December 5, 1994

Mr. David Johnson Mills College 5000 MacArthur Boulevard Oakland, CA 94613

Re: Ground Water Sampling Report Mills College Corporation Yard, Oakland, California Project No.: K275-H

Dear Mr. Johnson:

Harza is please to submit this report for the above referenced project. We are submitting copies to Juliet Shin of Alameda County Health Care Services and to the Regional Water Quality Control Board on your behalf.

Should you have any questions or require additional information, please do not hesitate to contact me.

Sincerely,

Harza Consulting Engineers and Scientists

Dennis Laduzinsky, C.E.G. ^L Head, Geology and Hydrogeology

DL\lk:aa\encl. Copies: Addressee (2) Ms. Juliet Shin (ACHCSA - 1) Alameda County LUFT Case Officer (RWQCB - 1)

Ground Water Sampling Report Mills College Corporation Yard Oakland, California

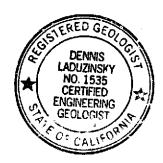
December 5, 1994

Prepared For:

Mills College 5000 MacArthur Boulevard Oakland, CA 94613

Prepared By:

Harza Consulting Engineers and Scientists 425 Roland Way Oakland, CA 94621



Derek D. Armentrout Project Chemist D_P____

Dennis Laduzinsky, C.E.G. Head, Geology and Hydrogeology



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- B Laboratory Analytical Reports and Chain-of-Custody Records



Ground Water Sampling Report Mills College Corporation Yard Oakland, California

1.0 INTRODUCTION

This report presents the results of ground water sampling performed at the Mills College Corporation Yard in Oakland, California. The project location is shown on the Site Vicinity Map (Figure 1).

The purpose of the investigation has been to evaluate the extent of petroleum hydrocarbons in ground water related to a previously removed gasoline underground storage tank (UST) at the site. This investigation was performed to comply with the continuing monitoring program under the jurisdiction of Alameda County Health Care Services Agency (ACHCSA).

2.0 BACKGROUND

In October 1988, a 1,000-gallon gasoline UST was removed from the Corporation Yard facility. A report prepared by Blaine Tech Services, Inc. of San Jose, California, indicated that soil samples collected from a depth of 21 feet below ground surface (bgs) following tank removal contained moderately high levels of total petroleum hydrocarbons as gasoline (TPHg). It is understood that 100 cubic yards of contaminated soils were excavated from the tank pit area at the time of tank removal and aerated on-site. The ACHCSA subsequently issued a letter, dated February 15, 1989, requesting investigation of the vertical and lateral extent of petroleum hydrocarbons in soil and ground water related to the former tank.

Beginning in June 1989, Harza (formerly Kaldveer Associates) performed soil and ground water quality investigations at the site, consisting of the installation and sampling of three ground water monitoring wells and two additional shallow soil borings.

The results of these investigations, presented in a report titled "Soil and Ground Water Testing Report For Mills College Corporation Yard", dated May 7, 1991, indicated that the majority of gasoline contamination in the unsaturated zone in the vicinity of the tanks appeared to have been removed during the soil excavation program conducted when the tanks were removed. Analysis of ground water samples collected from the monitoring wells since June 1989 have indicated the presence of TPHg at concentrations up to 11 parts per million (ppm).

K275-H reports\28054 12-05-94



The measured ground water flow direction at the site has usually been toward the south, beneath the existing Corporation Yard buildings, but has occasionally been directed southwesterly.

In May 1994, well MW-4 was installed downgradient of the Corporation Yard along Seminary Avenue. The well was installed to respond to the ACHCSA letter of April 23, 1993 requesting an additional downgradient monitoring point. In their September 7, 1994 letter, the ACHCSA expressed concern that well MW-4 was not screened in the same aquifer as wells MW-1 through MW-3, and requested an investigation to determine if well MW-4 is hydraulically connected to the other wells.

3.0 SCOPE OF SERVICES

The investigation consisted of the following tasks:

- Measuring ground water levels in all wells for use in developing a ground water elevation contour map.
- Collecting ground water samples from the four wells at the site.
- Analyzing ground water samples for TPHg using EPA Method 5030/GCFID, and for purgeable aromatic compounds (benzene, toluene, ethylbenzene, and xylenes or BTEX) using EPA Method 8020.
- Analyzing ground water samples from wells MW-3 and MW-4 for general minerals and preparing geologic cross sections to determine if the wells are screened in the same aquifer.
- Preparing this report.

4.0 FIELD INVESTIGATION

4.1 <u>Well Sampling</u>

The four monitoring wells were sampled on October 26, 1994. Following an initial ground water level measurement, a minimum of three well-casing volumes of water were purged from each well using a Teflon bailer. Purging consisted of the gradual removal of water from the well until physical parameters such as pH, temperature, and electrical conductivity stabilized.



Following purging, samples were decanted from the bailer into appropriate sample containers, labeled, and placed in refrigerated storage for transport to the laboratory under chain-of-custody control. The bailer was washed with trisodium phosphate (TSP) and rinsed with deionized water between wells to reduce the potential for cross contamination. Purge water was contained on-site in 55-gallon drums. Monitoring well sampling logs are attached to this report as Appendix A.

4.2 Ground Water Gradient

Well-top elevations have been surveyed to a common datum and water levels were measured in each well. Well-top elevations, depth to water, and calculated water-surface elevations are presented in Table 1. These data are used to generate the ground water elevation contours presented on Figure 2. Ground water elevation data collected during this investigation indicate ground water has a gradient of 0.0024 foot per foot to the southwest. This flow direction is slightly westward of the direction that has historically been observed at the site.

