "1" due to insufficient space in the column)

<sup>(1.0</sup> mg/kg) = Method Detection Limit

B6-0.5-10191

ND - Not Detected Above the Reported Detection Limit.

Table 3-2 Volatile Organic Compounds Date Sampled October 1, 1991 Date Analyzed October 4-8, 1991

Compound Name	Volatile Organic Compounds, EPA Method Detection Limit: 0.005 mg/kg				
	B6-0.5-10191	B7-0.5-10191			
V	_				
Chloromethane	NEO <sup>6</sup>	ND			
Vinyl Chloride	ND	ND			
Bromomethane	ND	ND			
Chloroethane	ND	ND			
Trichlorofluoromethane	ND	ND			
1,1-Dichloroethene	ND	ND			
Methylene Chloride	ND	ND			
1,2-Dichloroethene (Total)	ND	ND			
1,1-Dichloroethane	ND	ND			
Chloroform	ND	ND			
1,1,1-Trichloroethane	ND	ND			
Carbon Tetrachloride	ND	ND			
1,2-Dichloroethane	ND	ND			
Benzene	ND	ND			
Trichloroethene	ND	ND			
1.2-Dichloropropaue	ND	ND			
Bromodichloromethane	ND	ND			

PCC001304

<sup>&</sup>lt;sup>6</sup> ND = Not Detected Above the Reported Detection Limit.

# Table 3-2 (Continued) Volatile Organic Compounds Date Sampled October 1, 1991 Date Analyzed October 4-6, 1991

Compound Name	Volattie Organic Compounds, EPA Method 824 Detection Limit: 0.005 mg/kg				
	B6-0.5-10191	B7-0.5-10191			
2-Chloroethylvinylether	ND	ND			
Trans-1,3-Dichloropropene	ND	ND			
Toluene	ND	ND			
Cis-1,3-Dichloropropene	ND	ND			
1,1,2-Trichloroethane	ND	ND			
Tetrachkoroethene	ND	ND			
Dibromochloromethane	ND	ND			
Chlorobenzene	ND	ND			
Ethylbenzene	ND	ND			
Bromoform	ND	ND			
1,1,2,2-Tetrachloroethane	ND	ND			
1,3-Dichlorobenzene	ND	ND			
1,4-Dichlorobenzene	ND	ND			
1,2-Dichlorobenzene	ND	ND			
Total Xylenes	ND	ND			
Acetone	ND	ND			
Methyl Ethyl Ketone	ND	ND			
Methyl Isobutyl Ketone	ND	ND			

Table 3-3
Base/Neutrals/Acids-Soil
Date Sampled October 1, 1991
Date Analyzed October 4-8, 1991

Compound Name	Base/Neutrals/Acids-Soil, EPA Method 8270				
	B6-0.5-10191	B7-0.5-10191			
	7	:			
Phenol	ND(0.05) <sup>7</sup>	ND(0.05)			
Bis(2-Chloroethyl) Ether	ND(0.05)	ND(0.05)			
2-Chlorophenol	ND(0.05)	ND(0.05)			
1.3-Dichlorobenzene	ND(0.05)	ND(0.05)			
1,4-Dichlorobenzene	ND(0.05)	ND(0.05)			
Benzyl Alcohol	ND(0.10)	ND(0.10)			
1,2-Dichlorobenzene	ND(0.05)	ND(0.05)			
2-Methylphenol	ND(0.05)	ND(0.05)			
Bis(2-Chioroisopropyl)ether	ND(0.05)	NIX(0.05)			
4-Methylphenol	ND(0.05)	ND(0.05)			
N-Nitroso-Di-N-Propylamine	ND(0.05)	ND(0.05)			
Hexachloroethane	ND(0.05)	ND(0.05)			
Nitrobenzene	ND(0.05)	ND(0.05)			
Isophorone	ND(0.05)	ND(0.05)			
2-Nitrophenol	ND(0.05)	ND(0.05)			
2,4-Dimethylphenol	ND(0.05)	ND(0.05)			
Benzole Acid	ND(0.25)	ND(0.25)			

<sup>&</sup>lt;sup>7</sup> ND = Not Detected Above the Reported Detection Limit.

# Table 3-3 (Continued) Base/Neutrals/Acids-Soti Date Sampled October 1, 1991 Date Analyzed October 4-8, 1991

Compound Name	Base/Neutrals/Acids-Soil, EPA Method 8270			
	B6-0.5-10191	B7-0.5-10191		
Bis-(2-Chloroethoxy)Methane	ND(0.05)	ND(0.05)		
2.4-Dichiorophenol	ND(0.05)	ND(0.05)		
1,2,4-Trichlorobenzene	ND(0,05)	ND(0.05)		
Naphthalene	ND(0.05)	ND(0.05)		
4-Chloroaniline	ND(0.10)	ND(0.10)		
Hexachlorobutadiene	ND(0.05)	ND(0.05)		
4-Chloro-3-Methylphenol	ND(0.10)	ND(0.10)		
2-Methylnaphthalene	ND(0.05)	ND(0.05)		
Hexachlorocyclopentadiene	ND(0.05)	ND(0.05)		
2,4,6-Trichkorophenol	NID(0.05)	ND(0.05)		
2,4,5-Trichlorophenol	ND(0.05)	ND(0.05)		
2-Chloronaphthalene	ND(0.05)	ND(0.05)		
2-Nitroaniline	ND(0.25)	ND(0.25)		
Dimethyl Phthalate	ND(0.05)	ND(0.05)		
Acenaphthylene	ND(0.05)	ND(0.05)		
3-Nitroaniline	ND(0.25)	ND(0.25)		
Acenaphthene	ND(0.05)	ND(0.05)		
2,4-Dinitrophenol	ND(0.25)	ND(0.25)		
4-Nitrophenol	ND(0.25)	ND(0.25)		

# Table 3-3 (Continued) Base/Neutrals/Acids-Soil Date Sampled October 1, 1991 Date Analyzed October 4-8, 1991

Compound Name	Base/Neutrals/Acids-Soil, EPA Method 8270				
	B6-0.5-10191	B7-0.5-10191			
Dibenzofuran	ND(0.05)	ND(0.05)			
2,4-Dinitrotoluene	ND(0.05)	ND(0.05)			
2,6-Dinitrotoluene	ND(0.05)	ND(0.05)			
Diethyl Phthalate	ND(0.05)	ND(0.05)			
4-Chloro-Phenyl Phenyl Ether	ND(0.05)	ND(0.05)			
Fluorene	ND(0.05)	ND(0.05)			
4-Nitroaniline	ND(0.25)	ND(0.25)			
4-6-Dinitro-2-Methyl Phenol	ND(0.25)	ND(0.25)			
N-Nitrosodiphenylamine	ND(0.05)	ND(0.05)			
4-Bromophenyl Phenyl Ether	ND(0.05)	ND(0,05)			
Hexachlorobenzene	ND(0.05)	ND(0.05)			
Pentachlorophenol	ND(0.25)	ND(0.25)			
Phenanthrene	ND(0.05)	ND(0.05)			
Anthracene	ND(0.05)	ND(0.05)			
Di-N-Butyl Phthalate	ND(0.05)	ND(0.05)			
Fluoranthene	ND(0.05)	ND(0.05)			
Рутеле	ND(0.05)	ND(0.05)			
Butylbenzylphthalate	ND(0.05)	ND(0.05)			
3,3'-Dichlorobenzidine	ND(0.10)	ND(0.10)			

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See Enclosed Report.			
8			

#### SLIC PROGRAM

#### **NEW CASE DATA SHEET**

#### 1. Discharger Information

Name: PACO Pumps, Inc.
Location of sit: 9201 San Leandro Street, Oakland, California 94603-1237
Telephone number of Site or Consultant:

Site: (415) 639-3323 Consultants - Jonas & Associates Inc.: (510) 676-8554

- 2. Proposition 65 Reporting (check one)
- X No report deemed necessary (SL or DC initials)
- \_\_ Date Report completed
- 3. Site Characteristics

State specific levels of pollutants in soil and groundwater (e.g., 58 ppm TCE in soil and 5 ppm in groundwater)

Concentrations found in soil:

Barium:10.7 mg/kg - 138 mg/kg; Cadmium: 0.595 mg/kg - 1.99 mg/kg; Cobalt:0.804 mg/kg - 0.991 mg/kg; Chromium:2.63 mg/kg - 57.8 mg/kg; Nickel:1.28 mg/kg - 348 mg/kg; Lead:85.4 mg/kg; Zinc:22.5 mg/kg; Molybdenum:8.85 mg/kg; Vanadium:146 mg/kg.
PCB: 0.40 mg/kg - 0.67 mg/kg.

#### 4. Comments

State how the information was received. State any plans for additional study. State any staff opinions as to whether additional study is required. Add any additional screening information such as "large site", "historic problem", "Federal facility", etc.

Soil Characterization report dated October 30, 1991 submitted by PACO Pumps, Inc.

#### 5. Lead Agency

State whether any agency has assumed the lead role for investigation and/or cleanup.

Alameda County Health Care Services Agency, Department of Environmental Health, Hazardous Materials Program.

#### 6. Status

Regional Board assignment status (e.g., Unassigned or RB Lead). If RB lead, give name of staff person.

Unassigned.

# Table 3-3 (Continued) Base/Neutrals/Acids-Soil Date Sampled October 1, 1991 Date Analyzed October 4-8, 1991

Compound Name	Base/Neutrals/Acids-Soil, EPA Method 8270		
	B6-0.5-10191	B7-0.5-10191	
Benzo(A)Anthracene	ND(0.05)	ND(0.05)	
Bis(2-Ethylhexyl)Phthalate	ND(0.05)	ND(0.05)	
Chrysene	ND(0.05)	ND(0.05)	
Di-N-Octylphthalate	ND(0.05)	ND(0.05)	
Benzo(B)Fluoranthene	ND(0.05)	ND(0.05)	
Benzo(K)Fluoranthene	ND(0.05)	ND(0.05)	
Benzo(A)Pyrene	ND(0.05)	ND(0.05)	
Indeno(1,2,3 C,D)Pyrene	ND(0.05)	ND(0.05)	
Dibenzo(A,H)Anthracene	ND(0.05)	ND(0.05)	
Benzo(G,H,I)Perylene	ND(0.05)	ND(0.05)	

### Table 3-4 Chlorinated Pesticides and PCB Analysis<sup>8</sup> Date Sampled October 1, 1991 Date Analyzed October 4-8, 1991

Samula	ato in Constitution	Chlorinated Pesticides and PCBs (EPA Method 8080) - Unit in Mg/kg																		
Sample #	Aldrin	Dieldrin	Endrin Aldehyde	:	Heptachlor		DDT		P,P DDD	Endo sulfan I		α-	β-	γ-	BHC	Endo sulfan Sulfate		phene		Chlor- dane
B6-0.5- 10191	ND	ND	ND	ND	ND	ND	ND	ND	ND	מא	ND	ND	ND	ND	ND	ND	ND	ND	0.40	ND
B7-0.5- 10191	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	NID	ND	0.67	ND

METHOD DETECTION LIMITS: Aldrin (0.01 mg/kg); Dieldrin (0.01 mg/kg); Endrin Aldehyde (0.05 mg/kg); Endrin (0.01 mg/kg); Heptachlor (0.01 mg/kg); Heptachlor Epoxide (0.01 mg/kg); P,P' - DDT (0.05 mg/kg); P,P' - DDE (0.01 mg/kg); P,P' - DDD (0.05 mg/kg); Endosulfan II (0.05 mg/kg); α - BHC (0.01 mg/kg); β - BHC (0.01 mg/kg); γ - BHC (0.01 mg/kg); δ - BHC (0.01 mg/kg); Endosulfan Sulfate (0.1 mg/kg); P,P' - Methoxychlor (0.1 mg/kg); Toxaphene (0.1 mg/kg); PCB's (0.1 mg/kg); Chlordane (0.1 mg/kg). Toxaphene (0.1 mg/kg); PCB's (0.1 mg/kg); Chlordane (0.1 mg/kg).