5.0 ANALYTICAL RESULTS

5.1 Laboratory Procedures

Ground water samples were analyzed by American Environmental Network (AEN) of Pleasant Hill, California. AEN is certified by the California Environmental Protection Agency for the analyses performed. Samples from each well were analyzed for TPHg using EPA Method 5030/GCFID, and for BTEX using EPA Method 8020.

5.2 Analytical Results

The results of the chemical analyses are presented in Table 2 and laboratory analytical results are attached to this report as Appendix B. A historical summary of ground water sample analytical results is also included in Table 2.

TPHg was detected in the sample from well MW-1 at a concentration of 8.7 ppm. BTEX compounds were detected in the sample from MW-1 at concentrations of 1.0, 0.29, 0.14 and 0.36 ppm, respectively. A petroleum odor and a slight hydrocarbon sheen on the water surface were recognized during the purging of the well.



TPHg was detected in the sample from well MW-2 at 0.2 ppm. Benzene was also detected in the sample at 0.13 ppm. Benzene (0.004 ppm) was the only compound detected in the sample from well MW-3. No TPHg or BTEX compounds were detected in the sample from well MW-4.

6.0 MW-4 INVESTIGATION

A geologic cross-section was prepared for the site to evaluate if well MW-4 is hydraulically connected to wells MW-1 through MW-3 (Figure 3). Boring logs from wells drilled at the site indicate the surficial soils at the site consist of sandy clay and clayey sand, underlain by deeply weathered (decomposed) granitic bedrock of the Leona Rhyolite to depths of approximately 20 feet. Weathering of the Leona Rhyolite is characterized by the presence of highly decomposed rock fragments loosely embedded in a clay matrix that may locally persist to a depth of 30 feet. Fresh bedrock is extremely hard, sheared, friable, and closely fractured. The fractures are generally open and interconnected except where filled with clay.

The upper 20 feet of bedrock shows intense decomposition of rock minerals to clay. This zone acts as a quasi-impermeable bed that confines the ground water level at or near the approximate depth where the degree of decomposition changes from intense to slight or moderate. The change in degree of mineral decomposition allows the water to flow more freely within the open and interconnected fractures.

As shown on Figure 3, stabilized water level measurements for all four wells show a distinct planar surface parallel to ground surface and the lower boundary of intense mineral decomposition. However, it should be noted that ground water in MW-4 was initially encountered at a depth of 45 feet at the time of drilling, and subsequently rose to a depth of 14 feet within about three hours. $\neg publicly dive to preserve head$

From a geologic standpoint, the characteristics of the Leona Rhyolite and the occurrence of the stabilized ground water levels observed at the site would suggest that all four wells are hydraulically connected by open and interconnected fractures in the deeper, less weathered bedrock. However, the strong confined ground water conditions observed in MW-4 may indicate that whereas ground water between MW-4 and MW-1, 2, and 3 is hydraulically connected at depth, specific weathering differences in the shallow bedrock zones may preclude direct connection of the upper surface of the saturated zone.



A chemical investigation was also performed for well MW-4. Ground water samples from wells MW-3 and MW-4 were analyzed for general minerals and water quality parameters to evaluate if the samples represent water from the same aquifer. Analytical results are shown on Table 3. The results were plotted on Piper and Stiff diagrams (Figures 4 and 5). These diagrams are useful for plotting mineral concentrations and visually comparing the concentrations for different wells. In general, the mineral concentrations and general water quality parameters for wells MW-3 and MW-4 are similar. The alkalinities of the two samples show the most significant difference. Alkalinity will increase as water flows through carbonaceous rock such as that found at the site. The increase in total dissolved solids (TDS) between wells MW-3 and MW-4 may also be due to dissolution of minerals as water moves through the aquifer. In summary, the chemistry of the ground water samples does not conclusively indicate that the samples represent different aquifers.

7.0 CONCLUSIONS

The sampling performed between June 1989 and October 1994 have shown a fluctuation in reported TPHg and BTEX concentrations, although no trends are apparent. Ground water elevations measured during this, and the previous investigation, indicate an apparent shift in flow direction toward the southwest, from the general southerly flow direction historically measured at the site. The gradient remains relatively flat. It is unclear at this time if the ground water elevations represent a permanent or temporary change in flow direction.

The geologic and chemical investigation of well MW-4 indicates that the well may be hydraulically connected to wells MW-1 through MW-3, but a conclusive determination could not be made. We therefore recommend that monitoring be continued at the four existing wells at the site on a quarterly schedule. The next monitoring event at the Corporation Yard will occur in January 1995. varyin

8.0 LIMITATIONS

The purpose of a geologic/hydrogeologic study is to reasonably characterize existing site conditions based on the geology/hydrogeology of the area. In performing such a study, a balance must be struck between a reasonable investigation into the site conditions and an exhaustive analysis of each conceivable condition. The following paragraphs discuss the assumptions and parameters under which such a study is conducted.



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No investigation is thorough enough to detect every geologic/hydrogeologic condition of interest at a given site. If conditions have not been identified during the study, such a finding should not therefore be construed as a guarantee of the absence of such conditions at the site, but rather as the result of the services performed within the scope, limitations, and cost of the work performed.

We are unable to report on or accurately predict events that may change the site conditions after the described services are performed, whether occurring naturally or caused by external forces. We cannot assume responsibility for conditions we were not authorized to evaluate, or conditions not generally recognized as predictable when services were performed.

Geologic/hydrogeologic conditions may exist at the site that cannot be identified solely by visual observation. Where subsurface exploratory work was performed, our professional opinions are based in part on interpretation of data from discrete sampling locations that may not represent actual conditions at unsampled locations.



TABLES

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Table 1 GROUND WATER ELEVATION DATA

Ground Water Sampling Report Mills College Corporation Yard, Oakland, California (Reported in feet)

Monitoring Well	Relative Well-Top Elevation (1)	Depth to Water	Ground Water Elevation		
June 1989:					
MW-1	100.00	19.44	80.56		
MW-2	99.98	19.36	80.62		
MW-3	100.01	19.40	80.61		
December 1990:	······································				
MW-1	100.00	22.05	77.95		
MW-2	99.98	21.96	78.02		
MW-3	100.01	22.00	78.01		
June 1991:					
MW-1	100.00	20.85	79.15		
MW-2	99.98	20.76	79.22		
MW-3	100.01	20.81	79.20		
March 1992:			······································		
MW-1	100.00	19.87	80.13		
MW-2	99.98	19.92	80.06		
MW-3	100.01	19.82	80.19		
October 1992:					
MW-1	100.00	21.69	78.31		
MW-2	99.98	21.60	78.38		
MW-3	100.01	21.65	78.36		
May 1994:					
MW-1	100.00	19.66	80,34		
MW-2	99.97	19.62	80.35		
MW-3	100.01	19.60	80.41		
MW-4	88.88	13.60	75.28		
June 1994;					
MW-1	100.00	19.72	80.28		
MW-2	99.97	19.65	80.32		
MW-3	100.01	19.65	80.36		
MW-4	88.88	14.01	74.87		

Table 1 GROUND WATER ELEVATION DATA

Ground Water Sampling Report Mills College Corporation Yard, Oakland, California (Reported in feet)

Monitoring Well	Relative Well-Top Elevation (1)	Depth to Water	Ground Water Elevation
October 1994:			
MW-1	100.00	20,17	79.83
MW-2	99.97	20,10	79.87
MW-3	100.01	20.08	79.93
MW-4	88.88	17.95	70.93

NOTES

(1): Well-top elevations are based on an arbitrary datum of 100.00 feet at MW-1.

Table 2

SUMMARY OF GROUND WATER SAMPLE ANALYSES

Ground Water Sampling Report Mills College Corporation Yard, Oakland, California (Reported in parts per million, or milligrams per liter)

Sample Date	ТРН	Benzene	Toluene	Ethylbenzene	Xylenes
	Gasoline				
<u>MW-1:</u>					
June 1989	11	2.1	1.9	0.031	1.4
December 1990	2.5	0.4	0.21	0.056	0.31
June 1991	16	2	1.1	0.41	2.8
March 1992	1.6	0.26	0.1	0.47	0.12
October 1992	2.8	0.33	0.13	0.06	0.2
October 1992(D)	4.2	0.54	0.23	0.08	0.36
May 1994	3.4	0.6	0.11	0.11	0.15
October 1994	8.7	1.0	0.29	0.14	0.36
<u>MW-2:</u>					
June 1989	ND	ND	ND	ND	ND
December 1990	ND	ND	ND	ND	ND
June 1991	ND	0.005	ND	ND	ND
March 1992	0.09	0.047	0.0005	ND	ND
October 1992	ND	0.003	ND	ND	ND
May 1994	0.2	0.084	0.0006	ND	ND
October 1994	0.2	0.13	ND	ND	ND
<u>MW-3:</u>					
June 1989	ND	ND	ND	ND	ND
December 1990	0.05	0.011	ND	ND	ND
June 1991	0.1	0.007	ND	ND	ND
March 1992	0.09	0.27	0.0009	ND	ND
October 1992	ND	0.005	ND	ND	ND
May 1994	ND	0.005	ND	ND	ND
October 1994	ND	0.004	ND	ND	ND
<u>MW-4:</u>				, <u>, , , , , , , , , , , , , , , , , , </u>	
May 1994	ND	ND	ND	ND	ND
October 1994	ND	ND	ND	ND	ND

NOTES

TPH: Total petrolum hydrocarbons

ND: Not detected at or above the laboratory method reporting limits

(D): Duplicate sample analytical results

Table 3

GROUND WATER GENERAL MINERALS ANALYSES

Ground Water Sampling Report

Mills College Corporation Yard, Oakland, California (Reported in parts per million, or milligrams per liter)