Table 3-5
Title 22 Metais - Soii
Date Sampled October 1, 1991
Date Analyzed October 4-8, 1991

Parameters	EPA Method 3050/6010 Detection Limits (mg/kg)	Sample Number B6-0.5-10191	Sample Number B7-0.5-10191	Total Threshold Limit Concentration (TTLC)
Silver (Ag)	0.004	ND°	ND	500
Arsenic (As)	0.088	ND	ND	500
Barium (Ba)	0.005	10.7	138	10000
Beryllium (Be)	0.001	ND	ND	75
Cadmium (Cd)	0.012	0.595	1.99	100
Cobalt (Co)	0.020	0.804	0.991	8000
Chromium (Cr)	0.006	2.63	57.8	2500
Copper (Cu)	0.004	67.6	ND	2500
Mercury (Mg)	0.200	ND	ND	20
Molybdenum (Mo)	0.940	ND	8.85	3500
Nickel (Ni)	0.026	1.28	348	2000
Lead (Pb)	0.044	85.4	ND	1000
Antimony (Sb) <sup>10</sup>	0.040	ND	ND	500
Selenium (Se)	0.200	ND	ND	100
Thallium (Tl)	0.088	ND	ND	700
Vanadium (V)	0.004	ND	146	2400
Zine (Zn)	0.006	22.5	ND	5000

ND - Not Detected Above the Method Detection Limit

<sup>10</sup> Method of Analysis: 3005/6010

### **SECTION 4.0**

#### REFERENCES

- Atwater, B., Hedel, C. and Helley, E., 1977, Late Quaternary Depositional History, Holocene Sea-Level Changes, and Vertical Crustal Movement, South San Francisco Bay, California U.S.G.S., Professional Paper 1014, U.S. Government Printing Office, Washington.
- California Department of Health Service, Hazardous Waste Management Branch, Toxic Substances Control Program, <u>Guidance for PCB Storage Facility Closure Plan</u>, March 1991.
- 3. County of Alameda, Public Works Agency, unpublished material on <u>Oakland Upland and Alluvial and Merritt San Outcrop</u>.
- 4. Dames & Moore, August 12, 1987, Letter Report, Site Contamination Study, PACO Pumps Facility, Oakland, California, for Amsted Industries, Inc.
- 5. Helley, E. J. and Lajole, K.R., 1979, Flatland deposits of the San Francisco Bay Region, California their geology and engineering properties, and their importance to comprehensive planning, Geological Survey Professional Paper 943, U.S. Government Printing Office, Washington.
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- 10. U.S. EPA, Office of Toxic Substance, <u>TSCA Guidance Manual for Commercial PCB Storage Facility Applications</u>, October 18, 1989.

### Jonas & Associates Inc.

ATTACHMENT A
SOIL SAMPLING
ANALYTICAL LABORATORY RESULTS
CHROMALAB ANALYTICAL LABORATORY
October 1, 1991

**5 DAYS TURNAROUND** 

Analytical Laboratory (E694)

October 9, 1991

ChromaLab File No.: 1091017

JONAS & ASSOCIATES

Attn: R. Jonas

RE: Seven soil samples for Gasoline/BTEX and Diesel analysis

Project Name: PACO PUMPS SOIL

Date Sampled: Oct. 1, 1991

Date Submitted: Oct. 2, 1991 Date Extracted: Oct. 4-7, 1991 Date Analyzed: Oct. 4-7, 1991

### RESULTS:

Sample I.D.	Gasoline (mg/kg)	Diesel (mg/kg)	Benzene (µg/kg)	Toluene (µg/kg)	Ethyl Benzene (ug/kg)	Total Xylenes (µg/kg)
B1-3.5-1019 B2-3.5-1019 B3-3.5-1019 B4-3.5-1019 B5-3.5-1019 B6-0.5-1019 B7-0.5-1019	N.D. N.D. N.D. N.D. N.D. N.D.	N.D. N.D. N.D. N.D. N.D. N.D.	N.D. N.D. N.D. N.D. N.D. N.D.	N.D. N.D. N.D. N.D. N.D. N.D.	N.D. N.D. N.D. N.D. N.D. N.D.	N.D. N.D. N.D. N.D. N.D. N.D.
BLANK SPIKE REC. DUP SPIKE REC DET. LIMIT METHOD OF ANALYSIS	N.D. 101.3% 98.0% 1.0 5030/ 8015	N.D. 93.0% 100.7% 1.0 3550/ 8015	N.D. 95.4% 89.3% 5.0	N.D. 103.9% 88.3% 5.0	N.D. 94.6% 89.5% 5.0	N.D. 103.1% 90.8% 5.0

ChromaLab, Inc.

David Duong Chief Chemist

Erretam (by 10) Eric Tam Laboratory Director

Analytical Laboratory (E694)

**5 DAYS TURNAROUND** 

October 09, 1991

ChromaLab File # 1091017 F

Client: Jonas & Associates, Inc.

Date Sampled: Oct. 01, 1991 Date Analyzed: Oct. 08, 1991

Attn: Romena Jonas

Date Submitted: Oct. 02, 1991

Project Name: PAGO PUMPS SOIL

Sample I.D.: B6-0.5-1019

Method of Analysis:

8240

Detection Limit: 5.0 µg/kg

COMPOUND NAME	ug/kg	Spike Recovery
CHLOROMETHANE	N.D.	their olds plant
VINYL CHLORIDE	N.D.	that was may
BROMOMETHANE	N.D.	Aller right plays
CHLOROETHANE	N.D.	~~~
TRICHLOROFLUOROMETHANE	N.D.	= = =
1,1-DICHLOROETHENE	N.D.	91.8% 93.5%
METHYLENE CHLORIDE	N.D.	Anne date
1,2-DICHLOROETHENE (TOTAL)	N.D.	
1,1-DICHLOROETHANE	N.D.	
CHLOROFORM	N.D.	·
1,1,1-TRICHLOROETHANE	N.D.	THE RES 148
CARBON TETRACHLORIDE	N.D.	- 1e w
1,2-DICHLOROETHANE	N.D.	حد سه بله
BENZENE	N.D.	94.7% 96.0%
TRICHLOROETHENE	N.D.	
1,2-DICHLOROPROPANE	N.D.	
BROMODICHLOROMETHANE	N.D.	anne see pain
2-CHLOROETHYLVINYLETHER	N.D.	
TRANS-1,3-DICHLOROPROPENE	N.D.	The size day
TOLUENE	N.D.	92.5% 93.6%
CIS-1,3-DICHLOROPROPENE	N.D.	32.13% 33.0%
1,1,2-TRICHLOROETHANE	N.D.	
TETRACHLOROETHENE	N.D.	
DIBROMOCHLOROMETHANE	N.D.	
CHLOROBENZENE	N.D.	91.0% 91.8%
ETHYLBENZENE	N.D.	מאט, ויצי מאט. ויצי
BROMOFORM		
1,1,2,2-TETRACHLOROETHANE	N.D.	440k dam 4015
1,3-DICHLOROBENZENE	N.D.	
1,4-DICHLOROBENZENE	N.D.	red who are
1,2-DICHLOROBENZENE	N.D.	
	N.D.	TO 400 AND
TOTAL XYLENES	N.D.	alian valle vann
ACETONE METHYL ETHYL KETONE	N.D.	
METHYL ISOBUTYL KETONE	и.о.	
THE TOORDITE RETURE	N.D.	<b>4 5 4</b>

Chromatab, Inc.

Dayld Duong Senior Chemist

Eric Tam Lab Director

5 DAYS TURNAHOUND

## CHROMALAB, INC.

Analytical Laboratory (E694)

October 09, 1991

Chromatab File # 1091017 G

Client: Jonas & Associates, Inc.

Attn: Pomena Jonas

Date Sampled: Oct. 01, 1991

Date Submitted: Oct. 02, 1991

Date Analyzed: Oct. 08, 1991

Project Name: PACO PUMPS SOIL Sample 1.D.:

B7-0.5-1019

Method of Analysis: 8240 Detection Limit: 5.0 µg/kg

COMPOUND NAME	us/kg	Spike Recovery
CHLOROMETHANE	N.D.	, was 10 mm
VINYL CHLORIDE	N.D.	~
BROMOMETHANE	N.D.	
CHLOROETHANE	N.D.	<b>□ #</b> 44k
TRICHLOROFLUOROMETHANE	N.D.	
1,1-DICHLOROETHENE	N.D.	91.8% 93.5%
METHYLENE CHLORIDE	N.D.	# <b>*</b>
1,2-DICHLOROETHENE (TOTAL)	N.D.	<b>-</b>
1,1-DICHLOROETHANE	N.D.	=
CHLOROFORM	N.D.	
1,1,1-TRICHLOROETHANE	N.D.	~~-
CARBON TETRACHLORIDE	N.D.	
1,2-DICHLOROETHANE	N.D.	
BENZENE	N.D.	94,7% 96,0%
TRICHLOROETHENE	N.D.	<b></b>
1,2-DICHLOROPROPANE	N.D.	پېچ. منعب ننعتر
BROMODICHLOROMETHANE	N.D.	
2-CHLOROETHYLVINYLETHER	N.D.	/44 de la
TRANS-1,3-DICHLOROPROPENE	N.D.	<b>*</b> ** **
TOLUENE	N.D.	92.5% 93.6%
CIS-1,3-DICHLOROPROPENE	N.D.	
1,1,2-TRICHLOROETHANE	N.D.	
TETRACHLOROETHENE	N.D.	e= == ==
DIBROMOCHLOROMETHANE	N.D.	
CHLOROBENZENE	N.D.	91.0% 91.8%
ETHYLBENZENE	N.D.	An ==
BROMOFORM	N.D.	±
1,1,2,2-TETRACHLOROETHANE	N.D.	W
1,3-DICHLOROBENZENE	N.O.	
1,4-DICHLOROBENZENE	N.D.	ert aten ann
1,2-DICHLOROBENZENE	N.D.	
TOTAL XYLENES	N.D.	and apple days
ACETONE	N.D.	
METHYL ETHYL KETONE	N.D.	yet date das
METHYL ISOBUTYL KETONE	N.D.	عيد بعد يت

ChromaLab, Inc.

Day id Buong

Sentor Chemist

Eric Tam Lab Director

2239 Omega Road, #1 . San Ramon, California 94583 510/831-1788 • Facsimile 510/831-9798 Federal ID #68-0140157

**5 DAYS TURNAROUND** 

Analytical Laboratory (E694)

October 10, 1991

ChromaLab File # 1091017 F

Client:

Jonas & Associates, Inc.

Attn:

Romena Jonas

Date Sampled:

Oct. 01, 1991

Date Submitted: Oct. 02, 1991

Date Extracted: Oct. 07, 1991

Date Analyzed: Oct. 9, 1991

Project Name: PACO PUMPS SOIL Sample I.D.: 86-0.5-10191

Method of Analysis: EPA 8270

Matrix: soil

	C1_	1451	, 11
COMPOUND NAME	Sample mg/Kg	MDL.	Spike
PHENOL	N.D.	ma/Ka	Recovery
BIS(2-CHLOROETHYL) ETHER	N.D.	0.05	******
2-CHLOROPHENOL	N.D.	0.05	91.5% 89.2%
1,3-DICHLOROBENZENE	N.D.	0.05	فبو بيش يبها شيد
1,4-DICHLOROBENZENE	N.D.	0.05	
BENZYL ALCOHOL	N.D.	0.05	No Più sab dhi sab
1,2-DICHLOROBENZENE	N.D.	0.10	STEP TOTAL copy about days
2-METHYLPHENOL	N.D.	0.05 0.05	
BIS(2-CHLOROISOPROPYL)ETHER	N.D.		
4-METHYLPHENOL	N.D.	0.05	<del>** ** ** **</del>
N-NITROSO-DI-N-PROPYLAMINE	N.D.	0.05	
HEXACHLOROETHANE	N.D.	0.05	
NITROBENZENE	N.D.	0.05	
ISOPHORONE	,	0.05	
2-NITROPHENOL	N.D.	0.05	
2,4-DIMETHYLPHENOL	N.D.	0.05	
BENZOIC ACID	N.D.	0.05	
BIS(2-CHLOROETHOXY)METHANE	N.D.	0.25	
2,4-DICHLOROPHENOL	N.D.	0.05	91.5% 93.2%
1,2,4-TRICHLOROBENZENE	N.D.	0.05	## ## ## ## ##
NAPHTHALENE	N.D.	0.05	~~~~
4-CHLOROANILINE	N.D.	0.05	
HEXACHLOROBUTADIENE	N.D.	0.10	
4-CHLORO-3-METHYLPHENOL	N.D.	0.05	
2-METHYLNAPHTHALENE	N.D.	0.10	
HEXACHLOROCYCLOPENTAD   ENE	N.D.	0.05	की गर्म हमा का का
2,4,6-TRICHLOROPHENOL	N.D.	0.05	
2,4,5-TRICHLOROPHENOL	N.D.	0.05	~ ~ ~ ~ ~
2-CHLORONAPHTHALENE	N.D.	0.05	
2-NITROANILINE	N.D.	0.05	
DIMETHYL PHTHALATE	N.D.	0.25	
ACENAPHTHYLENE	N.D.	0.05	<del></del>
	N.D.	0.05	
3-NITROANILINE .	N.D.	0.25	~=
ACENAPHTHENE	N.D.	0.05	87.5% 88.2%
2,4-DINITROPHENOL	N.D.	0.25	
4-NITROPHENOL	N.D.	0.25	44 (49) by ma ma
DIBENZOFURAN	N.D.	0.05	
(continued on next page)			