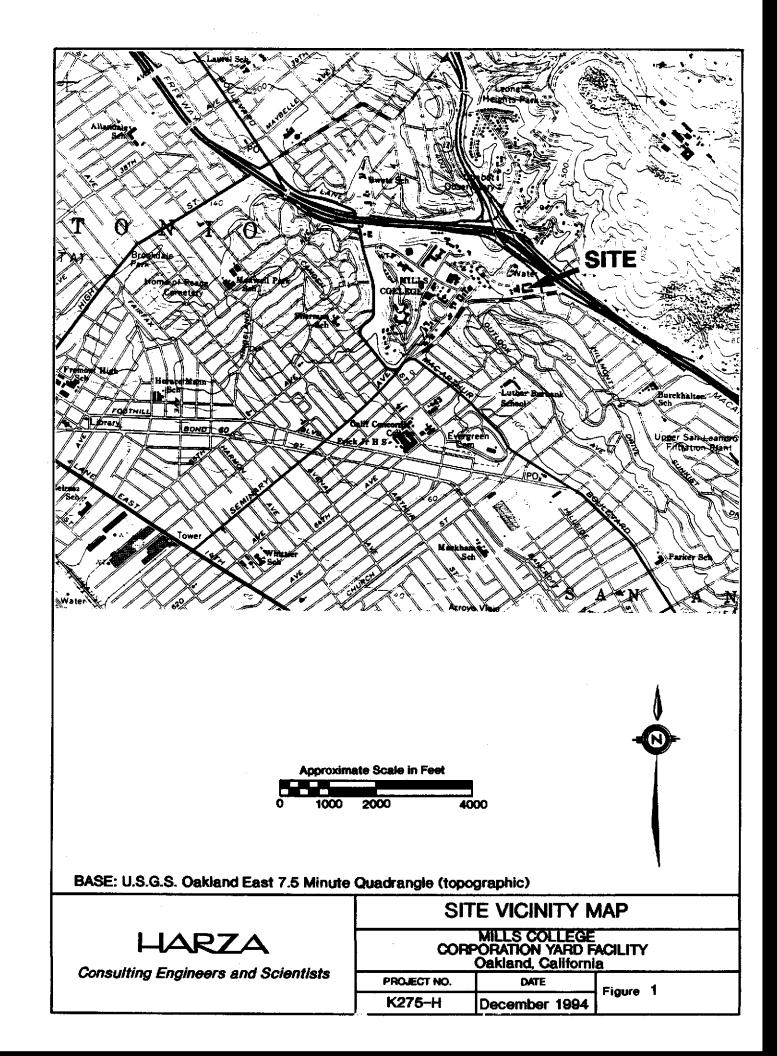
	Monitor	ring Well
Parameter	MW-3	MW-4
Nitrate-Nitrogen	0.3	<0.1
Phosphate-Phosphorus	<0.5	<0.5
Chloride	42	48
Sulfate	61	51
Bicarbonate Alkalinity	90	510
Carbonate Alkalinity	<2	<2
Hydroxide Alkalinity	<2	<2
Calcium	64	140
Copper	0.48	0.19
Iron	140	140
Magnesium	68	92
Manganese	1.3	1.3
Sodium	18	41
Zinc	0.11	0.16
Hardness	440	730
Total Dissolved Solids	280	670
Electrical Conductivity (micromhos per centimeter)	480	1200
pH (units)	6.6	7.2

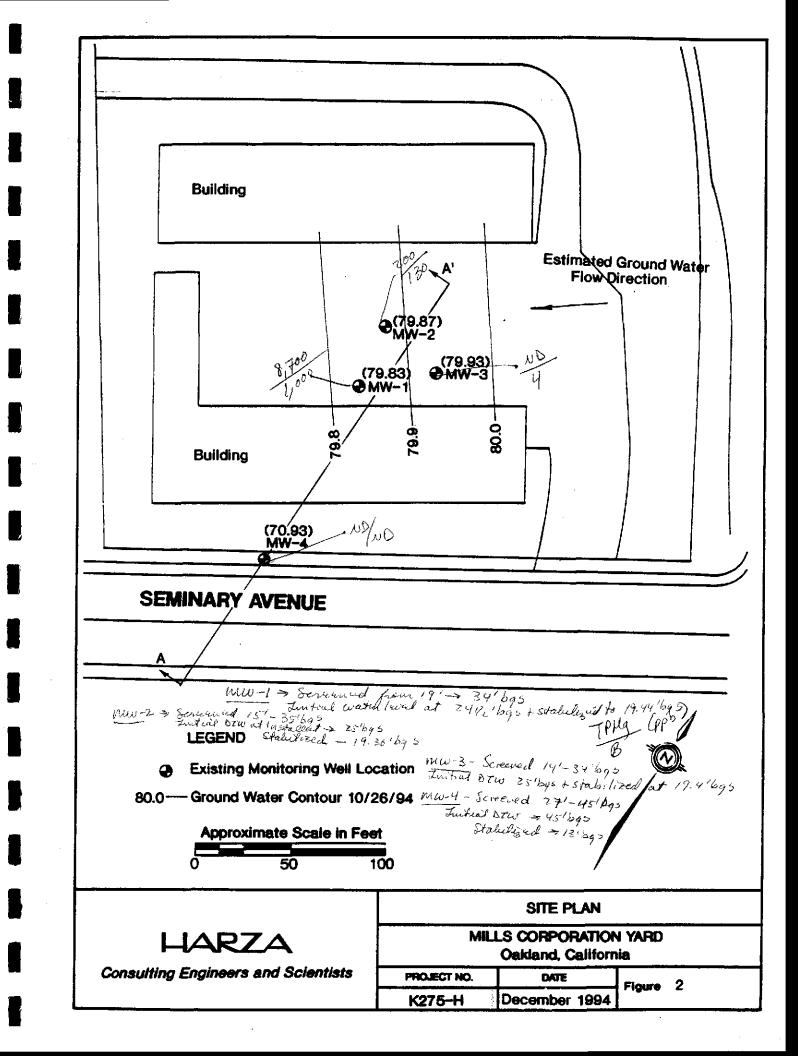
NOTE

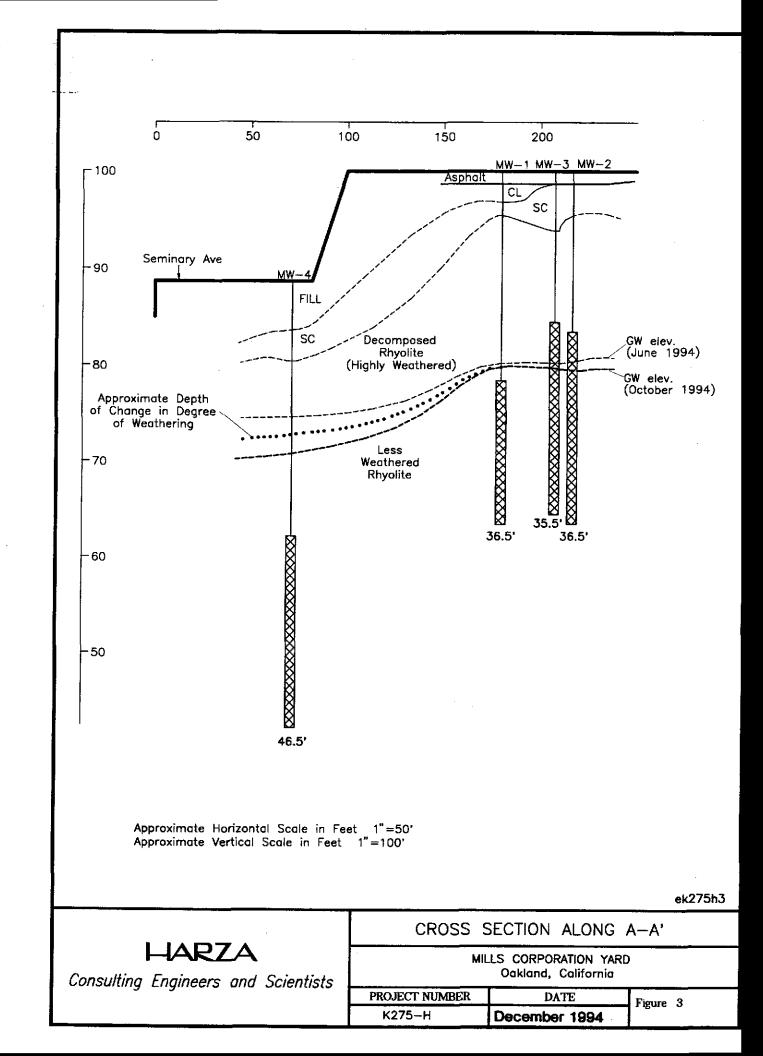
Samples were collected on 10/26/94

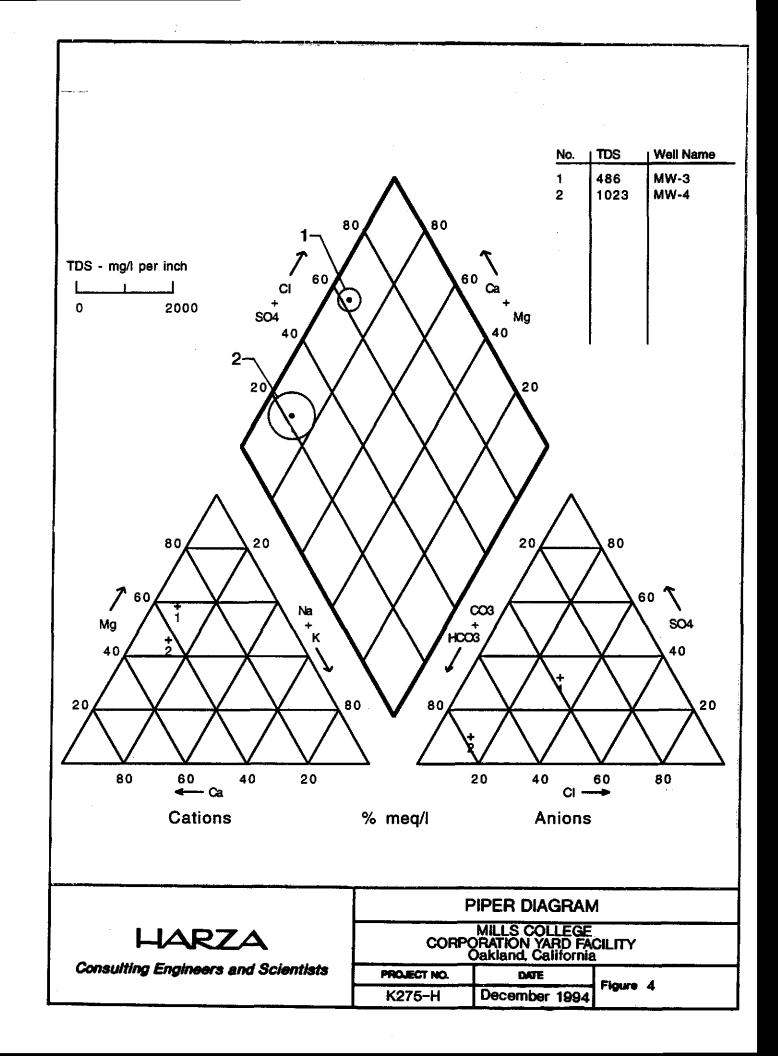
K275-H spread\28054.XLS 12/5/94 **FIGURES**