Analytical Laboratory (E694)

**5 DAYS TURNAROUND** 

Page 2

ChromaLab File # 1091017 F

Project Name: PACO PUMPS SOIL Sample 1.D.: B6-0.5-10191 Method of Analysis: EPA 8270

Matrix: soil

	•		
	Sample	MDL	Spike
COMPOUND NAME	mg/Kg	mg/Kg	Recovery
2,4-DINITROTOLUENE	N.D.	0.05	
2,6-DINITROTOLUENE	N.D.	0.05	89.1% 87.1%
DIETHYL PHTHALATE	и.D.	0.05	بيب څنه سه مله
4-CHLORO-PHENYL PHENYL ETHER	N.D.	0.05	
FLUORENE	N.D.	0.05	
4-NITROANILINE	N.D.	0.25	
4,6-DINITRO-2-METHYL PHENOL	N.D.	0.25	
N-NITROSODIPHENYLAMINE	N.D.	0.05	
4-BROMOPHENYL PHENYL ETHER	N.D.	0.05	
HEXACHLOROBENZENE	N.D.	0.05	
PENTACHLOROPHENOL	N.D.	0.25	
PHENANTHRENE	N.D.	0.05	
ANTHRACENE	N.D.	0.05	***
DI-N-BUTYL PHTHALATE	N.D.	0.05	
FLUORANTHENE	N.D.	0.05	***
PYRENE	N.D.	0.05	
BUTYLBENZYLPHTHALATE	N.D.	0.05	
3,3'-DICHLOROBENZIDINE	N.D.	0.10	
BENZO(A)ANTHRACENE	N.D.	0.05	
BIS(2-ETHYLHEXYL)PHTHALATE	N.D.	0.05	
CHRYSENE	N.D.	0.05	86.1% 85.1%
DI-N-OCTYLPHTHALATE	N.D.	0.05	war was story such such
BENZO(B)FLUORANTHENE	N.D.	0.05	
BENZO(K)FLUORANTHENE	N.D.	0.05	خصد منذ عمل الله على
BENZO(A)PYRENE	N.D.	0.05	<b>**</b> ** ** **
INDENO(1,2,3 C,D)PYRENE	N.D.	0.05	
DIBENZO(A,H)ANTHRACENE	N.D.	0.05	*** *** *** ***
BENZO(G,H,I)PERYLENE	N.D.	0.05	

ChromaLab, Inc.

Oavid buong

Senior Chemist

Eric Tam Lab Director

**5 DAYS TURNAROUND** 

Analytical Laboratory (E694)

October 10, 1991

ChromaLab File # 1091017 G

Client: Jonas & Associates, Inc.

Romena Jonas

Date Sampled:

Oct. 01, 1991 Date Extracted: Oct. 07, 1991

Date Analyzed: Oct. 9, 1991

Date Submitted: Oct. 02, 1991

Project Name: PACO PUMPS SOIL

Sample 1.D.: B7-0.5-10191

Method of Analysis: EPA 8270

Matrix: soil

	Sample	MDL	Spike
COMPOUND NAME	mg/Kg	mg/Kg	Recovery
PHENOL	N.D.	0.05	
BIS(2-CHLOROETHYL) ETHER	N.D.	0.05	91.5% 89.2%
2-CHLOROPHENOL	N.D.	0.05	
1,3-DIGHLOROBENZENE	N.D.	0.05	
1,4-DICHLOROBENZENE	N.D.	0.05	
BENZYL ALCOHOL	N.D.	0.10	
1,2-DICHLOROBENZENE	N.D.	0.05	
2-METHYLPHENOL	N.D.	0.05	
BIS(2-CHLOROISOPROPYL)ETHER	N.D.	0.05	
4-METHYLPHENOL	N.D.	0.05	
N-NITROSO-DI-N-PROPYLAMINE	N.D.	0.05	مع مد مد بگ
HEXACHLOROETHANE	N.D.	0.05	
NITROBENZENE	N.D.	0.05	
ISOPHORONE	N.D.	0.05	*
2-NITROPHENOL	N.D.	0.05	
2,4-DIMETHYLPHENOL	N.D.	0.05	
BENZOIC ACID	N.D.	0.25	
BIS(2-CHLOROETHOXY)METHANE	N.D.	0.05	91.5% 93.2%
2.4-DICHLOROPHENOL	N.D.	0.05	
1,2,4-TRICHLOROBENZENE	N.D.	0.05	
NAPHTHALENE	N.D.	0.05	
4-CHLOROANILINE	N.D.	0.10	
HEXACHLOROBUTADIENE	N.D.	0.05	
4-CHLORO-3-METHYLPHENOL	N.D.	0.10	
2-METHYLNAPHTHALENE	N.D.	0.05	
HEXACHLOROCYCLOPENTADIENE	N.D.	0.05	====+
2,4,6-TRICHLOROPHENOL	N.D.	0.05	
2,4,5-TRICHLOROPHENOL	N.D.	0.05	====
2-CHLORONAPHTHALENE	N.D.	0.05	
2-NITROANILINE	N.D.	0.25	
DIMETHYL PHTHALATE	N.D.	0.05	
ACENAPHTHYLENE	N.D.	0.05	
3-NITROANILINE	N.D.	0.25	<b>□ = = =</b> = =
ACENAPHTHENE	N.D.	0.05	87.5% 88.2%
2,4-DINITROPHENOL	N.D.	0.25	
4-NITROPHENOL	N.D.	0.25	
DIBENZOFURAN	N.D.	0.05	- P4
(continued on next page)			

5 DAYS TURNAROUND

Analytical Laboratory (E694)

Page 2

ChromaLab File # 1091017 G

Project Name: PACO PUMPS SOIL Sample I.D.: B7-0.5-10191 Method of Analysis: EPA 8270

Matrix: soil

	Sample	MDL	Spike
COMPOUND NAME	mg/Kg	mg/Kg	Recovery
2.4-DINITROTOLUENE	N.D.	0.05	
2,6-DINITROTOLUENE	N.O.	0.05	89.1% 87.1%
DIETHYL PHTHALATE	N.D.	0.05	شقة نصد جيبة بندر ووقد
4-CHLORO-PHENYL PHENYL ETHER	N.D.	0.05	And was task thin
FLUORENE	N.D.	0.05	
4-NITROANILINE	N.D.	0.25	
4,6-DINITRO-2-METHYL PHENOL	N.D.	0.25	
N-NITROSODIPHENYLAMINE	N.D.	0.05	
4-BROMOPHENYL PHENYL ETHER	N.D.	0.05	
HEXACHLOROBENZENE	N.D.	0.05	
PENTACHLOROPHENOL	N.D.	0.25	AND WAS DEED AND
PHENANTHRENE	N.D.	0.05	
ANTHRACENE	N.D.	0.05	
DI-N-BUTYL PHTHALATE	N.D.	0.05	with sout Add with with
FLUORANTHENE	N.D.	0.05	
PYRENE	N.D.	0.05	***
BUTYLBENZYLPHTHALATE	N.D.	0.05	
3,3'-DICHLOROBENZIDINE	N.D.	0.10	
BÉNZO(A)ANTHRÁCENE	N.D.	0.05	
BIS(2-ETHYLHEXYL)PHTHALATE	N.D.	0.05	
CHAYSENE	N.D.	0.05	86.1% 85.1%
DI-N-OCTYLPHTHALATE	N.D.	0.05	***
BENZO(B)FLUORANTHENE	N.D.	0.05	
BENZO(K)FLUORANTHENE	N.D.	0.05	
BENZO(A)PYRENE	N.D.	0.05	
INDENO(1,2,3 C,D)PYRENE	N.D.	0.05	هد هده المدير هجم نشتر
DIBENZO(A,H)ANTHRACENE	N.D.	0.05	
BENZO(G,H,I)PERYLENE	N.D.	0.05	

Chromatab, Inc.

Pavid Buong Senior Chemist Eric Tam Lab Director

**5 DAYS TURNAROUND** Analytical Laboratory (E694)

October 9, 1991

ChromaLab File No.: 1091017 F

JONAS & ASSOCIATES

Attn: R. Jonas

RE: 8080 ANALYSIS

Project Name: PACO PUMPS SOIL

Date Sampled: Oct. 1, 1991 Date Analyzed: October 4-8, 1991 Date Submitted: Oct. 2, 1991

RESULTS:

Sample I.D.: B6-0.5-10191

### CHLORINATED PESTICIDE ANALYSIS

	Concentration	Detection	Spike
<u>Compounds</u>	(µq/kq)	Limit (ua/ka)	Recovery
ALDRIN	N.D.	10	
DIELDRIN	N.D.	10	87.4%
ENDRIN ALDEHYDE	N.D.	50	
ENDRIN	N.D.	10	90.2%
HEPTACHLOR	N.D.	10	## <del>***</del> *** ***
HEPTACHLOR EPOXIDE	N.D.	10	
p,p' - DDT	N.D.	50	84.6%
p,p' - DDE	N.D.	10	93.1%
p,p' - DDD	N.D.	50	***
ENDOSULFAN I	N.D.	50	100.8%
ENDOSULFAN II	N.D.	50	
α - BHC	N.D.	10	
β - ВНС	N.D.	10	
$\gamma$ - BHC (LINDANE)	N.D.	10	86.9%
δ - BHC	N.D.	10	
ENDOSULFAN SULFATE	N.D.	100	
p,p' - METHOXYCHLOR	N.D.	100	
TOXAPHENE	N.D.	100	102.2%
PCB'S	400*	100	
CHLORDANE	N.D.	100	

\*PCB 1260

ChromaLab, Inc.

David Duong Chief Chemist

Eristam (by 00)

Eric Tam

**5 DAYS TURNAROUND** 

Analytical Laboratory (E694)

October 9, 1991

ChromaLab File No.: 1091017 G

JONAS & ASSOCIATES

Attn: R. Jonas

RE: 8080 ANALYSIS

Project Name: PACO PUMPS SOIL

Date Sampled: Oct. 1, 1991

Date Submitted: Oct. 2, 1991

Date Analyzed: October 4-8, 1991

RESULTS:

Sample I.D.: B7-0.5-10191

### CHLORINATED PESTICIDE ANALYSIS

Compounds	Concentration (ug/kg)	Detection Limit (49/kg)	Spike <u>Recovery</u>
ALDRIN	N.D.	10	
DIELDRIN	N.D.	10	87.4%
ENDRIN ALDEHYDE	N.D.	50	
ENDRIN	N.D.	10	90.2%
HEPTACHLOR	N.D.	10	
HEPTACHLOR EPOXIDE	N.D.	10	
p,p' - DDT	N.D.	50	84.6%
p,p' - DDE	N.D.	10	93.1%
p,p' - DDD	N.D.	50	
endosulfan i	N.D.	50	100.8%
endosulpan II	N.D.	50	-
α - BHC	N.D.	10	
$\beta$ - BHC	N.D.	10	
$\gamma$ - BHC (LINDANE)	N.D.	10	86.9%
δ - BHC	N.D.	10	
ENDOSULFAN SULFATE	N.D.	100	
p,p' - METHOXYCHLOR	N.D.	100	
TOXAPHENE	N.D.	100	102.2%
PCB'S	670*	100	
CHLORDANE	N.D.	100	

\*PCB 1260

ChromaLab, Inc.

Chief Chemist

ErzTam (by Do) Eric Tam

Analytical Laboratory (E694)

ChromaLab File No.: 1091017 G

5 DAYS TURNAROUND

JONAS & ASSOCIATES, INC.

Attn: R. Jonas

October 9, 1991

RE: One soil sample for TTLC CAM 17 Metals analysis

Project Name: PACO PUMPS SOIL

Date Sampled: Oct. 1, 1991

Date Submitted: Oct. 2, 1991

Date Analyzed: October 8, 1991

RESULTS:

Sample I.D.: B7-0.5-10191

<u>Metals</u>	Concentration (mg/kg)	Detection Limit (Mg/kg)	<pre>% Spiked Recovery</pre>	Regulatory Levels (mg/kg)
*Ag	N.D.	0.004	07.68	
		0.004	97.6%	500
As	N.D.	0.088	90.3%	500
Ba	138	0.005	102.3%	10000
Be	N.D.	0.001	103.2%	75
Cd	1.99	0.012	72.0%	100
Co	0.991	0.020	97.0%	8000
Cr	57.8	0.006	97.3%	2500
Cu	N.D.	0.004	105.6%	2500
Нg	N.D.	0.200	91.4%	20
Мо	8.85	0.040	87.1%	3500
ni.	348	0.026	97.2%	2000
₽b	N.D.	0.044	93.6%	1000
*Sb	N.D.	0.040	101.5%	500
Se	N.D.	0.200	87.6%	100
Tl	N.D.	0.088	100.1%	700
V	146	0.004	104.4%	2400
Zn	N.D.	0.006	102.2%	5000

Method of Analysis: 3050/6010 \*Method of Analysis: 3005/6010

ChromaLab, Inc.