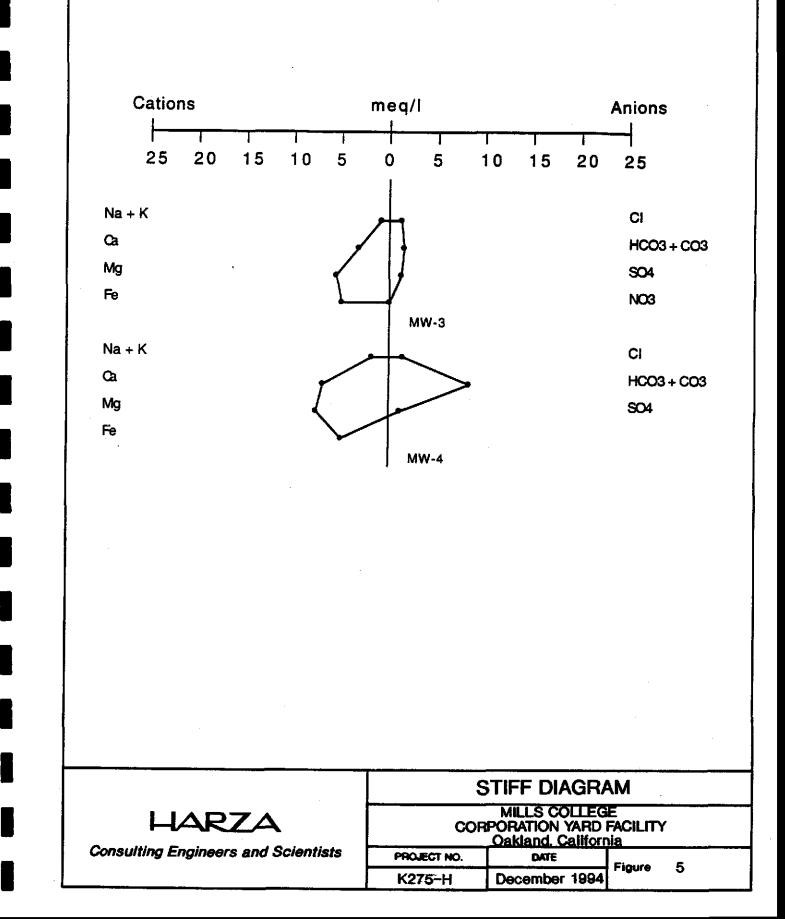
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APPENDIX A Well Sampling Logs

WATER SAMPLE LOG

Project Name: Project Number: Well Number: Well Location:	Mills College - G. Rr. Yaro K275 - H MW - J	Date: Sampier: Weather:	10/26/94 Derek Armentrout
Well Construction		Sampling Equipment 8	2 Cleaning
Date Completed:		Sampler Type:	Teflon bailer
Total Depth of Wel	1: ³ 2,5	Method of Cleaning:	TSP wash/DI rinse
Diameter:	2 inches	Pump/Bailer Type:	Teflon bailer
Well Elevation and	Reference:	Method of Cleaning:	TSP wash/DI rinse
	······································	pH Meter:	Hydac
		Conductivity Meter:	Hydac
Ground Water Leve	<u>els:</u>	Comments:	
Initial: 20. Final: 20 Reference Point: Well Volume of Wa	2.76 top of casing		

SAMPLING MEASUREMENTS

	Dischar	ge (gal.)		Temp	1 -	nductance os/cm)	Color/		
Time	Per Time Period	Cumulative	рН	(T F)	Field	@ 25°C	Turbidity	Odor	
1439	start	0			1				
1447		2.5	6.86	67.7	1280		BLACK / HIGH	STR. PETR	
14 54		5	6.80	67.0	1300		4	4	
1459		7	6.91	66.9	1270		¥	ખ	
		SAMPLED							
							. <u>.</u>		
							÷		
L								x	
		A.		······					
Casing Vol	Total Discharge: 7 991 Casing Volumes Removed: 3.5				Comments:	Sheen	ON WATER		
Method of	Disposal: 55-0	irum at Corpo	ration Yard	<u> </u>			<u>.</u> .		
					WATER SAMPLE LOG				
	HARZA Consulting Engineers and Scientists					Project No. Date Figure			

WATER SAMPLE LOG

Sampling Equipment &	c Cleaning
Sampler Type:	Teflon bailer
Method of Cleaning:	TSP wash/DI rinse
Pump/Bailer Type:	Teflon bailer
Method of Cleaning:	TSP wash/DI rinse
pH Meter:	Hydac
Conductivity Meter: Comments:	Hydac
	Method of Cleaning: Pump/Bailer Type: Method of Cleaning: pH Meter: Conductivity Meter:

SAMPLING MEASUREMENTS

					1 -	nductance			
		rge (gal.)	1	Temp	·	os/cm)	Color/		
Time	Per Time	Cumulative	pH	(°F)	Field	@ 25°C	Turbidity	Odor	
 	Period	Į'			<u> </u>				
	start	0					6		
1413		2,5	7,45	70.6	650		ORANGE/HIGH	NONE	
1420		5	7,77	69.5	630		ų	<i>i</i> (
1426		7.5	7.34	69.5	610		ધ	ţe	
			PLED						
					ļ	· · · · · ·			
								1	
	lumes Remove			- 17	Comments:				
Method of	Disposal: 55-0	drum at Corpo	ration Yard	<u></u>					
						WAT	ER SAMPLE LOG		
		HARZA Engineers an	-		Project No. Date Figure				

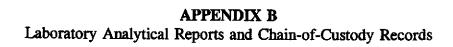
WATER SAMPLE LOG

Project Name:Mills College - CoRPVARDProject Number:K275 - HWell Number:M W - 3	Date: Sampler: Weather:	10/26/94 Derek Armentrout
Well Location:	Weather	
Well Construction	Sampling Equipment &	¿ Cleaning
Date Completed:	Sampler Type:	Teflon bailer
Total Depth of Well: 32.5	Method of Cleaning:	TSP wash/DI rinse
Diameter: 2 inches	Pump/Bailer Type:	Teflon bailer
Well Elevation and Reference:	Method of Cleaning:	TSP wash/DI rinse
	pH Meter:	Hydac
· · · · · · · · · · · · · · · · · · ·	Conductivity Meter:	Hydac
Ground Water Levels:	Comments:	
Initial: $2^{\circ} \cdot 08$		
Final: 22.02		
Reference Point: top of casing		
Well Volume of Water: 2 gal		