David Duong Chief Chemist

Erritam (by De) Eric Tam

Analytical Laboratory (E694)

ChromaLab File No.: 1091017 F

**5 DAYS TURNAROUND** 

JONAS & ASSOCIATES, INC.

Attn: R. Jonas

October 9, 1991

RE: One soil sample for TTLC CAM 17 Metals analysis

Project Name: PACO PUMPS SOIL

Date Sampled: Oct. 1, 1991 Date Submitted: Oct. 2, 1991

Date Analyzed: October 8, 1991

RESULTS: Sample I.D.: B6-0.5-10191

<u>Metals</u>	Concentration (mg/kg)	Detection Limit (mg/kg)	% Spiked Recovery	Regulatory Levels (mg/kg)
W	V 75			
*Ag	N.D.	0.004	97.6%	500
As	N.D.	0.088	. 90.3%	50 <b>0</b>
Вa	10.7	0.005	102.3%	10000
Be	N.D.	0.001	103.2%	75
Cđ	0.595	0.012	72.0%	100
Co	0.804	0.020	97.0%	8000
Cr	2.63	0.006	97.3%	2500
Cu	67.8	0.004	105.6%	2500
Hg	N.D.	0.200	91.4%	20
Mo	N.D.	0.040	87.1%	3500
Ni	1.28	0.026	97.2%	2000
Pb	85.4	0.044	93.6%	1000
*Sb	N.D.	0.040	101.5%	500
Se	N.D.	0.200	87.6%	100
T1	N.D.	0.088	100.1%	700
v	N.D.	0.004	104.4%	2400
Zn	22.5	0.006	102.2%	5000

Method of Analysis: 3050/6010 \*Method of Analysis: 3005/6010

ChromaLab, Inc.

David Duong

Chief Chemist

Eriztam (by \$3)

Eric Tam

JONAS & ASSOCIATES INC.

P. O. Box 27153

Concord, California 94527 Telephone: (415) 676-8554

FAX #: (415) 680-6511

### SAMPLE CONDITION REPORT FORM

Page of Of Date of Request 16/19/ Job No. 10-217-d-Soil Cilent 14 CO VIMPS / Trans J Association	ZNC ·
Source of Samples Soil Shipper Arrival Date 16/2/9	

Analysis Authorized by P. Company of Associate Inc.

Affiliation/Address Phone 5/6-571-8554

Report Results to LAA Phone 316-676-8554

Lab # (Lab to fill in)	Sample Type'	Identify Information (e.g. site taken, field LD. number, etc.)	Condition Received at Laboratory <sup>2</sup>	Date of Collection	Specific Analyses Requested	
	5011	B1-3-5-1619) B2- 11 B3 4 B4 4 B5 11 B6 4 B7 11	1. Custody Form? 2. Package O.K.? 3. In Ice? 4. Container Type Suitable? 5. Air bubbles present/absent in VOA? 6. Proper preservation? 7. Custody Seai?  Remarks:  Gaughts retained in good Condition		8240; 8270; 8554/86 8240; 8556/846; 6636/ 8065; 8086 9240; 8274; 3556/84/ 54446 40 BZ 54416 40 B] 8240; 8270; 3554/86 54464 40 B6	5; 53 80 80/5; 248;

Specify as water (W), soil (S), or other (describe on back if need to).

Questions are to be answered (Y) or no (N) if possible. Special concerns (re. condition of sample, detection limits desired, important information re, sample history, etc.) should be noted under remarks.

CHROMALAB FILE # 1091017

## CHROMALAB, INC.

2239 Omega Road, #\* 415/831-1788 \* ORDER # 3653

**Chain of Custody** 

ANALYSIS REPORT Metal NUMBER OF CONTAINERS ence + Assocrates Inc. BASE/NEUTRALS ACIOS (EPA 625/627, 6270) PRIORITY POLLUTANT METALS (13) METALS: Cd, Cr, Pb, CAM METALS (18) W/Cr VI 5/0-676-8554 SAMPLEES (SIGNATURE) TIME MATRIX LABID. SAMPLEIDE STATE RELINCUISHED BY RELINQUISHED BY RELINQUISHED BY PROJECT INFORMATION SAMPLE RECEIPT TOTAL NO. OF CONTAINERS (TIME SIGNATURE) (UME) CHAIN OF CUSTODY SEALS REC'D GOOD CONDITION/COLD PRINTED NAME (CATE) PRINTED NAME SHIPPING IO. NO., CONFORMS TO RECORD LAS NO. RECEIVED BY (LABORATORY) RECEIVED BY RECEIVED BY SPECIAL INSTRUCTIONS/COMMENTS: 11:30 Am ISIGNATURE (BIGNATURE) (PRINTED NAME) (PRINTED NAME) (COMPANY) (COMPANY)

# Indemnification Agreement

### INDEMNIFICATION AGREEMENT

Paco Pumps, Inc. a Delaware Corporation ("Indemnitor") enters into this indemnification Agreement in order to induce and encourage Heller First Capital Corp. ("Lender") and Bay Area Employment Development Company ("BAEDC") and the United States Small Business Administration ("SBA") to provide loans to GP Holdings, LLC a California limited liability company ("Borrower")

For valuable consideration and with the intention to be bound legally, the undersigned parties agree as follows:

### Definitions.

For the purposes of this Agreement, the following definitions apply:

- A. The term "environment" incorporates by reference the definition of the term at 42 U.S.C. S 9601.
- B. The term "release shall means any spilling, leaking, pumping, pouring, emitting, emptying, discharging, injecting, escaping, leaching, dumping, or disposing into the environment.
- C. The term "contaminant" means any substance that is hazardous or detrimental to the environment and which is regulated by law, including, without limitation, any hazardous substance (as that term is defined at 42 U.S.C. S 9601), and petroleum (as that term is defined at 42 U.S.C. SS 6991).
- D. The term "Agency" shall means any federal, state or local governmental agency with statutory or regulatory authority to enforce the laws regarding the release of any contaminant into the environment.
  - E. The term "Property" means 9201 San Leandro St, Oakland, CA now owned by

Indemnitor, and which is described in greater detail in the mortgages that will be given by Botrower to Lender, Bay Area Employment Development Company, and SBA in the event that the loans are made to finance botrower's purchase of the Property from Indemnitor.

F. The term "ENSR Report" means a Report showing the results of a Environmental Assessment regarding-the Property dated May 15, 1995 that has been prepared by ENSR Consulting & Engineering prior to this Agreement, a copy of which is attached hereto as Exhibit A.

### II. Recitals

- A. Borrower desires to purchase the Property, and has submitted an application to Lender. Bay Area Employment Development Company, and SBA for loans which are to be secured by the Property.
- B. The <u>ENSR Report</u> indicates that there has been a release of a contaminant or contaminants into or on the Property and that remediation of such contaminants is or may be necessary under law.
- C. Lender, Bay Area Employment Development Company and SBA will not make a loan to Borrower unless they are provided with indemnification from indemnitor against losses they may sustain as the result of the release of any contaminant into or on the Property prior to the transfer of such Property to Borrower.
- D. Indemnitor desires to cooperate with and assist Borrower in its efforts to secure a loan from Lender, Bay Area Employment Development Company and SBA by agreeing to perform remediation of all contaminants released into or on the Property prior to the transfer of such Property to Borrower.

### III. Terms and Conditions

### A. Indemnity From Remediation.

- (1) Indemnitor agrees, at its sole cost and expense, to cause to be performed such investigation, monitoring and/or, remediation of all contaminants released into or on the Property as described in the <u>ENSR\_Report.</u>
- (2) In the event that any contaminant, not identified in the ENSR Report, is found to have been released into or on the Property prior to the transfer of the Property to Borrower, Indemnitor further agrees, at its sole cost and expense, to cause to be performed such investigation, monitoring and/or remediation of all such contaminants as may be requested or directed by any Agency or that an owner or operator of the Property may be required to perform under law, now or at any time in the future.
- (3) Indemnitor's obligations under paragraphs III.A.(1) and III.A.(2) shall continue until such time as any Agency exercising authority over the investigation, remediation and/or monitoring at the Property approves such actions and provides written notice that no further action will be required.
- B. Indemnity from Claims. Indemnitor further agrees to indemnify, defend and hold harmless Borrower, Lender, Bay Area Employment Development Company and SBA, and any of their assigns and successors in interest, which may take title to the Property, from and against any liabilities, damages, fees, penalties or losses arising out of any demand, claim or suit by any Agency or any other party relating to any contaminants found to have been released into or on the Property prior to the transfer of the Property to Borrower.
- C. Manner of Performance. Indemnitor shall perform its investigation, monitoring and/or remediation in a prompt manner and in compliance with all applicable laws and in a manner and at times which will not unreasonably interfere with Borrower's use of the Property.

D. Cooperation To Indemnitor. Borrower, Lender, Bay Area Employment

Development Company, SBA, and any of their assigns and successors in interest who may
take title to the Property shall provide reasonable cooperation to Indemnitor (1) to allow

Indemnitor to cause to be performed any investigation, monitoring and/or remediation pursuant
to paragraphs III.A.(1) and III.A.(2) to allow Indemnitor to conduct its defense of any claim
or suit referenced in paragraph III.B.(1).

### ., E. Reporting Requirements

- (1) Indemnitor shall promptly provide Lender, BAEDC. SBA and Borrower with copies of any written communications between Indemnitor and any Agency regarding the investigation, monitoring and/or remediation described in paragraphs III.A.(1) and III.A.(2), unless, at its sole option, Lender, BAEDC, SBA and/or Borrower agree to waive their right to obtain some or all of such communications.
- (2) Indemnitor shall promptly provide Borrower, Lender, BAEDC and SBA with a copy of (a) any written communication received from any Agency regarding any contaminants released into or on the Property for which Indemnitor may be responsible under paragraphs III.A. (1) or III.A. (2), and (b) any notice of any demand, claim or suit referenced in paragraph B(1) for which Indemnitor may be responsible.
- (3) Borrower, Lender and SBA, and any of their assigns and successors in interest who may take title to the Property, shall promptly provide indemnitor with a copy of (a) any written communication received from any Agency regarding any contaminants released into or on the Property for which Indemnitor may be responsible under paragraphs III.A. (1) or III.A. (2), and (b) any notice of any demand, claim or suit referenced in paragraph III.B.(1) for which Indemnitor may be responsible.
  - (4) In the event that the Property has been transferred or assigned to a party other-

than Borrower, Lender, BAEDC or SBA and that indemnitor has been notified in writing of this transfer or assignment, Indemnitor's obligations under paragraphs III.E.(1) and III.E.(2) also extend to such assignee or transferee.

- P. Attorney Fees and Costs of Litigation. Indemnitor agrees to pay all attorney fees and costs of litigation incurred by Borrower, Lender, BABDC and/or SBA, and any of their assigns and successors in interest who may take title to the Property, in any case brought to enforce the terms of this Agreement in which judgment is entered against Indemnitor.
  - G. Notices

All notices required under this agreement shall be made to the following:

(1) All notices to SBA shall be made to:

District Counsel 211 Main St, 4th floor San Francisco, CA 94105

- (2) All notices to Lender shall be made to:
  Kathryn J. Sennott, Senior Loan Officer
  Heller First Capital Corp.
  650 California St., 23rd Fl., San Francisco, CA 94108
- (3) All notices to Borrower shall be made to: Leonard M. Silvani, GP Holdings, LLC 9201 San Leandro St, Oakland, CA 94603
- (4) All notices to BAHDC shall be made to: James Baird Bay Area Employment Development Company 1801 Oakland Blvd #300 Walnut Creek, CA 94596
- (5) All notices to indemnitor shall be made to:

  Newflo Corporation (for Paco Pumps, Inc.)
  301 Camp Craft Road #100

  Westlake Hills, Austin, TX 78746

  Attn: John Lilla
  Phone 512-314-8500 FAX 512-314-8511

## H. Authority To Sign.

By signing below, each individual represents and warrants that he or she has proper authority to execute this Agreement.