SAMPLING MEASUREMENTS

	Dischar	ge (gal.)		Temp	Spec. Conductance (µmhos/cm)		Color/			
Time	Per Time Period	Cumulative	pН	(°F)	Field	@ 25°C	Turbidity	Odor		
	start	0								
1324		2	7.50	68.7	4200		CRAY/MODRATE	None		
		4	6.87	67.6	Grans S	30				
1345		65	681	68-0	550					
	SA	mpced								
	•									
							a.	_		
Casing Vol	otal Discharge: 6.5 94				Comments:	· · · · · · · · · · · · · · · · · · ·				
Method of	Disposal: <u>55-d</u>	rum at Corpor	ration Yard							
						WATER SAMPLE LOG				
	HARZA Consulting Engineers and Scientists					Project No. Date Figure				

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American Environmental Network Certificate of Analysis

DOHS Certification: 1172

AIHA Accreditation: 11134

PAGE 1

HARZA 425 ROLAND WAY OAKLAND, CA 94621

ATTN: DEREK ARMENTROUT CLIENT PROJ. ID: K275-H REPORT DATE: 11/08/94 DATE(S) SAMPLED: 10/26/94 DATE RECEIVED: 10/26/94 AEN WORK ORDER: 9410322

PROJECT SUMMARY:

On October 26, 1994, this laboratory received 4 water sample(s).

Client requested sample(s) be analyzed for inorganic and organic parameters. Results of analysis are summarized on the following page(s).

Please see quality control report for a summary of QC data pertaining to this project.

If you have any questions, please contact Client Services at (510) 930-9090.

.an⁄cy Klein

Laboratory Director

HARZA

SAMPLE ID: MW-1 AEN LAB NO: 9410322-01 AEN WORK ORDER: 9410322 CLIENT PROJ. ID: K275-H

DATE SAMPLED: 10/26/94 DATE RECEIVED: 10/26/94 REPORT DATE: 11/08/94

ANALYTE	METHOD/ CAS#	RESULT	REPORTING LIMIT	UNITS	DATE ANALYZED
BTEX & Gasoline HCs Benzene Toluene Ethylbenzene Xylenes, Total Purgeable HCs as Gasoline	EPA 8020 71-43-2 108-88-3 100-41-4 1330-20-7 5030/GCFID	1,000 * 290 * 140 * 360 * 8.7 *	1 1 4 0.1	ug/L ug/L ug/L ug/L mg/L	11/02/94 11/02/94 11/02/94 11/02/94 11/02/94

Reporting limits elevated for gasoline/BTEX due to high levels of target compounds; sample run at dilution.

HARZA

SAMPLE ID: MW-2 AEN LAB NO: 9410322-02 AEN WORK ORDER: 9410322 CLIENT PROJ. ID: K275-H

DATE SAMPLED: 10/26/94 DATE RECEIVED: 10/26/94 REPORT DATE: 11/08/94

ANALYTE	METHOD/ CAS#	RESULT	REPORTING LIMIT	UNITS	DATE ANALYZED
BTEX & Gasoline HCs Benzene Toluene Ethylbenzene Xylenes, Total Purgeable HCs as Gasoline	EPA 8020 71-43-2 108-88-3 100-41-4 1330-20-7 5030/GCF1D	130 * ND ND 0.2 *	0.5 0.5 0.5 2 0.05	ug/L ug/L ug/L mg/L mg/L	11/02/94 11/02/94 11/02/94 11/02/94 11/02/94

HARZA

SAMPLE ID: MW-3 AEN LAB NO: 9410322-03 AEN WORK ORDER: 9410322 CLIENT PROJ. ID: K275-H

DATE SAMPLED: 10/26/94 DATE RECEIVED: 10/26/94 REPORT DATE: 11/08/94

ANALYTE	METHOD/ CAS#	RESULT	REPORTING LIMIT	UNITS	DATE ANALYZED
BTEX & Gasoline HCs Benzene Toluene Ethylbenzene Xylenes, Total Purgeable HCs as Gasoline	EPA 8020 71-43-2 108-88-3 100-41-4 1330-20-7 5030/GCFID	4 * ND ND ND ND	* 0.5 0.5 0.5 2 0.05	ug/L ug/L ug/L ug/L mg/L	11/02/94 11/02/94 11/02/94 11/02/94 11/02/94
Nitrate-Nitrogen	EPA 300	0.3 *	.1	mg/L	10/28/94
Phosphate-P in water	EPA 300	ND	0.5	mg/L	10/28/94
General Minerals Bicarbonate Alkalinity Carbonate Alkalinity Hydroxide Alkalinity Calcium Chloride Copper Iron Magnesium Manganese pH Sodium Sulfate Conductivity Total Dissolved Solids Hardness Zinc	EPA 310.1 EPA 310.1 EPA 310.1 EPA 6010 EPA 6010 EPA 6010 EPA 6010 EPA 6010 EPA 6010 EPA 6010 EPA 9040 EPA 6010 EPA 120.1 EPA 160.1 SM 2340B EPA 6010	90 * ND 64 * 42 * 140 * 140 * 1.3 * 6.6 * 18 * 480 * 280 * 440 *	2 0.05 0.5 0.01 0.05 0.04 0.005 NA 0.1 0.5 20 10	mg CaCO3/L mg CaCO3/L mg CaCO3/L mg/L mg/L mg/L mg/L std. units mg/L umhos/cm mg/L mg/L umhos/cm mg/L mg/L	11/01/94 11/02/94 11/02/94 11/02/94 11/02/94 11/02/94 11/02/94 10/26/94 10/26/94 10/28/94 10/28/94 10/31/94 11/01/94