Indemnitor has attached as Exhibit "B" to this Agreement a valid, certified resolution confirming such authority.

	s, die berneg im a estaten der Whitesidelie in die everatied St
of the latest date set forth below.	On ON Behalf of and
Dated: 9/28/95	Antilla for PALO PUMPS, INC
	Paco Pumps, Inc. a Delaware corporation
Dated:	
	Kathryn J. Sennott Heller First Capital Corp.
Dated:	
	Richard Jones U.S. SMALL BUSINESS ADMINISTRATION
Dated: 9-25-95	Leonard M. Silvani
	GP Holdings, LLC
Dated: 9/25/95	James Baird
•	Bay Area Employment Development Company

Report Site Contamination Study PACO Pump Facility Oakland, California For Amsted Industries, Inc.

August 12, 1987 Job No. 15215-008-43

# Dames & Moore



Page 2 of 47 Schedule 4.16

#### PRIVILEGED AND CONFIDENTIAL

August 12, 1987 Job No. 15215-008-43

Mr. Edward J. Brosius Senior Attorney Amsted Industries, Inc. 44th Floor Boulevard Towers South 205 North Michigan Avenue Chicago, Illinois 50501

Dear Mr. Brosius:

Report
Site Contamination Study
PACO Pump Facility
Oakland, California
For Amsted Industries, Inc.

This letter report presents the results of a site investigation performed at the PACO Pump facility located at 9201 San Leandro Street, Cakland, California (Figure 1). The purpose of the study was to explore an area on the west side of the PACO Pump machinery shop for potential contamination. The study area is bound by a cyclone fence on the east and the machine shop foundation on the west (Figure 2). The scope of our investigation was limited to excavating four shallow exploratory pits across the area, examining the sidewalls of each pit for evidence of discolored or oily soil, and analyzing two soil samples collected from each pit for concentrations of volatile and semi-volatile organic compounds as well as volatile and extractable petroleum hydrocarbons. All work was conducted as proposed. The details of our investigation, as well as our findings and recommendations, are presented below.

### FIELD INVESTIGATION

Four exploratory pits were excavated in the study area on Monday, July 27, 1987. Each pit, measuring approximately 3 feet by 3 feet in plan view, was excavated with a backhoe to a depth of about 3 feet (Figure 2). The work was performed by a licensed backhoe operator and monitored by a Dames & Moore geologist. Prior to beginning each excavation, the backhoe bucket was thoroughly steam cleaned to minimize the potential for cross contamination between pits.

Page 3 of 47 Schedule 4.16 Amsted Industries, Inc. August 12, 1987 Page two

Soil samples were collected from the sidewall of each pit at depths of 1.5 and 3.0 feet. Samples were collected by scooping soil into 3-inch stainless steel tubes with a trowel. The rings were covered at each end with 2 mil Teflon sheets, capped, taped, labeled and placed on ice. Samples were transported under chain-of-custody to Anatec Laboratories in Santa Rosa, California for analysis. Once samples were collected and sidewall observations were completed, exploratory pits were backfilled.

A total of eight soil samples were delivered to Anatec Laboratories for analysis. Each sample was analyzed for volatile and semi-volatile organic compounds (E.F.A. Methods 8240 and 8270, respectively) and extractable petroleum hydrocarbons (EPA Method 3550/8015). In addition, two samples were analyzed for volatile petroleum hydrocarbons (EPA Method 5020/8015) as well as polychlorinated biphenyls (EPA Method 8080). One sample was also analyzed for heavy metals (EPA Method 6010). For quality control purposes, analytical results were duplicated for one soil sample. A summary of analytical results is provided in Tables 1.0 and 2.0 and the actual laboratory results are attached to this report.

## CONCLUSIONS AND RECOMMENDATIONS

pata from the exploratory pits indicate that shallow soils in the study area consist of a dark brown to black gravelly fill material containing glass, bottles, bolts and garbage. At a depth of roughly 1.5 feet, the soil begins to increase in moisture, apparently the result of saturation with an oily substance. A visible oily sheen on the soil as well as a strong hydrocarbon odor was noted in each pit (Figure 3). At a depth of about three feet, the soil grades into a dark brown, stiff, silty clay. A hydrocarbon odor was still present. A free floating black oily looking substance was observed in pit number three at a depth of two feet.

Chemical analyses of the pit samples indicate that subsurface soils are contaminated with motor oil, creosote, and toluene. A variety of indentifiable, non-target organic compounds were also reported by the lab. Further analytical research of these compounds suggest they are various hydrocarbon residues typical of petroleum products.

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# DAMES & MOORE A PROFESSIONAL LIMITED PARTNERSHIP

Amsted Industries, Inc. August 12, 1987 Page three

As shown on Table 1.0, creosote, a wood preservative, was detected at a concentration of 790 mg/kg (ppm) in pit "3" at a depth of 1.5 feet. Motor oil was found in exploratory pits "1", "3" and "4" at concentrations ranging from 130 mg/kg (pit "1" at 3 feet) to 1100 mg/kg (pit "4" at 3 feet). Toluene, a solvent, was found at concentrations ranging from 110 ug/kg (ppb) (pit "4" at 1.5 feet) to a maximum concentration of 600 ug/kg (pits "1" and "2" at depths of 1.5 and 3.0 feet, respectively) (Table 1.0). No concentrations of PCB's or other organic compounds were detected. Metal concentrations were below California Total Threshold Limit Concentration (TTLC) and Soluble Threshold Limit Concentration (STLC), used to classify hazardous wastes (Table 2.0).

In summary, our investigation indicates that the nature of subsurface contamination at the site is motor oil, petroleum hydrocarbons and associated solvents used as additives to fuel products or solvents mixed with waste oils. Although creosote was identified in one sample, its presence may be isolated contamination linked to the adjacent railroad ties. The source and extent of hydrocarbon and solvent contamination is not known. However, the source may be attributed to spillage of waste oils. Further, the analytical results indicate that the northern portion of the study area (pits "3" and "4") have higher concentrations of extractable petroleum hydrocarbons than the southern area (pits "1" and "2").

To assess the extent of site soil contamination and the impact on groundwater (anticipated to be shallow), we recommend that additional site exploration be performed. Future activities should include the following:

- o Drill five to seven exploratory borings to a depth of approximately 15 feet. Samples should be collected at approximately 4, 8, 12, and 15 feet.
- o Complete one to two borings as groundwater monitoring wells. The wells should be screened at an appropriate depth to penetrate approximately the upper 10 feet of groundwater.
- o Inspect soil samples and groundwater wells for evidence of petroleum products.
- o Analyze selected soil and groundwater samples for total petroleum hydrocarbons, benzene, toluene and xylene.

Page 5 of 47 Schedule 4.16

# DAMES & MOORE A PROFESSIONAL LIMITED PARTNERSHIP

Amsted Industries, Inc. August 12, 1987 Page four

If evidence of deeper soil or groundwater contamination is discovered, additional borings may be necessary. Once the horizontal and vertical extent of contaminantion are better understood, appropriate clean up measures can be recommended. In applying guidelines set forth by the California Regional Water Quality Control Board, we anticipate that soils contaminated with petroleum hydrocarbons in excess of 1000 mg/kg may have to be excavated and disposed of at a Class I disposal facility. On-site treatment and disposal at Class II facilities can be considered for those materials contaminated between 100 mg/kg and 1000 mg/kg. Remediation of groundwater contamination will depend on the beneficial uses of the ground water, the feasibility of remedial technologies, the potential environmental impact and agency approvals.

We have enjoyed the opportunity to assist you on this investigation. If you have any questions concerning our findings or desire that we further investigate your facility, please do not hesitate to contact me at 415/896-5858.

very truly yours.

DAMES & MOORE

Dan M. Klinley

David M. Klimberg Associate

DMK:fs

Attachments: Table 1.0-Summary of Analytical Results.

Table 2.0-Summary of Metal Results

Plate 1-Site Location Plate 2-Site Plan

Plate 3-Logs of Exploratory Pits

Anatec Laboratories Report of Analytical

Results, August 5, 1987.

cc: Mike Ander - Dames & Moore, Chicago

Page 6 of 47 Schedule 4.16

TABLE 1.0 Summary of Analytical Results
PACO Pump Facility
Oakland, California

			<u>Oaklan</u>	d' Curroun	<u>a</u>		424 E1	nit 4/3.01
Analyte		pit 1/3.0°	Pit 2/1.5'	<u>Pit 2/3.0'</u>	Pit 3/1.5' 780 <sup>d</sup> (800 <sup>d</sup> )	<u>Pit 3/3.0'</u>	780	1,100
Extractable Petroleum b Hydrocarbons (mg/Kg) EPA Method 3550/8015			•	NER.	NR.	<10	NIR	NR
Volatile Petroleum Hydrocarbons (mg/Kg) EPA Method 5020/8015	NR <sup>C</sup>	NER.	<10		230	380	110	<b>45</b>
Tolume (ug/Kg)	600	470	420	600	2.40			ND
Pyrene (ug/Kg) EPA Method 8270	59	ND <sup>e</sup>	ND.	ND	Œ	<b>59</b>	ND .	. <b>6</b> 2
•								

as received basis

b - Data are quantitated as motor oil, unless otherwise noted

c - NR - Analysis not requested.

d - Quantitated as craosote

e - ND - not detected

<sup>( ) -</sup> Duplicate analysis

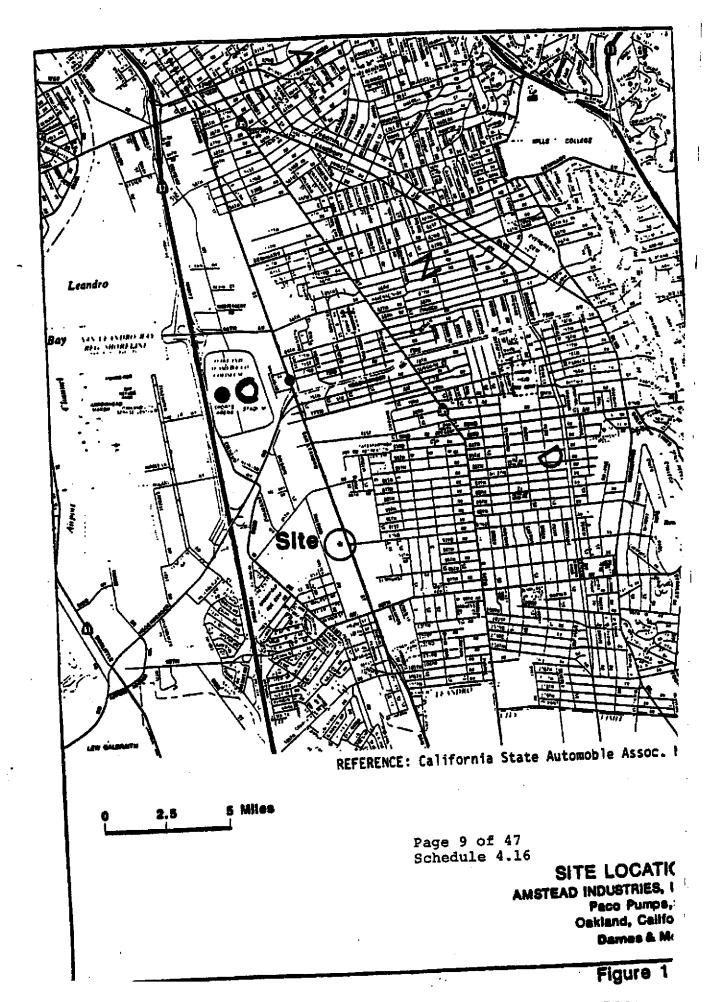
### TABLE 2.0 SUMMARY OF METAL RESULTS PIT #3 AT 3.0 FEET

Parameter	Results (mg/Kg
	₹50
Antimony	14
Arsenic	190
Barium	<b>&lt;2</b>
Beryllium	
-	∢3.
Cadmium	NAB
Chromium (VI)	41
Chromium (total)	-6
Cobalt	J
	22
Copper	
Lead	<20
Helcala	(0.05
WolApqeurw	<20
MOLADDAM	
	41_
Nickel	<0.5
Selenium	<1
silver	<30
Thallium	
	36
Vanadium	42
Zinc	