HARZA

SAMPLE ID: MW-4 AEN LAB NO: 9410322-04 AEN WORK ORDER: 9410322 CLIENT PROJ. ID: K275-H

DATE SAMPLED: 10/26/94 DATE RECEIVED: 10/26/94 REPORT DATE: 11/08/94

ANALYTE	METHOD/ CAS#	RESULT	REPORTING LIMIT	UNITS	DATE ANALYZED	
BTEX & Gasoline HCs Benzene Toluene Ethylbenzene Xylenes, Total Purgeable HCs as Gasoline	EPA 8020 71-43-2 108-88-3 100-41-4 1330-20-7 5030/GCFID	ND ND ND ND ND	0.5 0.5 0.5 2 0.05	ug/L ug/L ug/L ug/L mg/L	11/02/94 11/02/94 11/02/94 11/02/94 11/02/94	
Nitrate-Nitrogen	EPA 300	ND	0.1	mg/L	10/28/94	
Phosphate-P in water	EPA 300	ND	0.5	mg/L	10/28/94	
General Minerals Bicarbonate Alkalinity Carbonate Alkalinity Hydroxide Alkalinity Calcium Chloride Copper Iron Magnesium Manganese pH Sodium Sulfate Conductivity Total Dissolved Solids Hardness Zinc	EPA 310.1 EPA 310.1 EPA 310.1 EPA 6010 EPA 6010 EPA 6010 EPA 6010 EPA 6010 EPA 6010 EPA 6010 EPA 300 EPA 120.1 EPA 160.1 SM 2340B EPA 6010	510 * ND ND 140 * 48 * 0.19 * 140 * 92 * 1.3 * 7.2 41 * 51 * 1.200 * 670 * 730 * 0.16 *	2 2 0.05 0.5 0.01 0.05 0.04 0.005 NA 0.1 0.5 20 10 1 0.01	mg CaCO3/L mg CaCO3/L mg CaCO3/L mg/L mg/L mg/L mg/L std. units mg/L umhos/cm mg/L mg/L mg/L mg/L mg/L	11/01/94 11/02/94 10/28/94 11/02/94 11/02/94 11/02/94 11/02/94 10/26/94 10/26/94 10/28/94 10/28/94 10/31/94 11/01/94	

American Environmental Network

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AEN (CALIFORNIA) QUALITY CONTROL REPORT

AEN JOB NUMBER: 9410322

CLIENT PROJECT ID: K275-H

<u>Quality Control and Project Summary</u>

All laboratory quality control parameters were found to be within established limits.

Definitions

Laboratory Control Sample (LCS)/Method Spike(s): Control samples of known composition. LCS and Method Spike data are used to validate batch analytical results.

Matrix Spike(s): Aliquot of a sample (aqueous or solid) with added quantities of specific compounds and subjected to the entire analytical procedure. Matrix spike and matrix spike duplicate QC data are advisory.

Method Blank: An analytical control consisting of all reagents, internal standards, and surrogate standards carried through the entire analytical process. Used to monitor laboratory background and reagent contamination.

Not Detected (ND): Not detected at or above the reporting limit.

Relative Percent Difference (RPD): An indication of method precision based on duplicate analysis.

Reporting Limit (RL): The lowest concentration routinely determined during laboratory operations. The RL is generally 1 to 10 times the Method Detection Limit (MDL). Reporting limits are matrix, method, and analyte dependent and take into account any dilutions performed as part of the analysis.

Surrogates: Organic compounds which are similar to analytes of interest in chemical behavior, but are not found in environmental samples. Surrogates are added to all blanks, calibration and check standards, samples, and spiked samples. Surrogate recovery is monitored as an indication of acceptable sample preparation and instrumental performance.

- D: Surrogates diluted out.
- #: Indicates result outside of established laboratory QC limits.

QUALITY CONTROL DATA

AEN JOB NO: 9410322 DATE(S) ANALYZED: 10/28-11/02/94 MATRIX: WATER

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Method Spike Recovery Summary

		Spike	Average		QC Limits				
Analyte	Inst./ Method	Added (mg/L)	Percent Recovery	RPD	Percent Recovery	RPD			
Cu, Copper Ca Calcium Fe, Iron Mg, Magnesium Mn, Mangenese Na, Sodium Zn, Zinc	ICP/6010 ICP/6010 ICP/6010 ICP/6010 ICP/6010 ICP/6010 ICP/6010	$\begin{array}{c} 0.13 \\ 10.0 \\ 0.50 \\ 10.0 \\ 0.25 \\ 10.0 \\ 0.25 \\ 10.0 \\ 0.25 \end{array}$	92 103 102 102 105 101 104	2 2 1 2 1 2 2	85-113 80-120 80-120 80-120 80-120 80-120 80-120 87-117	6 15 15 15 15 15 15 7			
Chloride Nitrate-N Phosphate-P Sulfate	DIONEX/300 DIONEX/300 DIONEX/300 DIONEX/300	$5.0 \\ 1.0 \\ 10.0 \\ 10.0 \\ 10.0 $	108 89 100 101	2 <1 1 4	80-120 80-120 80-120 80-120 80-120	15 15 15 15			

Daily method blanks for all associated analytical runs showed no contamination over the reporting limit.

*** END OF REPORT ***

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Sample Tea Lab Number	am Member (s) <u>DEREK</u> AR Client Sample Indetification	Air Volume	Date/ Time Collected	Sample Type*	Pres.	No. of Cont.	Type of Cont.			Avion Press		The second se					Commo	nts / Hazards
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