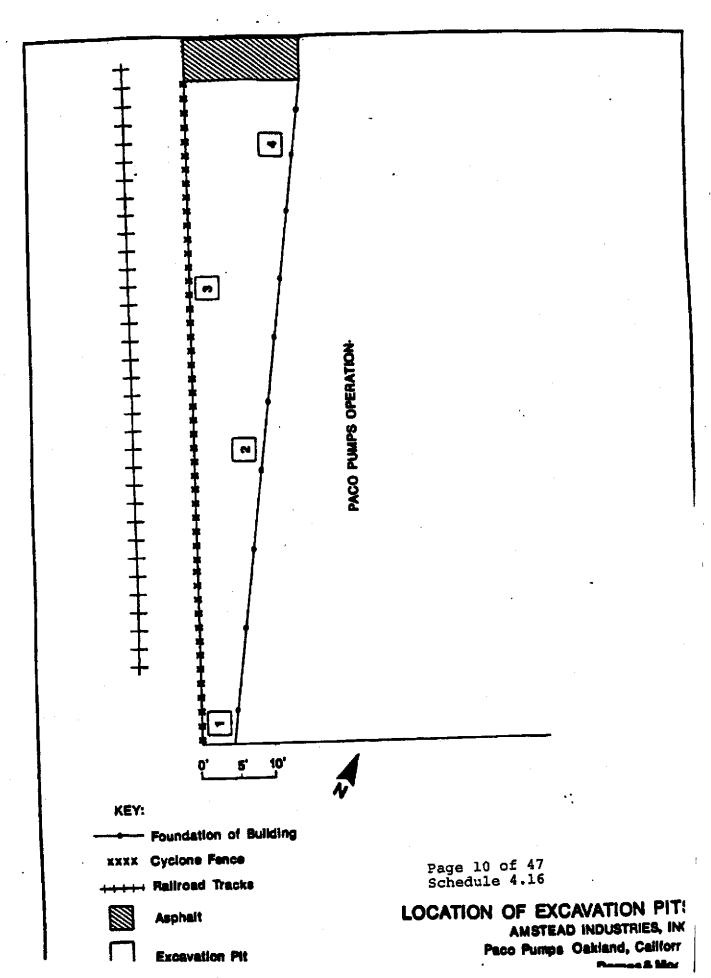
amg/Kg--Data are expressed as milligrams analyte per kilogram sample, as-received

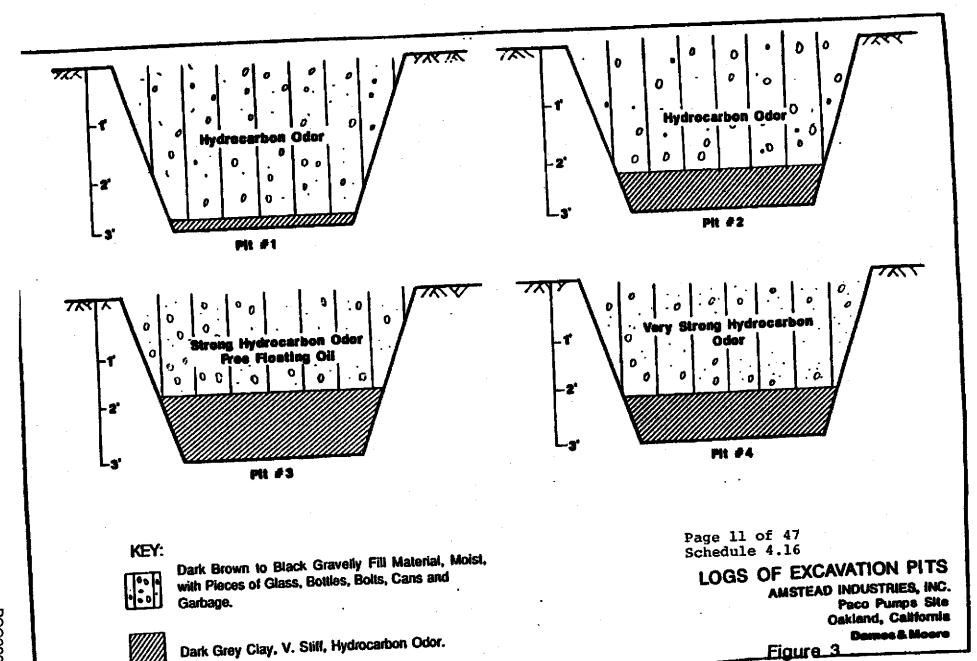
bNA--Not applicable; total chromium level is below regulatory limit (Section 66699,

Article II, California Administrative Code)
C-Arsenic - EPA Method 7060 Mercury - EPA Method 7471 Selenium - EPA Method 7740 Metals - all others - EPA Method 6010



PCC000931





Figure\_3



435 Tesconi Circle

Santa Rosa, California 95401

707-526-7200

Kris Franklin Dames & Moore 221 Main Street, Ste 600 San Francisco, CA 94105 August 5, 1987 ANATEC Log No. 9853 (1-8) Series No: 338/053 Client Proj No: 15215-008-043

Subject: URGENT Priority Analysis of Eight Soil Samples Identified as "PACO PUMPS, Project No. 15215-008-043" Received July 28, 1987.

Dear Ms. Franklin:

Analysis of the above referenced samples has been completed. This report provides details of analytical methodologies used to produce the results transmitted to you on August 4, 1987.

## Sample Receipt and Log-in

Eight soil samples were received at the laboratory under documented chain-of-custody on July 28, 1987. Sample custody was transferred to ANATEC Sample Control staff who initiated the laboratory log-in process. The log-in process consisted of the following activities:

- (1) Inspection and notation of the condition of all samples received;
- (2) Inspection of sample label information and reconciliation with information submitted on sample custody documents;
- (3) Assignation of ANATEC laboratory log and sample numbers;
- (4) Matching of analyses to be performed with sample containers prepared by methods compatible with analyses;
- (5) Documentation of processes listed above and any irregularities on the laboratory log sheet; and
- (6) Inspection and approval of the documents package and testing protocols by the Project Manager.

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Biological Studies • Laboratory Analysis • Research



Subsequent to completion of log-in procedures, all samples were placed in secure storage where they were maintained at 4 °C until analysis commenced.

It was noted that all samples were in good condition on arrival (cold and labeled legibly and completely).

One sample "#6, 3/3' 7/27 1600" was analyzed to measure various metals. All samples were analyzed to measure volatile and semivolatile organic compounds, volatile and extractable petroleum hydrocarbons, and polychlorinated biphenyls (PCBs). The methods used for these determinations are listed in Table 1 and are described in the following sections of this report.

## Metals Measurements

Concentrations of seventeen metals ("project metals") were measured using several atomic spectroscopic techniques. Project metals other than arsenic, selenium and mercury were measured using inductively coupled argon plasma atomic emission spectroscopy (ICP). Samples were prepared for ICP analysis by acid dissolution in accord with requirements of Method 6010 in "Test Methods for Evaluating Solid Waste, Physical/Chemical Methods," U.S. EPA, SW-846, 2nd edition, revised 1984. Briefly, 1-gram aliquots of samples were heated with nitric acid (without boiling) until the appearance of the digestate was constant. The solution was cooled, dilute hydrochloric acid added, and then warmed again. Reagent water was addded to a final volume of 50 mL, and the solution filtered.

The filtered digestates were analyzed on a Perkin-Elmer Model 6500 Plasma Emission Spectrometer. The spectrometer operates by nebulizing a solution which is then transported to an argon-plasma torch where characteristic atomic-line emission spectra are generated and intensities of characteristic wavelengths measured by the optical system. The following wavelengths were monitored for each element:

Element	Wavelength	Element	Wavelength (nm)
Antimony Barium Beryllium Cadmium Chromium Cobalt Copper	206.833	Lead	220.353
	233.527	Molybdenum	202.030
	313.042	Nickel	231.604
	226.502	Silver	328.068
	267.716	Thallium	190.864
	228.616	Vanadium	292.402
	324.754	Zinc	213.856

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338/053 LOG 9853

August 5, 1987

The spectra were compared by computer to spectra obtained from calibration solutions containing the elements of interest.

Preparation of the sample for arsenic and selenium measurement was similar to that for other metals except that hydrochloric acid was omitted from the procedure; prior to analysis nickel nitrate was added to digestates to reduce analyte volatility. Arsenic and selenium were quantitated using the following atomization program:

Drying Time - Temperature - 30 sec at 125 °C
Ashing Time - Temperature - 30 sec at 1,200 °C
Atomizing Time - Temperature - 6 sec at 2,700 °C
Wavelength 193.7nm (arsenic), 196.0 nm (selenium)
Background correction: Deuterium arc

Mercury content of samples was measured using the "Manual Cold" Vapor" atomic absorption spectroscopic method. Procedural details are available as Method 7471 in "Test Methods for Evaluating Solid Waste, Physical/Chemical Methods," U.S. EPA, SW-846, 2nd edition, revised 1984. Samples were prepared for mercury analysis by gentle heating with nitric and sulfuric acids and potassium permanganate and potassium persulfate. Analysis of digests involved reduction of mercuric ions generated during preparation to elemental mercury. Elemental mercury was swept from the reduction flask into an absorption cell on the optical bench of an atomic absorption spectrometer. Absorbance was monitored at 253.7nm with deuterium arc background correction.

# Petroleum Hydrocarbons Testing - Volatile and Extractable

Volatile Petroleum Hydrocarbons

Samples were analyzed to measure volatile petroleum hydrocarbon (e.g. gasoline, jet fuel) residues by headspace sampling/flame ionization-gas chromatography. In performing the analysis a portion of sample was measured into and sealed in a septum-top glass vial. The vial was heated at 90 °C during which time volatile hdyrocarbons equilibrated with headspace gases. A two-milliliter portion of headspace gas was removed through the septum with a gas-tight syringe and injected onto the analytical column of a gas chromatograph equipped with a flame-ionization detector. The analytical system was calibrated by analysis of standards treated as described for samples. The standards were prepared with commercial (regular) gasolines.

Page 14 of 47 Schedule 4.16



#### 338<u>/053 LOG **9853**</u>

Interpretation of sample chromatograms consisted of first inspecting the chromatographic pattern or "fingerprint" and comparing it with those generated by calibration standards. Hydrocarbon mixtures if detected in samples were then quantitated using response factors generated by standards which match or approximate sample chromatograms.

Chromatographic operating conditions for measurement of volatile petroleum hydrocarbons were as follows:

Column: Carrier: 6' x 1/8" SP1500 Nitrogen at 20 mL/minute

Oven Program:

Initial Temperature-Time 150 °C - 2 min Program Rate - 10 °C/min

Final Temperature - 230 °C

Injector Temperature - 280 OC FID Temperature - 300 °C

Details of the procedure are consistent with "Method I. Total Fuel Hydrocarbons Analysis (Low to Medium Boiling Point Hydrocarbons)" in "Guidelines for Addressing Fuel Leaks," Regional Water Quality Control Board, San Francisco Bay Region, revised 1986. Further information regarding chromatographic interpretation may be found in "Method D3328-78," in "Comparison of Waterborne Petroleum Oils by Gas Chromatography." ASTM Standards on Chromatography, 1st edition, 1981. Additional information pertaining to headspace techniques may be found as Method 5020 in "Test Methods for Evaluating Solid Waste, Physical/Chemical Methods," U.S. EPA, SW-846, 2nd edition, revised 1984.

## Extractable Petroleum Hydrocarbons

Measurement of extractable petroleum hydrocarbon (e.g., diesel, creosote, motor oil) residues is performed by analysis of methylene chloride sample extracts by gas chromatography with flame ionization detection. Fifteen gram portions of sample are extracted by sonication with 30 milliliters of methylene chloride using ultrasonic agitation. The organic layer is drawn off and the sample extracted two additional times. Extracts are combined, dried over sodium sulfate and concentrated in Kuderna-Danish apparatus by evaporation of solvent at 70 °C. Concentrated extracts are injected into a capillary-column gas chromatograph equipped with a flame ionization detector. Interpretation of

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sample chromatograms is as described above for volatile petroleum hydrocarbons except that commercial diesel fuel, creosote and 10-40W motor oil are used as calibration standards. Chromatographic operating conditions are as follows.

- 5 -

Column dimensions:

Coating Head pressure

Temperature program

30m x 0.25 mm fused silica capillary

SPB-1

12 psi helium

40°C for 6 min, 10°C/min to 270°C,

hold for 11 min

Injection technique Temperature

Splitless 260°C

Detector

FID at 250°C

# Volatile Organic Compounds

Samples were tested in accord with U.S. EPA Method 8240 in "Test Methods for Evaluating Solid Waste, Physical/Chemical Methods," U.S. EPA, SW-846, 2nd edition, revised 1984 to measure contents of 31 volatile organic priority pollutants and other purgeable compounds. Briefly, the method involves making a slurry of 1-gram of sample and deionized water, dosing the slurry with internal and surrogate standards, and bubbling inert gas through the mixture at ambient temperature. The volatile compounds are transferred to the vapor phase which is swept through a sorbent trap. After purging is complete, the trap is heated and backflushed with inert gas to desorb the compounds onto a gas chromatographic column. The column is then temperature programmed, and the compounds detected with a mass spectrometer. The following instrument parameters were used:

Purge and Trap Device: Trap packing

Tekmar Model LSC-2 1 cm methyl silicone 15 cm 2,6-diphenylene oxide polymer (TENAX)

8 cm silica gel

Purge gas Purge time Desorb temp Desorb time (cont.)

Helium at 30 mL/min 11 min 180°C 4 min

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GC/MS Unit:
Column dimensions
Coating
Head pressure
Temperature program

HP 5970 MSD
6° x 0.1° stainless steel
1% SP-1000 on 60/80 mesh Carbopak B
50 psi helium
40°C for 4 min, 6°C/min to 220°C,
hold 15 min

Mass spectrometer mode Electron energy Mass range Scan time Calibration gas Data system Electron Impact
70 eV
35 - 260 amu
2.5 seconds
Perfluorotributylamine (PFTBA)
HP-1000

# Semi-volatile (Extractable) Organic Compounds

Samples were tested in accord with U.S. EPA Method 8270 in "Test Methods for Evaluating Solid Waste, Physical/Chemical Methods," U.S. EPA, SW-846, 2nd edition, revised 1984 to measure contents of 66 semi-volatile (extractable) priority pollutants and other extractable compounds. Briefly, a fifteen gram aliquot of sample which has been dosed with internal and surrogate standards is extracted with methylene chloride using ultrasonic agitation. The organic layer is drawn off and the sample extracted two additional times. The extracts are combined, dried over sodium sulfate and concentrated by gentle evaporation in Kuderna-Danish apparatus to a final volume of one milliliter. An aliquot of the resulting extract is analyzed by GC/MS under the following conditions:

Column dimensions
Coating
Head pressure
Temperature program

30m x 0.25 mm

DB-5 at 0.25 micron thickness

12 psi helium

40°C for 2.5 min, 8°C/min to

310°C, hold for 5 min

Injection technique Volume Temperature Splitless 2 microliters 275°C

Mass spectrometer mode Electron energy Scan range Scan time Calibration gas Electron Impact
70 eV
35 - 450 amu
0.5 sec
Perfluorotributylamine (PFTBA)

Data system

HP-1000

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It was noted that the total ion chromatograms for analyses of semi-volatile compounds by GC/MS contained peaks other than those specified by Method 8240 for most samples. A single sample was chosen (48, 4/3' 7/27 1600) and the ten most prominent non-target compound peaks in the chromatogram were compared with the computer library of mass spectra, which contains data compiled by the National Bureau of Standards, the Environmental Protection Agency and the National Institutes of Health. The five most likely identities, if five were available, of each unknown compound peak are listed in Table 7.

# Polychlorinated Biphenyls

Samples were tested in accord with U.S. EPA Method 8080 in "Test Methods for Evaluating Solid Waste, Physical/Chemical Methods," U.S. EPA, SW-846, 2nd edition, revised 1984 to measure contents of seven common (Aroclor) mixtures of polychlorinated byphenyl compounds (PCBs). Briefly, 15-grams of sample were extracted with methylene chloride using ultrasonic agitation. The organic layer was decanted and the extraction repeated twice more. Solvent extracts were combined and dried over sodium sulfate, then exchanged to hexane by repeated concentration in a Kuderna-Danish apparatus and dilution with hexane. Extracts were then passed through a column of partially deactivated Florisil and compounds of interest subsequently eluted using 40% ethyl ether in n-pentane. The isolated extract was concentrated by gentle evaporation in Kuderna-Danish apparatus. Reduced extracts were stored at -20°C until analyzed.

Analysis was performed by injection of an aliquot of the final extract onto the column of a gas chromatograph equipped with electron capture detectors. Any positive results were considered presumptive until confirmed by reanalysis on a second chromatographic column of quite different chromatographic quality. The following instrument parameters were used:

Carrier gas

Column 1 dimensions Coating Temperature program

(cont.)

5% methane in argon

30m x 0.75 mm fused silica capillary SPB-1 at 1 micron thickness 160°C for 2 min, 2°C/min to 220°C, hold for 17 min

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Column 2 dimensions

Coating

30m x 0.75 mm fused silica capillary SPB-5 at 21 micron thickness

same as for column 1

Temperature program same as for economic same as fo

Detector ECD at 250°C

Decec for

# Presentation of Results

Results of all sample analyses are contained in the following tables:

TABLE 2 - Metals Analyses for "Sample 6, 3/3' 7/27 1600"

TABLE 3 - Volatile Organics Analyses

TABLE 4 - PCBs

TABLE 5 - Extractable and Volatile Petroleum Hydrocarbons

TABLE 6 - Semi-volatile Organics Analyses

TABLE 7 - Possible Identities of Non-target Compound Peaks Obtained During Analyses of Organic Compounds by GC/MS

APPENDIX A - Total Ion Chromatograms of Sample Analyses for Semi-volatile Organic Compounds by GC/MS

APPENDIX B - Chromatograms of Sample Analyses for Extractable Petroleum Hydrocarbons by GC/FID

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## Quality Assurance

Analysis of samples was accompanied by various quality control procedures. These included preparation and analysis of method blanks and standards, and replicate and analyte-fortified ("spiked") sample portions. Results of quality control procedures are available on request but are not included in this report.

Please feel welcome to contact us should you have questions regarding procedures or results.

Submitted by:

David Hirano Project Chemist

Enc: Custody Documentation

Approved by:

Greg Adderson, Director Analytical Laboratories



SUMMARY OF ANALYTICAL METHODS AND METHOD REFERENCES TABLE 1. FOR "PROJECT NO. 15215-008-043, PACO PUMPS" SOIL SAMPLES RECEIVED JULY 28, 1987

Parameter	Method Description1	Method Number	Method Reference <sup>2</sup>
Arsenic Mercury Selenium Metals- all others	AAS-HGA AAS-CV AAS-HGA ICPAES	7060 7471 7740 6010	1 1 1
Organic compounds- Semi-volatile Volatile	GC/MS (purge & trap)	8270 8240	1
Petroleum hydrocarbons- Extractable Volatile	GC-FID GC-FID (headspace)	3550/8015 5020/8015	1
PCBs	GC-EC	8080	1

### labbreviations:

AAS-- Atomic absorption spectrophotometry

Heated graphite atomization

Cold-vapor generation

ICPAES -- Inductively coupled argon plasma atomic emission spectroscopy

GC-EC-- Gas chromatograph with electron capture detector

GC-FID--Gas chromatography with flame ionization detection.

GC/MS-- Gas chromatography/mass spectrometry

<sup>2</sup>References:

1--"Test Methods for Evaluating Solid Waste, Physical/Chemical Methods," U.S. EPA, SW-846, 2nd edition, revised 1984.



SUMMARIZED RESULTS OF METALS ANALYSES FOR "SAMPLE 6, 3/3' 7/27 1600" SOIL SAMPLE RECEIVED JULY 28, 1987 TABLE 2.

Parameter	Results (mg/Kg)a
Antimony	<50
	14
Arsenic	190
Barium	<2
Beryllium	1.2
m = .1	<3
Cadmium	41
Chromium (total)	6
Cobalt	•
_	22
Cobber	<20
Lead	₹0.05
Mercury	<20
Molybdenum	<20
	41
Nickel	<0.5
Selenium	<b>&lt;1</b>
Silver	<30
Thallium	(30
	36
Vanadium	42
zinc	

amg/Kg--Data are expressed as milligrams analyte per kilogram sample, as-received basis.



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<u> August 5. 1987</u>

TABLE 3. SUMMARIZED RESULTS OF VOLATILE ORGANICS ANALYSES FOR "PROJECT NO. 15215-008-043, PACO PUMPS" SOIL SAMPLES RECEIVED JULY 28, 1987

		Descriptor, Lab No. & Results (ug/kg)&						
	мскр	1/1.51	2/1.5' 7/27 1600	3/1.5' 7/27 1600 Sample #5	3/1.5	4/1.5' 7/27 1600 Sample 87	1/3' 7/27 1600 Sample #2	2/3° 7/27 1600 Sample #4
Analyte	( <u>uq/Kg</u> )	(9853-1)	<u>(9853-2)</u>	(9853-3)	(9653- <u>3</u> R)	(9853-4)	(9853-5)	<u>(9853-6)</u>
	25	KDG	ND	ND	ND	ИD	ND	ND .
Bensene Bronodichloromethane	10	ND	ND	ND	ND	ND.	ЖD	ND
	25	ND	ND	ND	ЖD	ND	ND	ИD
Brunoform	15	ND	ND	ND	ND	ND	30	ЖD
Bronzmethene Carbon tetrachloride	15	ND	ND	УD	ND	ЖD	ЖD	ND
CULTUM COCKACHIOCTION	_							
***	25	ND	, ND	ND	ND	ИD	ND	ND.
Chlorobanaere	15	ND	ND	ND	ND.	НD	ND	KD '
Chlorosthane	35	ND	ND	ND	ND	ND	ND	КD
2-Chlorosthylvinyl ether	10	ND	ND	KD ·	ND	ND	ND	<b>XD</b>
Chloroform	15	ND	ND	ЖD	ND	ND	ND	NO
Chloromethane								
	15	ND	ND	ND	ND	ND:	· ND	_ ND
Dibranchloromethane	25	ND	ND	ND	ND	ND	ИD	· ND
1,2-Dichlorobenzene	25 25	ND	ND	ND	ND	ND	Ж	ND
1,3-Dichlorobenzene	25	32D	NO	ND	ND	ND	ХD	ND
1,4-Dichlorobenzene	20	)ED	190	ND	ND	ND	ND	ХD
1,1-Dichlorosthane	20		•-					
	15	. ND	ND	ND	ND	ND	ND	ND .
1,2-Dichlorosthans	15	15D	ND	ND	ND.	ND:	ND	ND
1,1-Dichlorosthess	10	ИD	ND	ND	<b>SD</b>	ND	ND	ND
trans-1,2-Dichlorosthens	25	180	ND	ND	ND	NO	₽ <b>D</b>	ND
1,2-pichlo <del>ropropans</del>	20	ND	10	ND.	ND	ND	ND	ЖD
cis-1,3-bichloropropene	ىم							
	24	ND	ND	ND	ХD	ND	ND	ND ND
trans-1,3-Dichloropropens	25	ND ND	15D	190	ND	ND	ND	ИD
Ethyl benzane	30	ND ND	ND:	190	NO	ND	XD	ND
Mathylane chloride	15	ND ND	)ED	)	ND	ND	ND	ND
1,1,2,2-Tetrachlorosthene	30	ND ND	XD	ND	ND	ND	1 <del>2</del> D	KD
Tetrachlorouthens	20	K		,				
	45	600	420	230	300	110	470	600
Tolume	25		7D	ND	ND	ND	ND.	ĬΦ
1,1,1-Trichlorosthana	20	HD	ND	ND TO	ЖD	ND	ND	ND
1,1,2-Trichlorosthane	25	HD	ND ND	ND OK	ND	ND	ЖD	ND
Trichlososthens	10	ND	150	₩D	ND	ND	ИD	NO
Trichlorofluoremethene	15	ND		)D	).D	ND	ND	ИD
Vinyl chloride	15	ND	ЖD					•

<sup>&</sup>quot;mg/Kg-Data are expressed as milligrams analyte per kilogram sample, as-received basis.

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Des Hethod detection limit.

On that detected at the method detection limit.



338/053 LOG 9853 TABLE 3. (cont.)

### Descriptor, Lab No. & Results (ug/Kg)1

•		3/31	4/3*
		7/27 1600	7/27 1600
	MDL <sup>2</sup>	Sample #6	Sample 48
	(pg/ <u>kg</u> )	(9853-7)	(9853 <del>-8)</del>
Analyte	(Bay va)	7298	
	25	KD3	)AD
Benzene	10	ND	ND
Bromodichloromethane	25	ND	MD
Bromoform	15	MD	ND
Branamethane	15	ND	ND
Carbon tetrachloride	7-3		
	25	ЖĎ	ND
Chlorobensene	15	ND	KD
Chloroethane	35	ND	190
2-Chloroethylvinyl ether	10	KD	ND
Chloroform	15	ND	ND
Chloromethane			
	15	ND	ND
Dibromochloromethane	25	ND	ЖD
1,2-Dichlorobenzene	25	ND	ИD
1,3-Dichlorobenzene	25	ND	ИD
1,4-Dichlorobenzene	20	ND	ND
1,1-Dichloroethene		•	
_	15	ND	ND
1,2-Dichloroethane	15	ИD	ND
1,1-Dichloroethene	10	ND	ND
trans-1,2-Dichloroethene	25	ND	ND
1,2-Dichloropropens	20	ND	N <b>D</b>
cis-1,3-Dichloropropens	20		
	25	ND	ND
trans-1,3-Dichloropropene	30	<b>10</b>	ND
Ethyl benzene	15	NÓ	ND
Methylene chloride		ND	ИD
1,1,2,2-Tetrachloroethane	20	ND	ND
Tetrachloroethene			
	25	380	45
Toluene	20	ND	ИD
1,1,1-Trichloroethane	25	ND	₩D
1,1,2-Trichloroethane	10	ND	ЖD
Trichlorosthens	15	ND	140
Trichlorofluoromethane	15	KD	ND
Vinyl chloride	73		

lpata expressed in units of micrograms analyte per liter sample. 2MDI--Method detection limit.

<sup>3,00-</sup>Not detected at the method detection limit.



SUMMARIZED RESULTS FOR PCBS ANALYSIS FOR "PROJECT NO. 15215-008-043, PACO PUMPS" SOIL SAMPLES RECEIVED JULY 28, 1987 TABLE 4.

Descriptor, Lab No. & Results (mg/Kg)1

	_					
Parameter	MDL <sup>2</sup> (mg/Kg)	2/1.5' 7/27 1600 Sample #3 (9853-2)	3/3' 7/27 1600 Sample #6 (9853-7)			
PCB-1016 PCB-1221 PCB-1232 PCB-1242 PCB-1248 PCB-1254 PCB-1260	0.2 0.2 0.2 0.2 0.2 0.2	ND3 ND ND ND ND ND	ND ND ND ND ND			

lmg/Kg--Data are expressed as milligrams analyte
per kilogram sample, as-received basis. liter sample.

<sup>2</sup>MDL--Method detection limit.

<sup>3</sup>ND--Not detected at the listed method detection limit.



TABLE 5. SUMMARIZED RESULTS FOR SEMI-VOLATILE AND VOLATILE PETROLEUM HYDROCARBON ANALYSES FOR "PROJECT NO 15215-008-043, PACO PUMPS" SOIL SAMPLES RECEIVED JULY 28, 1987

						Results	mg/Kg)a
ANATEC		Sample	Descr	iptor		Semi-volatile Petroleum Hydrocarbonsb	Volatile Petroleum Hydrocarbons
<u>Lab No.</u> 9853-1 9853-2	1/1.5'	7/27 7/27	1600 1600	Sample Sample	#1 #3	250 <10	NRC <10
9853-3 9853-3R	3/1.5° 3/1.5°	7/27 DUPLI		Sample		780 <b>đ</b> 800 <b>đ</b> 780	NR NR NR
9853-4	4/1.51		1600 1600	Sample Sample		130	nr
9853-5 9853-6	1/3' 2/3'	7/27 7/27	1600	Sample	#4	<10	NR . <10
9853-7 9853-8	3/3' 4/3'	7/27 7/27	1600 1600	Sample Sample	#6 #8	1,100	NR

amg/Kg--Data are expressed as milligrams analyte per kilogram sample, as-received basis. bData are quantitated as motor oil, unless otherwise noted.

CNR--Analysis not requested.

dQuantitated as creosote.

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			Pes	eriotor, L	ab No. 6 Re	eults (ug	/Kq)1	
٠	MDI,2 (UG/Kg)	1/1,5' Sample 1 (9853-1)	2/1.5' Sample 3 (9853-2)	3/1,5° Sample 5 (9853-3)	WAST-R	4/1,5' Sample 7 (9853-4)	32701 2 (9853-5)	2/3' sample 4 (9853-6)
Aconspictment Aconspictment Aconspictment Aconspictment Benzicine							<del></del>	35555555555555555555555555555555555555

essed in units of micrograms analyte per kilogram sample. od detection limit. etected at the method detection limit.

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338/053 LOG 9853

# Table 6, (cont.)

	<u> </u>	Descri	ults (ug/Kg)1
Analyte	MDI.2 (ug/Kg)	3/3' (9853-7)	4/3' {9853-8}
a consultation	33		16666666666666666666666666666666666666
Anthracens Senzidine	33	<b>36</b>	100 100 100
Benzo a diluoranthene Benzo k lluoranthene Benzo k lluoranthene	33	200 200 200 200 200 200 200 200 200 200	10 10 10
Benzo(ghi)perylene Benzy(butyl phthalate	#	99 99 99 99	10 10 10
gamma-BBC Bis (2-chloroethyl)ether Bis (2-chloroethoxy)methane	33	20 20 20	755 750 750
Bis(2-chloroisoproy) halate Bis(2-ethylhexyl) chthalate 4-Bromophanyl phanyl ether	3,300	<b>356</b>	190 190 190
2-Chlorophenyl phenyl ether Chrysone		<b>3</b> 8868	100 200 200
1,1'-008 4'-007 Dibenzo(a,h)anthracens	1,630 1,630	6666	ND ND
Di-n-butyl phenalate 1,2-Dichlorobenzene 1,3-Dichlorobenzene	33	<b>3888</b>	366 366
3 3 - Dichlorobensidine pieldrin predata	33	25 25 25 25	20 20 20
Dimethyl phthalate 2,4-Dinitrotoluene 3,6-Dinitrotoluene	133	20 20 20 20 20 20 20 20 20 20 20 20 20 2	110 110 110
Di-n-octylphthalate Endosulfan sulfate Endrin aldehyde	33	<b>686866866</b> 66	8688 8688
Pluorentene Fluorene Heptachlor	33	888 888	ND ND ND
Hexachlorobenzene Hexachlorobutadiene Hexachlorocyclopentadiene	44 43	的形	29 29 20 20 20
Hazachloroethane Indeno(1,2,3-cd)pyrane Isophorone	33	9666 8666	
Naphthalene Nitrobenzene Nanitrosodi-n-propylamine	1,3	<b>66</b> 6	ND ND ND
Access here and a constraint a c	133 23	59 90 90 90 90 90 90 90 90 90 90 90 90 90	100 100 100
2-Chlorophenol	76. 1,65		100 100 100 100
Phenanthrene Pyrene 12.4-Trichlorobensene 12.4-Trichlorobensene 2.4-Chloro-3-methylphenol 2.4-Dichlorophenol 2.4-Dimethylphenol 2.4-Dimethylphenol 2.4-Dimethylphenol 2.4-Dimethylphenol 2.4-Nitrophenol Pentachlorophenol Phenol 2.4-6-Trichlorophenol	ol 7층		
4-Ni trophenol Pentachlorophenol Phenol	A STATE OF THE STA		MD MD MD
2,4,6-Trichlorophenol			

loats expressed in units of micrograms analyte per kilogram sample.

AMDI-Hethod detection limit.

AND-Hot detected at the method detection limit.

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TABLE 7. POSSIBLE IDENTITIES OF THE TEN MOST-PROMINENT NON-TARGET PEAKS OBTAINED DURING GC/MS ANALYSIS OF "SAMPLE #8, 4/3' 7/27 1600" RECEIVED JULY 28, 1987

•	Possible Identity of Compounds	Percent Probability <sup>2</sup>	Molecular Weight	Molecular Formula
Peak No.1	LOSSIDIA ITALIA	61	490	C35H70
_	17-pentatriacontene	60 61	278	C20H3B
1		<del>5</del> 5	350	C25H50
	NEOPHITADIEAL Heptadecane, 9-(2-cyclohexylethyl)-		424	C29H600
	Nonacosanol	52	324	C21H4002
	OLEIC ACID, PROPYL ESTER	46	447	-
-	Office states and the state of	60	424	C298600
_	Honacosanol	59 58	324	C21H4002
2	OLEIC ACID, PROPYL ESTER		336	C24H48
	CYCLODOCOSANE, ETHYL-	53	204	C15824
	.alphaEumulene	53	364	C26H52'
	1-Hexacosene	44	304	7
	T-ESTERANCE	69	490	C35H70
_	17-Pentatriacontene	67	336	C24H48
3	CYCLODOCOSANE, ETHYL-	53	718	C2021030
	1-Pentacontanol		124	C9H16
	3,4-Octadiene, 7-methyl-	49	424	C29H600
	Nonacosanol	49	76-	•-
	Minedanana	65	718	C50H1020
<u>;</u>	1-pentacontanol	•••	268	C18H360
4	Octadecanal	60	242	C15H3002
	Oxirane, [(dodecyloxy)methyl}-	59	490	C35H70
	17-Pentatriacontene	50	326	C12H24Br2
	Dodecane, 1,2-dibromo	47	704	<b>-</b>
	Dodgerne) -1- 4-1-1		718	C50H102O
	1-Pentacontanol	46	240	C17H36
. 5	Heptadecane	38	490	C35H70
	17-pentatriacontene	37	266	C19H38
	TRIDECANE, 6-CYCLOHEXYL-	36	232	C14H29C1
	Tetradecane, 1-chloro-	35	646	<b>V</b>

<sup>1</sup> Number of compound peak in decreasing order of prominence.
2 Probability of correct identity of unknown compound, expressed as percentage.

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Table 7, (con.t)

- <u>19 -</u>

Peak No.1	Possible Identity of Compounds	Percent Probability <sup>2</sup>	Molecular Weight	Molecular Formula
		30	490	C35H70
6	17-Pentatriacontene Cyclopentane, 1,1'-[3-(2-cyclopentylethyl	· ·		
	idene)-1,5-pentanediyl]bis-	29	302	C22H38
	Heptadecane, 9-(2-cyclohexylethyl)-	28	350	C25H50
	Heptadecane, 7-(2-Cyclonex)2	26	111	C6H3NO
	3-METHOXY-2-METHYLFYRROL	25	252	C18H36
	3-Octadecene, (E)	_ <del>-</del>	•	
		47	490	C35H70
7	17-Pentatriacontene	47	266	C73H38
	7-CYCLOHEXYLTRIDECANE	43	350	C25H5Q
	Heptadecane, 9-(2-cyclohexylethyl)-	42	718	C\$0H1020
	1-Pentacontanol	39	466	C32#660
	1-Dotriacontanol			•
	·	42	124	C9H16
8	5-Ethylnorbornane	40	138	C10H18
_	endoisocamphane	35	138	C9H14O
	CIS-1-ETHINYL-2-METHYL-1-CYCLOHEXANOLE	30	264	C17H28O2
	8-epiambreinolide	. 25	138	C9H14O
	3,8-Nonadien-2-one, (E)-	25		
	·	60	362	C26H50
g	Pentalene, octahydro-1-(2-octyldecyl)-	69	718	C50H1020
•	1-Pantacontanol	66	350	C25850
	Cyclopentane, (4-octyldodecyl)-	47	256	C19H3B
	7-CVCLORRYYLTRIDECANE	36	374	C27H50
	1,1':3',1''-Tercyclopentane, 2'-dodecyl-	- 35	3/4	42,200
			192	C14H24
10	1,5,9-DECATRIENE, 2,3,5,8-TETRAMETHYL-	85	360	C26H48
10	Phononthrone. 2-dodecyltetradecanydro-	36	278	C20E38
	2-M-DUTYL-3-N-HEXYLDECAHYDRONAPHTHALEND	34		C20H38
	2-H-BUTYL-8-N-HEXYLDECAHYDROHAPHTHALENE	34	278	C26H50
	Pentalene, octahydro-1-(3-octyldecyl)-	31	362	C20830

<sup>1</sup>Number of compound peak in decreasing order of prominence.
2Probability of correct identity of unknown compound, expressed as percentage.

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### SAN FRANCISCO

### OFFICE HEMORANDUM

	ACTION	INFO		
	1	1	FILE:	
0:		<u> </u>		
	RDD		11	
	SD		<u>                                 </u>	
	BDD	<u> </u>		
	AMH	<u> </u>	<u></u>	
	KJE	1		
	DCK			
	WTL		<u></u>	
	MKP			
	AFR		- 11	
	S. SAUNDERS	<u> </u>	11	
	JRT	1 1	11	
	ији		11	
	MCY	<u>                                     </u>	DATE:	AUGUST 12, 1987
		T P#	REPLY	REQUIRED BY:
FRO	4: BILLY VIL	<u> </u>		
SUB	JECT: MINI-CONE	& EARTH TECHNOLOGY		

The moment you have been eagerly awaiting!! Without even knowing it!!!

Earth Technology will be displaying their mini-cone (a CPT unit on a four-wheel drive truck) in our very own parking lot (Howard & Beale) at 4:30 pm, Thursday, August 13. The unit can be used for rapid exploration to 20-foot depths.

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